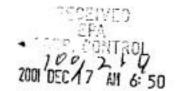
ARZO1-13468

COURTNEY M. PRICE VICE PRESIDENT CHEMSTAR





December 14, 2001

Christine Todd Whitman, Administrator U.S. Environmental Protection Agency P. O. Box 1473 Merrifield, VA 22116

> RE: Phthalate Esters Panel Test Plans for Phthalate and Trimellitate Esters; HPV registration number

#### Dear Ms. Whitman:

The Phthalate Esters Panel HPV Testing Group of the American Chemistry Council submits its test plans for Phthalate and Trimellitate Esters under the High Production Volume (HPV) Challenge Program. However, they are not volunteered in the HPV program, because they are described in the test plans and noted here for information purposes and are already part of the OECD SIDS program.

Chemical Name	CAS Number
1,2-benzenedicarboxylic acid, dibutyl ester	84-74-2
1.2-benzenedicarboxylic acid, butylbenzyl ester	85-68-7
1,2-benzenedicarboxylic acid, di(2-ethylhexyl)ester	117-81-7
1,2,4-benzenetircarboxylic acid, tris (2-ethylhexyl) ester	3319-31-1
1,2-benzenedicarboxylic acid diisononyl ester	28553-12-0
1,2-benzenedicarboxylic acid, di-C8-10-branched alkyl esters, C9 rich	68515-48-0
1,2-benzenedicarboxylic acid, di-C9-11-branched alkyl esters, C10 rich	68515-49-1

Thus, data presentation and any needed testing for these seven chemicals will occur through the OECD program. The Phthalate Esters Panel HPV Testing Group will, where appropriate, use these chemicals for data read across purposes.

In preparing this test plan, the Panel has given careful consideration to the principles contained in the letter EPA sent to all HPV Challenge Program participants on October 14, 1999. As requested by EPA in that letter, the Panel has sought to maximize the use of scientifically appropriate categories of related chemicals and of structure activity relationships. Additionally, and also as requested in EPA's letter, in analyzing the adequacy of existing data, the Panel has conducted a thoughtful, qualitative analysis rather than use a rote checklist approach. The Panel has taken the same thoughtful approach when developing this revised test plan and believes it conforms to those principles.

Christine Todd Whitman, Administrator December 14, 2001 Page 2

If you have any questions, please call Marian Stanley, Manager Phthalate Esters Pane, lat (703-741-5623), e-mail Marian\_Stanley@americanchemistry.com.

Courtney M. Price Vice President, CHEMSTAR

CC:

C. Auer, EPA

B. Leczynski

R. Hefter, EPA

S. Russell - ACC

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# HIGH PRODUCTION VOLUME (HPV) CHEMICAL CHALLENGE PROGRAM

**TEST PLAN** 

For The Trimellitate Category

Prepared by:

ExxonMobil Biomedical Sciences, Inc.

For The

Phthalate Esters Panel HPV Testing Group of the American Chemistry Council

**December 13, 2001** 

CONTAIN NO CBI

# THE PHTHALATE ESTERS PANEL

The American Chemistry Council Phthalate Esters Panel sponsoring this test plan includes the following member companies:

Eastman Chemical Company
ExxonMobil Chemical Company
Sunoco Chemicals
Teknor Apex Company

# TRIMELLITATE CATEGORY

CAS Number	CAS Number Description
3319-31-1	1,2,4-benzenetricarboxylic acid, tris (2-ethylhexyl) ester
27251-75-8	1,2,4-benzenetricarboxylic acid, triisooctyl ester
53894-23-8	1,2,4-benzenetricarboxylic acid, triisononyl ester
67989-23-5	1,2,4-benzenetricarboxylic acid, decyl octyl ester

# PLAIN ENGLISH SUMMARY

The trimellitates category contains four U.S. HPV trimellitates. These substances are 1,2,4 benzenetricarboxylic acids with side chain esters ranging from C8-C10. Of these, the one most extensively tested, Tris-2(ethylhexyl) trimellitate (TOTM), has been shown to have a low order of toxicity. Existing toxicology data on these substances were supplemented with information on phthalate esters (1,2 benzenedicarboxylic acids) with side chains of similar length.

The American Chemistry Council Phthalate Esters Panel HPV Testing Group believes that there is a sufficient amount of information available on trimellitates to substantially characterize the human health effects and environmental fate and effects endpoints for the remaining members of this category under the HPV program. TOTM has been sponsored under the OECD SIDS program through ICCA. A full SIDS data set exists for TOTM and is being used to support the hazard assessment of the remaining trimellitates in this category. No additional toxicology tests are proposed for these materials.

# **EXECUTIVE SUMMARY**

The American Chemistry Council Phthalate Esters Panel HPV Testing Group and its member companies hereby submit for review and public comment the test plan for the Trimellitate category under the Environmental Protection Agency's (EPA) High Production Volume (HPV) Chemical Challenge Program (Program). It is the intent of the Phthalate Esters Panel and its member companies to use existing data and scientific judgment/analyses to meet the requirements of the Screening Information Data Set (SIDS) for human health, environmental fate and effects, and physical/chemical properties for this category.

This test plan addresses the 4 HPV trimellitates listed in Table 1. Trimellitates are produced by esterification of trimellitic anhydride (TMA) with various linear and branched alcohols in the presence of an acid catalyst to form 1,2,4—benzenetricarboxylic acids. Because the side chains for all substances in this category are of similar carbon number (C8-C10) and structure, all four of the HPV substances were grouped into a single category.

Trimellitates are used predominantly as plasticizers for production of flexible PVC. Because of their relatively high molecular weight (>500 g/mole) and bulky structure, they have lower volatility and greater resistance to migration than the corresponding phthalate ester plasticizers. They are predominantly used in the manufacture of high temperature PVC cables (Wilson, 1996). Since these chemicals are produced in closed systems, there is essentially no occupational exposure to these substances except at the flexible PVC production facility. Usually, these substances have been already blended to the compound as plasticizer, so it is not expected that downstream users or consumers are directly exposed to trimellitates.

# **Testing Rationale:**

Because of the similarity in chemical structure, the Panel believes that the toxicological properties of the substances in this category will be similar as well. Thus, the Panel considers that the data for the best tested member of this category, tris-2(ethylhexyl) trimellitate (TOTM), also represent the potential for human and environmental effects of the other members of this category. In addition, data on TOTM indicate that it is hydrolyzed very poorly in rodents to the di-2-ethylhexyl ester and a mono-2-ethylhexyl ester. Therefore, "read across" for trimellitates would consist of comparisons to the similar phthalate esters, which are also being sponsored by the Panel under the HPV program. Existing toxicology data on these substances were supplemented with information on phthalate esters (1,2 benzenedicarboxylic acids) with side chains of similar length (see test plan for phthalate esters category).

TOTM has been sponsored by Japan under the OECD SIDS program. A review of the available data for TOTM (Table 2) indicates that all endpoints have been adequately addressed, and that TOTM exhibits a low order of toxicity. Further, a comparison of the relative toxicity of TOTM to its corresponding phthalate ester, di-ethylhexyl phthalate

(DEHP), indicates that trimellitates are much less active than phthalate esters with side chains of similar length. Due to their higher molecular weight and bulky side chains, the remaining members of this category are expected to demonstrate a lower order of toxicity than TOTM. This is supported by a similar structural-activity relationship observed with phthalate ester compounds, i.e., the higher molecular weight phthalates (ester side chains ≥C7) are less active that the transitional phthalates (ester side chains C4-C6). Thus, the use of TOTM to represent the potential hazards of the other category members is a conservative position. No additional toxicity tests are proposed for this category.

# TEST PLAN FOR THE TRIMELLITATE CATEGORY

# **INTRODUCTION**

The American Chemistry Council Phthalate Esters Panel HPV Testing Group and its member companies have committed voluntarily to develop screening level human health effects, environmental fate and effects, and physicochemical data for the trimellitates category under the Environmental Protection Agency's (EPA's) High Production Volume (HPV) Challenge Program.

This plan identifies CAS numbers used to characterize the SIDS endpoints for this category, identifies existing data of adequate quality for substances included in the category, and provides the Panel's rationale for utilizing the available SIDS data to characterize the potential hazards of all category members. The objective of this effort is to identify and adequately characterize the physicochemical properties along with human health, environmental fate and effects, to satisfy the EPA HPV program.

# **DESCRIPTION OF THE TRIMELLITATES CATEGORY**

The trimellitates comprise a family of chemicals synthesized by esterifying trimellitic anhydride with alcohols with average carbon numbers ranging from approximately C7-C10, in the presence of an acid catalyst. The category includes the four trimellitates listed in Table 1. Trimellitates in this category are all 1,2,4-benzenetricarboxylic acids with side chain ester groups ranging from C8 to C10. The structural formula for trimellitates varies somewhat depending on the isomeric composition of the alcohols used in their manufacture. The specific alcohols used are 2-ethylhexanol (TOTM), iso-octyl alcohol (TIOTM), iso-nonyl alcohol (TINTM), and a mixture of linear and branched decyl (40%) and octyl (60%) alcohols (DOTM).

Trimellitates are colorless to slightly yellow liquids with high boiling points (> 250°C) and low vapor pressures; these properties contribute to their high physical stability. They are readily soluble in most organic solvents and miscible with alcohol, ether and most oils, but essentially insoluble in water. Because of the similarity in structure as well as physicochemical properties, the trimellitates were grouped into a single category containing four substances with carboxylic side chain ester groups ranging from C8-C10.

# **DATA ADEQUACY REVIEW**

# Literature Search:

Literature searches were conducted by EMBSI Information Services on the environmental and mammalian toxicity endpoints for four trimellitates using the CAS numbers supplied by the Phthalate Esters Panel. A review of these substances was recently published (David et al., 2001). Therefore, the search was conducted using the MEDLINE and TOXLINE databases and limited to studies published since 1995. The TSCATS database was searched for relevant unpublished studies on these chemicals. In addition, a complete SIDS information package on TOTM was kindly provided by Dainippon Ink & Chemicals, Japan, as part of it's OECD SIDS submission. Standard handbooks and other reference material (CRC Handbook on Chemicals; IUCLID) were consulted for physical/chemical properties. Information on manufacture and use was taken from EPA (1981) and Wilson (1996).

In addition, modeled data were entered into the robust summaries for all of the physical properties. There are a number of reasons for this approach:

- The EPA guidance (www.epa.gov/opptintr/chmrtk/robsumgd.htm) allows inclusion of calculated values in the robust summaries for physical/chemical elements,
- The need for a complete set of physical property data in order to calculate environmental distribution, and
- Supplement measured physical properties for these trimellitates.

The physical properties were modeled using the SRI/EPA computer program EPIWIN, a modeling package that includes a number of algorithms developed at or for the EPA. EPIWIN is the program used and advocated by the EPA. Because the model is a structure-property model a specific discrete structure is required and EPIWIN contains a CAS number database which contains the structures for the chemicals. For mixtures, a single representative structure is contained in the database and in this work, these surrogate chemical structures were accepted for further modeling. It should be remembered that the resultant physical properties are for a single structure not a mixture so the values are discrete numbers rather than ranges.

The existing data for environmental and mammalian toxicology endpoints were reviewed using the literature searches to identify the most relevant studies for each chemical in the group. A number of the listed individual chemicals had no relevant studies identified in the searches. For the listed chemicals for which there were relevant data, all studies were reviewed using the criteria outlined in the EPA's method for determining the adequacy of existing data for the HPV program and the ranking system proposed by Klimisch et al. (1997). A list of the most relevant studies that were available for environmental and mammalian health endpoints is presented in **Appendix 1**.

Studies that were chosen for robust summaries represented the best available data for each specific endpoint. Published studies from the general literature, as well as a number of unpublished company reports, were obtained and summarized. Some endpoints include multiple summaries in order to present a more complete data set.

# **TESTING RATIONALE**

## Overview:

The trimellitates category contains four U.S. HPV trimellitates. These substances are 1,2,4 benzenetricarboxylic acids with side chain esters ranging from C8-C10. Of these, the one most extensively tested, TOTM, will be used as a representative chemical to assess the potential environmental and health effects of the other trimellitate category members. A review of the available data for TOTM (Table 2) indicates that all endpoints have been adequately addressed, and that TOTM exhibits a low order of toxicity. Due to their higher molecular weight and bulky side chains, the remaining members of this category are expected to demonstrate a lower order of toxicity than TOTM. This is supported by a similar structural-activity relationship observed with phthalate ester compounds, i.e., the higher molecular weight phthalates (ester side chains ≥C7) are less active than the transitional phthalates (ester side chains C4-C6). Thus, the use of TOTM to represent the potential hazards of the other category members is a conservative position. No additional toxicity tests are proposed for this category.

## Manufacturing and Use

Trimellitates are produced by esterification of trimellitic anhydride (TMA). The basic structure is an aromatic ring with side chains in the 1, 2 and 4 positions. Trimellitate plasticizers are based on alcohols with (average) carbon numbers in the range 7-9. The relatively high molecular weight and bulky structure of these molecules gives them low volatility and makes them relatively resistant to migration. Their main application is in high temperature PVC cables (Wilson, 1996).

# **Category Justification**

The four trimellitates in the HPV category, tris-2(ethylhexyl) trimellitate (TOTM), tri-isooctyl trimellitate (TIOTM), tri-isononyl trimellitate (TINTM) and decyl,octyl—trimellitate (DOTM). The distinguishing feature of these substances is in the alcohol side chains. TOTM has side chains with a 2-ethylhexyl moiety, TIOTM has iso-octyl side chains, TINTM has isononyl side chains and DOTM has mixed decyl (40%) and octyl (60%) side chains. These molecules are of the same general structure, differing only in side chains, and the side chains themselves are very similar, containing carbon numbers ranging from C8 to C10. These molecules also have similar physical and chemical properties; in particular because of their high molecular weights and aliphatic character,

they have very low vapor pressures and very low water solubilities. Because of the similarity in structure and physical/chemical properties, the Panel believes that it is reasonable to consider this group of substances as a category and to rely on data for one representative member (TOTM) for all other representatives in this category.

## Physicochemical Properties

Physicochemical properties for trimellitates are shown in Table 2A. The 2-ethylhexyl trimellitate ester (TOTM) is representative of this group of trimellitates as the other members are quite similar triesters of mellitic acid with C8 through C10 alcohols. TOTM has a melting point of -46°C and a boiling point of >300°C at 1 atmosphere (measured values are >300°C at reduced pressure). Vapor pressure measurements are only possible for TOTM at very high temperatures due to its low order of volatility. Measured values are <1 Pa at 100°C and 13 Pa at 200°C; thus, the vapor pressure at 25°C is extrapolated to be < 0.01 Pa. The vapor pressure calculated for TOTM by EPIWIN is  $5 \times 10^{-9}$  Pa. The water solubility of TOTM is also quite low. A measured value of  $4 \times 10^{-4}$  mg/L is available. However, water solubility is difficult to measure at such low concentrations, particularly for esters with densities near that of water and which tend to form dispersions in water and for that reason, standard test methods tend to over-estimate water solubility. The EPIWIN calculated water solubility value for TOTM is  $4.5 \times 10^{-8}$  mg/L. The log of the octanol/water partition coefficient (log  $K_{ow}$ ) for TOTM is calculated (EPIWIN) as 11.6. Measured values of 5.94 and 4.35 are also available.

Structure-property modeling has been done using the EPIWIN program recommended by EPA. This modeling has been used to estimate all of the required physicochemical parameters of all four of these HPV trimellitates. TOTM has a melting point below 0°C; it is expected that the other members of this group will have melting points below 0°C as well. Due to their high molecular weight, these trimellitates are expected to boil at a much higher temperature than TOTM and than the corresponding phthalate esters all of which boil at >300°C at atmospheric pressure. EPIWIN estimates boiling points >500°C for all four trimellitates. Thus, all boiling points are assuredly >300°C and measurement is not necessary.

For the phthalate esters, the EPIWIN model agrees well with measured values for the critical environmental fate properties of octanol-water partition coefficient, water solubility, and vapor pressure. By analogy with the phthalates and by EPIWIN calculations, these trimellitates are expected to be virtually water insoluble <1 part per billion) and non-volatile ( $\sim 10^{-9}$  Pa). Measured values on TOTM confirm the expected low water solubility and vapor pressures. The log  $K_{ow}$  values for all four trimellitates are calculated to be in the range of 11 to 13. The measured values reported for TOTM seem quite unlikely, since the measured value for the corresponding phthalate diester (DEHP) is 7.7 (which agrees well with the calculated value) and TOTM is expected to be much more hydrophobic due to the presence of a third ester group. Moreover, these measured values are more than 5 orders of magnitude lower than the calculated value.

<sup>&</sup>lt;sup>1</sup> US EPA (2000). The Use of Structure Activity Relationships (SAR) in the High Production Volume Chemicals Challenge Program, http://www.epa.gov/opptintr/chemrtk/sarfinl1.htm

The physical properties of vapor pressure and water solubility of the trimellitates are too low to measure accurately, as evidenced by the data for TOTM. Similarly the calculated log  $K_{ow}$  values are likely to be more accurate than laboratory measurements, due to high values beyond the range of applicability of the test methods. The water solubility is also so low as to make hydrolysis rate studies untenable. Thus, no further measurements of the physical properties of the trimellitates is necessary as the values calculated by QSAR models are likely to be as reliable or more reliable than the measured values.

#### **Environmental Fate**

There are data available on the degradability of TOTM. The measured abiotic degradation (hydrolysis) half-life of TOTM at pH 7 and 25°C is 17.5 days (0.05 year). The EPIWIN calculated hydrolysis half-life is 0.32 year. The atmospheric degradation half-life (hydroxyl radical attack) calculated by EPIWIN (AOP module) is 0.33 day. The measured biodegradability data on TOTM are an inherent 28 day degradation of 68% by <sup>14</sup>C-TOTM loss in a shake flask test and a degradation of 4.2% in the Japanese MITI test (OECD 301C, ready biodegradability). Since the phthalate esters are readily biodegradable, it is likely that the low result in the MITI test is due to lack of bioavailability since that test has a relatively high solids content and since TOTM is much less soluble than the corresponding phthalate diester (DEHP). By analogy with the phthalates, degradation of the trimellitates is expected to proceed through step-wise hydrolysis of the ester groups to free alcohol and mellitic acid. These metabolites, in turn, are known to be rapidly degraded. No further degradation testing is necessary.

The calculated environmental distribution of TOTM (Mackay level 1) indicates that negligible fractions of TOTM will partition to air or water, with the major fractions partitioning to soil (97.82%) and sediment (2.17%). Due to closely similar physical properties, exactly the same environmental distribution is calculated for the other trimellitates in this group. Environmental fate properties are shown in Table 2A.

## Toxicokinetics and Metabolism

Absorption and metabolism were studied for TOTM administered in corn oil by gavage in a single dose of 100 mg/kg of body weight in 4 male SD rats. Urine and feces were collected over the following 144 hour period, after which animals were sacrificed and residual carcass levels determined. About 75% of the dose was excreted in the feces, 16% in the urine as metabolites and 1.9% was expired as <sup>14</sup>CO<sub>2</sub>. Radioactivity was mostly excreted in the feces as unchanged TOTM (85% of the fecal radioactivity), with 6% as isomers of the diester and 1% as the mono-2-ethylhexyl trimellitate (MEHT). Metabolites in the urine were identified as MEHT and metabolites of 2-ethylhexanol. Less than 0.6% of the dose remained in the tissues. Elimination of <sup>14</sup>CO<sub>2</sub> was biphasic with half-lives of 4.3 and 31 hrs, and excretion of radioactivity in the urine was biphasic with half-lives of 3.4 hrs and 42 hrs. (Eastman Kodak Company, unpublished report 1984).

These data indicate that TOTM is poorly hydrolyzed and absorbed across the gastrointestinal tract. By comparison, numerous absorption and metabolism studies on DEHP indicate that DEHP is readily hydrolyzed in the gut prior to absorption, with ~50% of the DEHP dose absorbed by rodents following oral administration (Albro and Lavenhar, 1989). Hydrolysis in the gut appears to be an obligatory step for systemic absorption of phthalate esters. Thus, the relatively poor hydrolysis and systemic absorption of TOTM may in part explain the observed lower toxicity of trimellitates as compared to phthalate esters.

# **Mammalian Toxicity Data**

A summary of the available toxicity data on trimellitates is shown in Table 2B.

## **Acute Toxicity**

TOTM exhibits very limited acute toxicity with an oral  $LD_{50} > 2$  g/kg, a dermal  $LD_{50} > 20$  ml/kg (approximately 20 g/kg), and an acute inhalation  $LC_{50}$  in the range of 0.23 to 2.64 mg/L (nominal). There is, in addition, an acute oral  $LD_{50}$  value for TINTM of > 10 g/kg. Although some of these data are from older studies that may not have been fully consistent with current guidelines, these results are consistent with those from studies of phthalate esters produced from similar alcohol feedstocks. These data indicate that acute toxicity is not a concern for molecules of this type. No additional acute toxicity testing is planned for this category.

#### Repeated Dose Toxicity

TOTM was tested for repeated dose toxicity in rats. Exposure was by dietary administration at levels of 0.2, 0.67, and 2.0%. There was no effect on body weight or food consumption; liver weights in the 0.67% group were significantly increased, but this was judged to have been a spurious finding as liver weights were not increased in the 2.0% group. There was also evidence of increased metabolic enzymes and cholesterol. There were also some changes in blood parameters but these were inconsistent, and were judged to be without toxicological consequence. The NOAEL was 654 mg/kg/day based on a finding of slight peroxisome proliferation at the top dose (2%, ca. 1826 mg/kg/day). A NOAEL of 1000 mg/kg/day was similarly reported in an unpublished 28 day oral feeding study in rats (Japan Ministry of Health & Welfare, 1996). Based on these data, no additional subchronic toxicity testing is planned for this category.

# Mutagenicity

TOTM was not mutagenic in Salmonella and did not cause mutations in the HGPRT assay in CHO cells. Additionally, there was no increase in unscheduled DNA synthesis in rat hepatocytes. TOTM induced neither structural chromosomal aberrations nor polyploidy in CHL/IU cells up to the limit concentration of 5.0 mg/ml, in the absence or presence of an exogenous metabolic activation system. In addition, a wide range of phthalate esters produced from similar C8-C10 alcohol feedstocks have been evaluated

for both point mutations (Zeiger et al., 1987; Barber et al., 2000) and chromosomal aberrations (McKee, 2000) and have consistently been found to be inactive. Based on these data there is no need to conduct additional tests of trimellitates for point mutations or chromosomal aberrations.

# Reproductive/Developmental Toxicity

TOTM was studied for oral toxicity in rats in an OECD preliminary reproduction toxicity screening test at doses of 0, 100, 300 and 1000 mg/kg/day. Histopathological examination of the testes revealed decreases in spermatocytes and spermatids in males of the 300 and 1000 mg/kg groups. No effects of TOTM were detected on general appearance, body weight, food consumption, autopsy findings, and weights of the reproductive organs of both sexes, or on histopathological examination of the ovary. Except for the effects in males observed on histopathological examination, no influence of TOTM was detected regarding reproductive ability, organ weights or histopathological appearance of the ovaries, delivery or maternal behavior of dams. No effects of TOTM were detected on viability, general appearance, body weight or autopsy findings of offspring. On the basis of these findings, the NOELs of TOTM for reproductive/developmental effects were considered to be 100 mg/kg/day for males, 1000 mg/kg/day for females, and 1000 mg/kg/day for offspring.

A comparison of TOTM to DEHP indicates that TOTM is considerably less active than its diester analog. The LOAEL for DEHP-induced reproductive and developmental effects is 140 mg/kg/day and 750 mg/kg/day, respectively. In contrast, no reproductive or developmental effects were observed with TOTM at dose levels up to 1000 mg/kg/day.

Phthalate esters produced from similar C8-C10 alcohol feedstocks as used to produce trimellitates have also been extensively studied for potential reproductive and developmental effects. Phthalate esters with linear alkyl chains ≥C7 (High molecular weight phthalates), demonstrate neither reproductive nor developmental effects in rodents. Thus it is highly unlikely that the remaining trimellitates in this category will exhibit any reproductive or developmental effects. No further reproductive or developmental testing is proposed for this category.

#### **Environmental Toxicity**

A summary of the available toxicity data on trimellitates is shown in Table 2B. There are acute aquatic toxicity data available for TOTM in fish daphnia and algae. No acute toxic effects to fish (*Oryzias latipes*) were observed at the highest concentration tested (100 mg/L, NOEC > 100 mg/L). Similarly, no effects were observed in algae (*Selenastrum capricornutum*) at 100 mg/L (NOEC > 100 mg/L). The EC<sub>50</sub> for *Daphnia magna* was also above the highest concentration tested (180 mg/L). It should be noted that all of these toxicity tests were conducted at concentrations many orders of magnitude higher than the true water solubility of TOTM through the use of a chemical dispersants. The calculated values for TOTM acute toxicity also predict no effects at the limit of water

solubility. Chronic fish and daphnia exposure studies on TOTM also show no toxicity at and above its water solubility limit.

No measured aquatic toxicity data are available for the remaining members of this category. However, all of these are similar in structure to TOTM and to the higher phthalates and are expected to have the same characteristic aquatic toxicity, namely none. They are even less water soluble than the higher phthalates and like the higher phthalates, their solubility is too low to result in toxicity to aquatic organisms.

In addition, quantitative structure activity relationships (QSARs) are acceptable sources of ecotoxicity information for the evaluation for chemicals that belong to chemical classes with established QSARs.<sup>2</sup> Esters are such a class and the EPA's QSAR model "ECOSAR" may be relied upon to evaluate the potential aquatic toxicity of these trimellitate esters. No additional environmental testing is proposed for this category.

# **TEST PLAN SUMMARY**

The American Chemistry Council Phthalate Esters Panel HPV Testing Group believes that there is a sufficient amount of information available on TOTM to substantially characterize the human health effects and environmental fate and effects endpoints for the remaining members of this category under the HPV program.

Physicochemical properties and environmental fate for all category members were calculated using appropriate QSAR models, and supplemented with measured data from the literature. Due to the poor solubility of these materials, the values calculated by QSAR models are likely to be as reliable or more reliable than the measured data.

Mammalian toxicity data on TOTM is being used to characterize the potential hazards of the remaining category members. Sufficient SIDS data exists on TOTM to reliably assess acute toxicity, repeat dose toxicity, point mutations, chromosomal aberrations, and reproductive toxicity. The developmental effects of TOTM were indirectly measured in an OECD preliminary reproduction toxicity screening test. Extensive reproductive and developmental studies on phthalate esters were used as supportive information to characterize these endpoints.

Both calculated and measured values for TOTM environmental toxicity endpoints predict no effects at the limit of water solubility. As the remaining trimellitates are even less water-soluble than TOTM, their solubility is too low to result in toxicity to aquatic organisms.

No additional toxicology tests are proposed for these materials.

# Table 1. CAS Numbers and Descriptions

<sup>&</sup>lt;sup>2</sup> US EPA (2000). The Use of Structure Activity Relationships (SAR) in the High Production Volume Chemicals Challenge Program, http://www.epa.gov/opptintr/chemrtk/sarfinl1.htm.

CAS Number	CAS Number Description	Acronym
3319-31-1	1,2,4-benzenetricarboxylic acid, tris (2-ethylhexyl) ester	TOTM
27251-75-8	1,2,4-benzenetricarboxylic acid, triisooctyl ester	TIOTM
53894-23-8	1,2,4-benzenetricarboxylic acid, triisononyl ester	TINTM
67989-23-5		

Table 2. Summary of SIDS Information on Trimellitates

A. Physical/Chemical Properties of Trimellitates

(R)						7.0	Water	Photodeg	Hydrolysis	Transport (%) c			
Carbon Chain Length	CAS Number	Chemical Name	MP* (°C)	BP** (℃)	VP (hPa@25°C)	PC (log Pow)	Solubility (mg/L @25°C)	Half-life (days)	Half-life (yrs)	Soil	Air	Water	Sediment
C8	3319-31-1	tris 2-ethylhexyl (TOTM)	-46 97 c	>300 541 c	<0.0001*** 5.25E-11 c	5.94 11.59 c	3.9E-04 4.51E-08 c	0.33 с	0.05 0.32 c	97.8	3.6E - 6	2.8E - 7	2.17
C8	27251-75-8	triisooctyl ester	<0 197 c	541 c	5.25E-11 c	11.59 с	4.51E-08 c	0.35 с	0.43 с	97.8	3.64E - 6	2.8E - 7	2.17
С9	53894-23-8	triisononyl ester	<0 224 c	>300 575 c	3.17E-12 c	13.06 с	1.32E-09 c	0.31 с	0.86 с	97.8	2.74E - 7	9.61E -9	2.17
C8,C10	67989-23-5	decyl, octyl ester	<0 234 c	585 ¢	1.37E-12 c	12.79 ¢	2.78E-09 c	0.32 c	0.98 с	97.8	1.02E - 7	1.79E - 8	2.17

c = calculated data using EPWIN; all other values are derived from measurements

<sup>\* =</sup> All of these trimellitates are liquids at zero degrees C. Modeled data do not accurately reflect melting points for these substances

<sup>\*\* =</sup> Measured boiling points were determined to be >300°C at 0.66 kPa

<sup>\*\*\* =</sup> vapor pressure of TOTM 13 Pa @ 200°C

Table 2. Summary of SIDS Information on Trimellitates **B.** Toxicology Data on Trimellitates

(R) Carbon Chain Length	CAS Number	Chemical Name	Acute Oral LD50	Acute Dermal LD50	Acute Inhalation LC50	Repeated Dose Toxicity	GeneTox (Ames)	GeneTox (Chrom. Abs.)	Toxicity to Reproduction	Developmental Toxicity / Teratogenicity	Acute Fish (A) mg/L	Daphnia (B) mg/L	Algal (C) mg/L	Biodegradation %
C8	3319-31-1	tris 2-ethylhexyl (TOTM)	> 3.2 g/kg (rat, mouse)	>20 ml/kg (guinea pig) >2.0 ml/kg (rabbit)	<2.64 mg/L (rat, nominal)	NOAEL (rat, dietary) 654 mg/kg/day	Negative	Negative (CHL/IU cells)	NOAEL (rat, oral) 1000 mg/kg/day	NOAEL (rat, oral) 1000 mg/kg/day (3)	>100	>180	>100	68-71 (1) 4.2 (2)
C8	27251-75-8	Triisooctyl ester	R	R	R	R	R	R	R	R	R	R	R	R
C9	53894-23-8	Triisononyl ester	> 10 g/kg (rat)	R	R	R	R	R	R	R	R	R	R	R
C8, C10	67989-23-5	decyl, octyl ester	R	R	R	R	R	R	R	R	R	R	R	R

R = read-across to TOTM

- Footnotes: A) Japanese Medaka (Oryzias latipes), 96 hr LC50 & NOEC
  - B) Daphnia magna, 48-hr EC50
  - C) Selenastrum capricornutum, 72-hr EC50 & NOEC
  - (1) Inherent biodegradation by Shake Flask Method
  - (2) Ready biodegradation by MITI method (OECD 301C)
  - (3) OECD Preliminary reproduction toxicity screening test; indirect measure of develomental effects

#### References\*

Albro, P.W. and S.R. Lavenhar (1989). Metabolism of di(2-ethylhexyl)phthalate. Drug Metabolism Reviews, 21(1): 13-34.

Wilson, A., (1996). Plasticizers – Selection, Applications and Implications. Rapra Review Reports 8:15-16.

EPA (1981). Chemical Technology and Economics in Environmental Perspective. Task VI. A survey of plasticizers, epoxies, linear polyesters, and trimellitates. Prepared for Environmental Protection Agency by Midwest Research Inst. EPA

Klimisch, H. J., M. Andreae, and U. Tillmann. (1997). A Systematic Approach for Evaluating the Quality of Experimental Toxicological and Ecotoxicological Data. Regulatory Toxicol. and Pharmacol. 25:1-5.

CRC Handbook of Chemistry and Physics, 81st editing (2000). CRC Press LLC, Boca Raton, FL.

US EPA (2000). The Use of Structure Activity Relationships (SAR) in the High Production Volume Chemicals Challenge Program, http://www.epa.gov/opptintr/chemrtk/sarfinl1.htm

\*The list of references is not a comprehensive bibliography of all of the trimellitate literature, merely a series of papers that illustrate key points made in the text. The information in these papers also supplements the robust summaries developed for toxicology studies of listed substances in tests addressing specific SIDS endpoints.

# Appendix 1: Literature Search

# 3319-31-1 1,2,4-benzenetricarboxylic acid, tris (2-ethylhexyl)ester

#### Reviews

Japan dossier and robust summary for tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate. (DRAFT unpublished report, 2001).

Japan Ministry of Health & Welfare (1996). Toxicity Testing Reports of Environmental Chemicals, Vol. 4. http://wwwdb.mhlw.go.jp/ginc/dbfile1/file3319-31-1.html.

David, R. et al., (2001). Esters of aromatic mono-, di-, and tricarboxylic acids, aromatic diacids and di-, tri-, or polyalcohols. <u>In</u>: Patty's Toxicology, Fifth edition, Vol. 6, Bingham E., B. Cohrssen and C.H. Powell (eds.), John Wiley & Sons, Inc. pp. 635-932.

Environmental and Health Assessment of Alternatives to Phthalates and to Flexible PVC (2001). Danish EPA, Environmental project No. 590.

# Phys/Chem Data

IUCLID, International Uniform Chemical Information Database, European Chemicals Bureau, Ispra, Italy - Feb 2000.

Japan dossier and robust summary for tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate. (DRAFT unpublished report, 2001).

## **Ecotoxicity Data**

Japan dossier and robust summary for tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate. (DRAFT unpublished report, 2001).

# **Mammalian Toxicity**

BIBRA, The British Industrial Biological Research Association. (1985). 28-day toxicity study with tri(2-ethylhexyl)trimellitate in the rat. Report to the Chemical Manufacturers Association. Report No. 0496/1/85.

Eastman Kodak Company, Rochester NY (1971). Tri(2-ethylhexyl)trimellitate. Acute oral toxicity. Unpublished report.

Eastman Kodak Company, Rochester NY (1971). Tri(2-ethylhexyl)trimellitate. Acute dermal toxicity. Unpublished report.

Eastman Kodak Company, Rochester NY (1984). Absorption and Metabolism of (hexyl-2-<sup>14</sup>C)tris-(2 ethylhexyl) trimellitate in the rat. OTS 42040. Doc. ID 408465031.

Eastman Kodak Company, Rochester NY (1971). Tri(2-ethylhexyl)trimellitate. Acute inhalation toxicity. Unpublished report.

Japan Ministry of Health & Welfare (1996). Toxicity testing Reports of Environmental Chemicals, Vol. 4. http://wwwdb.mhlw.go.jp/ginc/dbfile1/file/file3319-31-1.html.

Zeiger, E., B. Anderson, S. Haworth, T. Lawlor and K. Mortelmans. (1988). Salmonella mutagenicity tests. IV. Results from the testing of 300 chemicals. *Environmental and Molecular Mutagenesis* 11(12):1-158.

# 27251-75-8 1,2,4-benzenetricarboxylic acid, triisooctyl ester

No relevant studies found.

# 53894-23-8 1,2,4-benzenetricarboxylic acid, triisononyl ester

Esso Research and Engineering Company (1969). Acute Oral Administration of MRD-69-31 in Rats. Unpublished Report.

# 67989-23-5 1,2,4-benzenetricarboxylic acid, decyl octyl ester

No relevant studies found.

# SIDS INITIAL ASSESSMENT PROFILE

CAS NO.	3319-31-1	
CHEMICAL NAME	Tris(2-ethylhexyl)benzene-1,2	,4-tricarboxylate
Structural formula	CH₃CH₂CH₂CH2CH2OOC CH₂CH3	CH2CH3 COOCH2CHCH2CH2CH2CH3 COOCH2CHCH2CH2CH2CH3
	ow priority for further work.	DEST WIG
SUMMARY CONCLUSION Human health	9: 56 ()	

Acute toxicity of TOTM is low,  $LD_{50} > 2,000$  mg/kg in rats. In the irritation-test for animals, this substance is slightly irritating to the skin and the eyes. Sensitization test on guinea pig showed no sensitization. Oral study in rats conducted for 28 days at doses of 0(0), 0.2(184), 0.67(650), 0.67(650), 0.67(650), 0.67(650), 0.67(650), 0.67(650), 0.67(650), 0.67(650), 0.67(650), 0.67(650), weights between control and TOTM. There were no statistically significant differences in body weights between control and treated groups in the following: hemoglobin concentration (lower in both sexes, 0.67 or 0.67 o

Preliminary reproductive toxicity screening test reveals moderate decrease of offermatocytes and spermatids in males at 100mg/kg/day. From these two test results, he NOAELs for repeated or all toxicity were considered to be 100 mg/kg/day for male rats. The NOAELs for reproductive/ developmental toxicity were considered to be 1,000 mg/kg/day for female rats and for offspring. TOTM was evaluated its genotoxicity by many assay systems. It was neithermutagenic in bacteria nor clastogenic in mammalian cells in vitro. All other in vitro and in vivo assays gave negative results. It is concluded that TOTM is not genotoxic in vitro and in vivo. The reported results of carcinogenecity was invalid.

Absorption and metabolism were studied for <sup>14</sup>C labeled TOTM and about 75% of the dose was excreted unchanged in the feces, 16% in the urine as metabolites and 1.9% was expired as <sup>14</sup>CO<sub>2</sub>.

#### Environment.

The Mackay level III fugacity Model was employed to estimate the environmental distribution of TOTM in air, water, soil and sediment. If released to air, TOTM will exist solely in the particulate phase in the ambient atmosphere. If released to soil, TOTM is not expected to have mobility. If released into water, TOTM is expected to adsorb to suspended solids and sediment in water.

TOTM has to be considered as weakly toxic against aquatic organisms. The substance is not readily biodegradable. Measured BCF of this chemical is reported as less than 1 to 2.7 in carp for 6 weeks, which suggest that bioconcentration in aquatic organisms is much lower than the value estimated from logPow(=5.94). The toxicity data to aquatic plants (algae; Selenastrum capricornutum) was >100 mg/L for EC<sub>50</sub> (72hr) and NOEC (72hr). The acute toxicity data in fish (medaka, Oryzias latipes) were >100 mg/L (96h, LC<sub>50</sub> and NOEC) and >75 mg/L (14d, LC<sub>50</sub> and NOEC). In Daphnia magna, acute toxicity was >180mg/L (48hr: EC<sub>50</sub>) and chronic toxicity was 55.6mg/L (21d, reproduction NOEC). All these data were obtained in supersaturated solution with the aid ofsolubilizer (HCO-40). The test solution was considered to be homogeneous substantially. Another chronic toxicity data inDaphnia magna (NOEC >0.082mg/L) was reported. Though this value is lower than the saturation point, the observed concentration data was less reliable. Assessment factor of 100was chosen to determine the lowest PNEC. Thus, calculated PNEC (=0.00082 mg/L) of TOTM is closely to the value of one hundredths (assessment factor) of saturation point. From these toxicity data, it is difficult to decide the exact PNEC, but we are sure of the practical safety of TOTM against aquatic organisms.

#### Exposure

TOTM is manufactured as the plasticizer of PVC applications.

The production volume of TOTM in Japan is approximately 20,000 tonnes/year and also, there are 5 manufacturers in Japan. Estimated global production is 40,000-100,000 tonnes/year. This substance is produced in closed system and mainly used asplasticizer for PVC electrical cable and wire. And so, this substance has been already blended to the compound asplasticizer, so it is not expected that downstream users or consumers of electric wire industry may expose to this substance.

Occupational exposure may occur through dermal contact and inhalation of mist. The process is constructed by closed system and workers wear protective gloves and goggles during the operation, so significant exposure is not expected.

# NATURE OF FURTHER WORK RECOMMENDED

No recommendation

**FULL SIDS SUMMARY** 

	SIDS SUMMARY		I may a series	
	O: 3319-31-1	SPECIES	PROTOCOL	RESULTS
	CAL-CHEMICAL	1		
2.1	Melting Point		OECD TG 102	<-50 °C (223 K)
2.2	Boiling Point		Other (unknown)	283 °C (at 4 hPa)
2.3	Density	}	Other (unknown)	0.987-0.990 g/cm³ at 20 °C
2.4	Vapour Pressure		OECD TG 104	< 2.8 x 10 <sup>-4</sup> Pa at 100 °C
2.5	Partition -		OECD TG 107	5.94 at 25 °C
	Coefficient			
1	(Log Pow)			
2.6A.	Water Solubility		OECD TG 105	0.13 mg/L at 25 °C
В.	pH	Į		None
	pKa	•		None
2.12	Oxidation:			None
	Reduction			
	Potential Potential			
	RONMENTAL			
	AND PATHWAY			
3.1.1	Photodegradation			None
3.1.2	Stability in Water		OECD TG 111	Stable at pH 4 at 50°C
				T <sub>1/2</sub> =17.5 days at pH 7 at 25°C
				T <sub>1/2</sub> =11.9 days at pH 9 at 25°C
3.2	Monitoring Data			None
3.3	Transport and		Calculated	(Release 100% to air)
	Distribution		(Level III	Air Water Soil Sediment
			Fugacity Model)	19.6% 4.7% 66.2% 9.5%
				(Release 100% to water) Air Water Soil Sediment
				1
		ļ ·		0.0% 32.7% 0.1% 67.2% (Release 100% to soil)
			}	Air Water Soil Sediment
		•	1	0.0% 0.0% 100% 0.0%
			(local exposure)	PEClosal = None
3.5	Biodegradation		OECD TG 302C	4.2 % after 28 days
3.7	Bioaccumulation		OECD TG 305C	BCF=1-2.7(Conc. 0.2 mg/L)
	OXICOLOGY			B - /
4.1 A	Acute Toxicity to		OECD TG 203	LC <sub>50</sub> (96 hr) > 100 mg/L
	Fish	Oryzias		
4.1 B	Prolonged	Latipes	OECD TG 204	LC <sub>50</sub> (14 day) > 75 mg/L
``- ~	Toxicity to Fish	Oryzias latipes		NOEC(14 day) > 75 mg/L
				LOEC(14 day) > 75 mg/L
4.2	Acute Toxicity to		OECD TG 202	EC <sub>50</sub> (24 hr) > 180 mg/L
<b>\</b>	Aquatic	Daphnia magna		EC <sub>s0</sub> (48 hr) > 180 mg/L
	Invertebrates			NOEC > 180 mg/L
	(Daphnia)			LOEC > 180 mg/L
4.3	Toxicity to	Selenastrum	OECD TG 201	$EC_{50}$ (72 hr) > 100mg/L
	Aquatic Plants	Capricornutum		NOEC(72 hr) > 100mg/L
[]	e.g. Algae	ATCC22662		)
4.5.1	Chronic Toxicity			None
	to Fish			1
<b>B</b>	** * ****	i	1	§

11	1		•	
4.5.2	Chronic Toxicity to Aquate Invertebrates (Daphnia)	Daphnia magna	OECD TG 211	NOEC(21d, reproduction) = 55.6 mg LOEC(21d, reproduction) > 100mg/L EC <sub>50</sub> (21d, reproduction) > 89.1mg/L LC <sub>50</sub> for parental Daphnia(21d) > 100 mg/L NOEC=0.0082 (21d, reproduction, parent Daphnia mortality)
4.6.1	Toxicity to Soil Dwelling Organisms			None
4.6.2	Toxicity to Terrestrial Plants			None
4.6.3	Toxicity to Other Non- Mammalian Terrestrial Species			None
TOYK	(Including Birds)			
5.1.1	Acute Oral Toxicity	Rat	OECD TG 401	LD <sub>50</sub> > 2,000 mg/kg (for both sexes)
5.1.2	Acute Inhalation Toxicity	Rat	Other	2,600 mg/m³ (4hr)
5.1.3	Acute Dermal Toxicity	Rabbit	Other	$LD_0 > 2.0 \text{ mL/kg}$
5.2.1 5.2.2 5.3 5.4 5.5	Skin Irritation Eye Irritation Skin Sensitisation Repeated Dose Toxicity Genetic Toxicity In Vitro	Rabbit Rabbit Guinea pig Rat	Other Other OECD TG 406 OECD TG 421	Slightly irritating Slightly irritating Not sensitizing NOAEL = 100 mg/kg bw LOAEL = 300 mg/kg bw
А.	Bacterial Test	S.typhimurium, E. coli	Japanese Guideline and OECD TG 471 & 472	- (With metabolic activation) - (Without metabolic activation)
В.	Non-Bacterial In Vitro Test	CHL/IU cells	Japanese Guideline	- (With metabolic activation) - (Without metabolic activation)
5.6	Genetic Toxicity In Vivo	Mouse ·	Other	No valid data
5.7 5.8	Carcinogenicity Toxicity to Reproduction	Mouse Rai	Other OECD TG 421 Preliminary toxicity screening test	No valid data  NOAEL = 100 mg/L (male)  NOAEL = 1,000 mg/L (female)  NOAEL = 1,000 mg/L (Offspring)
5.9	Developmental Toxicity/			None
5.11	Teratogenicity Experience with Human Exposure	·	·	None

[Note] Data beyond SIDS requirements can be added if the items are relevant to the assessment of the chemical, e.g. corrosiveness/irritation,carcinogenicity.

# SIDS Initial Assessment Report for 13th SIAM

(November 6-9, 2001)

Chemical Name: Tris(2-ethylhexyl) benzene-1,2,4-tricarboxylate

CAS No:

3319-31-1

Sponsor Country: Japan

National SIDS Contact Point in Sponsor Country: Mr. Koji Tomita, Ministry of Foreign Affairs, Japan

# HISTORY:

The original IUCLID documents were prepared by European Commission. Dainippon Ink and Chemicals Inc., Japan reviewed the documents after incorporation of Japanese testing results.

#### COMMENTS:

ICCA Initiative work led by Dainippon Ink and Chemicals Inc., Japan

Deadline for circulation:

Date of Circulation:

# SIDS INITIAL ASSESSMENT REPORT (SIAR)

Tris(2-ethylhexyl) benzene-1,2,4-tricarboxylate

1. IDENTITY

IUPAC Name:

Tris(2-ethylhexyl) benzene-1,2,4-tricarboxylate

CAS Number:

3319-31-1

Molecular formula:

 $C_{33}H_{54}O_6$  (MW=546.79)

Structural formula:

 $\begin{array}{c} \text{CH}_2\text{CH}_2\\\\ \text{COOCH}_2\text{CHCH}_2\text{CH}_2\text{CH}_2\text{CH}_2\\\\ \text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\\\\ \text{CH}_2\text{CH}_3\\\\ \text{CH}_2\text{CH}_3\\\\ \end{array}$ 

Synonym:

**TOTM** 

Tris(2-ethylhexyl) trimcllitate

Benzene-1, 2, 4-tricarboxylic acid tris-(2-ethylhexyl) ester

Purity:

>99.5%

Impurity:

Di(2-ethylhexyl) phthalate (DEHP) < 0.1%

Water

Additives:

None

Table 1. Physical and Chemical Properties

	India 11 my sicurate Chemical 110 perces						
Items	Protocol	Results					
Melting Point	OECD TG 102	< -50°C					
Boiling Point	Unknown	283°C(4 hPa)					
Density	Unknown	0.987 - 0.990 g/cm³(20°C)					
Vapor pressure	OECD TG 104	<2.8 x 10 <sup>-1</sup> Pa(100°C)					
Partition Coefficient (LogPow)	OECD TG 107	5.94 (25°C)					
Water Solubility	OECD TG 105	0.13 mg/ L (25°C)					

# 2. GENERAL INFORMATION ON EXPOSURE

The production volume of TOTM in Japan is approximately 20,000 tonnes /year and also, there are 5 manufacturers in Japan. Estimated global production is 40,000 – 100,000 tonnes/year. TOTM is produced in closed system andmainly used as plasticizer for PVC electrical cable and wire especially for high temperature application. TOTM is no source of potential release to the environment except for samplingand maintenance of the production activities.

## 2.1 Environmental Fate

Based upon the biodegradation measurement, the substance is not readily biodegradable. TOTM achieved 4.2 percent of its theoretical BOD using an activated studge inoculum during a 4 weeks incubation in a single screening study.

The Mackay levell!! fugacity model was employed to estimate the environmental distribution of TOTM in air, water, soil and sediment. The calculation results are shown in Table 2.If released to air, an estimated vapor pressure of less than 2.8 x 10<sup>-4</sup> Pa at 100°C indicates TOTM will exist solely in the particulate-phase in the ambient amosphere. Particulate-phase TOTM is removed from the atmosphere by wet and dry deposition. If released to soil, TOTM isnot expected to have mobility based upon the fugacity model calculation Volatilization from soil surfaces is not expected to be an important environmental fate process based on the stimated vapor pressure of this substance. If released into water, TOTM is expected to adsorb to suspended solids and sediment in water based upon the fugacity model calculation. [Dainippon Ink and Chemicals, Inc. (2001)]

Hydrolysis may be an important environmental fate process based orestimated hydrolysis half-lives of 17.5 and 11.9 days at pH 7 and 9, respectively. Measured BCF values of less than 1 to 2.7 in carp suggest that bioconcentration in aquatic organisms is low.

Table 2. Predicted distribution of TOTM using Fugacity level III (%)

Compartment	Release 100% w air	Release 100% to water	Release 100% to soil
Air	19.6	0.0	0.0
Water	4.7	32.7	0.0
Soil	66.2	0.1	100.0
Sediment	9.5	67.2	0.0

# 2.2 Human Exposure

# 2.2.1 Occupational exposure

The substance is produced and used in closed system. So, occupational exposure is limited in the case of sampling and maintenance at the production facilities. Moreover, the exposure time is very short. The major rout of occupational exposure is inhalation and demal.

The atmospheric concentration was measured at two production sites in Japan. The monitoring data are shown in Table 3. The maximum exposure level is estimated according to working schedules as follows. From Table 3, if a single worker (Body weight; 70 kg, respiratory volume; 1.25 m³/hour) is assigned to implement all daily operation without protection, the daily intake (EHE inh) is calculated as 1.77 x 10³ mg/kg/day as the worst case. On the other hand, a single worker (surface area of exposed skin 840 cm² for hands) daily dermal dose (EHE der) is calculated as 2.47 mg/kg/day based on below calculation as the worst case without protection. Workers wear protective gloves and goggles during the operation, so significant exposure is not expected.

Table 3. Available workplace monitoring data for TOTM (EHE inh)

Occupation	Frequency Times/day	Duration Hr	Working hr/day	Max concentration mg/m³	EHE inh mg/kg/day	Reference
Sampling	5	0.017	0.085	0.210	3.19x10 <sup>-4</sup>	ЛЅНА,
Analysis	5	0.067	0.335	0.053	3.17x10 <sup>-4</sup>	Japan
Charge to drum	1	0.833	0.833	0.076	1.13x10 <sup>-3</sup>	(2001)
Total	11	-	1.253	-	1.77x10 <sup>-3</sup>	

EHE inh: Estimated Human Exposure for inhalation

Calculation: EHE der = (Cder \* T \* S \* t ) /W

EHE der. Estimated Human Exposure for dermal

Cder = 990 mg/cm<sup>3</sup> (Rate in product contacted by worker)

T = 0.01 cm (Thickness of substance)

 $S = 840 \text{ cm}^2$  (Surface area of exposed skin) for hand

t = 0.0208 day/day (Exposure time per day; 10min/8Hr, [1day = 8Hr] assumed)

W = 70 Kg (body weight)

# 2.2.2 Consumer exposure

Usually, this substance has been already blended to the compound asplasticizer, so it is not expected that downstream users or consumers of electric wire industry may expose to this substance.

#### 3.HUMAN HEALTH HAZARDS

#### 3.1 Effects on Human Health

#### 3.1.1 Toxicokinetics and metabolism

Absorption and metabolism were studied for TOTM(14C-labeled on the 2-carbon atom of 2-ethylhexyl group) mixed with corn oil and administered by gavage in a single dose of 100 mg/kg of body weight in 4 male SDrats. About 75% of the dose was exercted unchanged in the feces, 16% in the urine as metabolites and 1.9% was expired as  $^{14}$ CO<sub>2</sub>. Radioactivity was excreted in the feces as unchanged TOTM (85% of the fecal radioactivity), mono- and di(2-ethylhexyl) trimellitate(MOTM and DOTM, respectively), and as unidentified polar metabolites. Metabolites in the urine were identified as MOTM and metabolites of 2-ethylhexanol less than 0.6% of the dose remained in the tissues. Elimination of  $^{4}$ CO<sub>2</sub> was biphasic with half-lives of 4.3 and 31 hrs, and excretion of radioactivity in the urinewas biphasic with half-lives of 3.4 hrs and 42 hrs. [Eastman Kodak Company]

# 3.1.2 Acute toxicity

Acute toxicity data are mainly reported for rat, mice and rabbits. We could find 12 acute toxicity data for animals (oral(6), inhalation(1), IP(2) and dermal(3)) test data, and one (oral) study (MHW, Japan (1996)) and two (oral and dermal) studies Nuodex Inc. (1981), Nuodex Inc. (1982c) ) were conducted by the method of OECD TG and similar method to OECD TG, respectively.

The data, which we feel informative to evaluate the acute toxicity, are listed in Table 4.

Table 4. Summary of effects of TOTM on animals (Acute Toxicity)

Route	Animals	Values	Type	References
Oral	Rat	>2000 mg/kg	$LD_{\mathbf{x}}$	MHW, Japan (1996)
	Rat	>5000 mg/kg bw	$LD_0$	Nuodex Inc.(1981)
Inhalation	Rat	>2600 mg/m <sup>3</sup>	LCa	Nuodex Inc.(1982b)
Dermal	Rabbit	>2 m1/kg	LDg	Nuodex Inc(1982c)
	Rabbit	>1970 mg/kg bw	$LD_0$	Tenneco Chemicals(1981))
I.P.	Rat	>3200 mg/kg bw	LDsa	Eastman Kodak (1983)
	Mouse	>3200 mg/kg bw	LD <sub>so</sub>	Eastman Kodak (1983)

It can be concluded that acute toxicity (Oral) of TOTM is LD<sub>10</sub> > 2000 mg/kg in rat.

# 3.1.3 Repeated dose toxicity

Among the eight available data, four were conducted under the GLP. Three studies were considered to be key study.

The first study was the oral study by CMA(1985). The subchronic toxicity of TOTM administered orally in the diet togroups of 5 male and 5 femaleFischer 344 rats at levels of O(0), 0.2(184), 0.67(650), 2.0(1826) % (mg/kg bw/day) for 28 days was determined There were no statistically significant differences in body weights between control and OTM treated groups. There was a significant difference between control and treated groups in theollowing: absolute and relative liver weights (higher in both sexes at all levels except 0 or 0.2%) serum albumin (higher in both sexes at 0.67 or 2.0%), serum cholesterol levels (higher in males at 0.67 or 2.0%). Liver biochemistry revealed statistically significant differences between treated and control groups as indicated by paimitoyl CoA oxidation (increased in both sexes at 20% and males at all dose levels), and catalase activity (increased in males at 2.0%). So, the NOAEL for repeated dose toxicity is considered to be 184 mg/kg and the LOAELis 650 mg/kg for both sexes.

The second study was the oral study by MHWJapan(1996). No test substances related changes were noted in terms of clinical signs, body weight, food consumption, and hematology, blood examination, urinalysis, and pathological findings. So, the NOEL for repeat dose toxicity is considered to be 1,000 mg/kg/day for both sexes.

The third study was the OECD preliminary reproduction toxicity screening test by MHW Japan(1998). Gavage study in SD rats conducted at doses of 100, 300 and 1,000 mg/kg/day (Male; 46days, Female; from 14days before mating to day 3 of lactation) of TOTM. The decreases in spermatocytes and spermatids in males was observed for 300 and 1.000 mg/kg groups by histopathological examination. No effects on general appearance, body weight, food consumption, autopsy findings, weights of the reproductive organs of both sexes, or histopathlogical features of the ovary were detected. So, the NAOEL is considered to be 100 mg/kg/day for males, and 1,000 mg/kg/day for females.

There is no available information on human toxicity.

#### Conclusions:

The NOAEL and the LOAEL for repeated oral toxicity are considered to be 100 and 300 mg/kg/day for rats, respectively.

# 3.1.4 Genotoxicity / Mutagenicity

We can find five reports for Ames Tests. One MHW, Japan: 1996) is conducted under GLP and others are not. The study of MHW is considered to be a key study.

TOTM has been investigated in vitro tests. This substance did not induce gene mutation in bacterial system (MHW, Japan: 1996), and chromosomal aberration in mammalian cultured cells (MHW, Japan: 1996), with and without an exogenous metabolic activation system. Among these studies, MHW study was identified to be a key study because it was well conducted in reported.

Reverse gene mutation assay was conducted by OECD TG 471 and 472, using pre-incubation method. TOTM was not mutagenic in Salmonella typhinurium TA100, TA1535, TA98, TA1537 and Escherichia coli WP2 uvrA at concentration of up to 5000 ug /plate, with or without an exogenous metabolic activation system (MHW, Japanit 996).

Chromosomal aberration test by OECD TG 473 was conducted in cultured Chinese hamster lung (CHL/IU) cells. Structural chromosomal aberrations and polyploidy were not inducedup to a maximum concentration of 5.0 mg/mL on continuous treatment, and with Short-term treatment, with and without an exogenous metabolic activation system (MHW, Japan: 1996).

And all other test results (HGPRT assay, Unscheduled DNA synthesis, Dominant Lethal Assay for example) shows that TOTM is not genetoxic

#### Conclusions:

This substance is considered to be not genotoxic with and without an exogenous metabolic activation system in bacterial test and chromosomal aberration testin vitro.

# 3.1.5 Carcinogenecity

One brief report states only that tests in mice, with a propensity to form pulmonary adenomas, were negative for TOTM, unlike those using urethane. The carcinogeneous tests revealed that the chemical is negative but test result was invalid.

# 3.1.6 Reproduction/developmental toxicity

The OECD Preliminary Reproduction Toxicity Screening Test was performed. [MHW, Japan: 1998]. This study was identified to be well conducted and eported.

Gavage study in SD rats conducted at doses of 100,300 and 1,000 mg/kg/day (Male; 46days, Female; from 14days before mating to day 3 of lactation) of TOTM.

Histopathological examination of the testes revealed decreases inpermatocytes and spermatids in males of the 300 and 1,000 mg/kg groups. No effects of TOTM were detected on general appearance, body weight, food consumption, autopsy findings, and weight of reproductive organs of both sexes, or on histopathological examination of the ovary. On the basis of these findings, the NOELs of TOTM for repeat dose toxicity are considered to be 100 mg/kg/day for males, and 1,000 mg/kg/day for females.

Except for the effects in males observed on histopathological examination, no influence of this substance was detected regarding reproductive ability, organ weights or histopathological appearance of the ovaries, delivery or maternal behavior dams. No effect of TOTM were detected on viability, general appearance, body weight or autopsy findings of offspring. On the basis of these findings, the NOELs for reproductive / developmental toxicity were considered to be 100 mg/kg/day for male rats, 1,000 mg/kg/day for female rats, and 1,000 mg/kg/day for offspring.

#### Conclusions:

The NOELs for reproductive/developmental toxicity were considered to be 100 mg/kg/day for male rats, 1,000 mg/kg/day for female rats, and 1,000 mg/kg/day for offspring, respectively.

#### 3.1.8 Other: Irritation and sensitization

Six and three results are reported for skin and eye irritation test, respectively. All these test results showed that TOTM is slightly irritating to the skin and the eye.

Sensitization test on guinea pig using OECD/TG 406 (Tenneco Chemicals, 1981) showed "no sensitization".

#### 3.2 Initial Assessment for Human Health

Acute toxicity of TOTM is considered to be LD<sub>a</sub>>2000 mg/kg in rat.

In the irritation-test for animals, TOTM is slightly irritating to the skin and the cyc.

Sensitization test on guinea pig using OECD/TG 406 showed "no sensitization".

The NOAEL and the LOAEL for repeated oral toxicity are considered to be 100 and 300 mg/kg/day for rats, respectively

The NOELs for reproductive/developmental toxicity were considered to be 100 mg/kg/day for male rats, 1,000 mg/kg/day for female rats, and 1,000 mg/kg/day for offspring, respectively.

This substance is not genotoxic with and without an exogenous metabolic activation system in bacterial test and chromosomal aberration testin vitro.

TOTM produces the same spectrum of morphological and biochemical change in the rat liver as DEHP. TOTM, however, was much less potent in its action, with a dietary level of 2.0%, causing less peroxisome proliferation and peroxisome-associated enzyme induction than 0.67% DEHP. Also, the level of peroxisome induction in rats given TOTM is less than in those receiving a metabolically equivalent dose of 2-ethylhexanol. Furthermore, on a molar basis, effects were lower than with DEHP. An effect of MEHP, a metabolite of DEHP, was not seen with TOTM. [The British Industrial biological Research Association (1985), EPA OTS0510637(1985), JOHN R. HODGSON. (1987)]

In addition, recently studies have determined that rodents (rats) are susceptible toperoxisome proliferation. After all, these results suggest that the effect of DEHP on liver are markedly different between other species (marmosets) and rodents (rats). Yoshimasa Kurata et al. (1998)] Therefore, DEHP was downgraded from Group 2B to Group 3 by the IARC Monographs Working Group. (February 2000) Group 3 is "cannot be classified as to its carcinogenicity to humans".

#### 4. Hazards to the Environment

# 4.1 Aquatic Effects

TOTM has to be considered as weakly toxic against aquatic organisms. Aquatic effects were tested and results are summarized in Table 5. As the lowest acute and chronic toxicity data, EC<sub>so</sub> (>100 mg/L, ,72hr) of Selenastrum capricornutum ATCC22662 and NOEC (55.6 mg/L, 21day) of Daphnia magna were adopted, respectively.

Table 5. Summary of effects of TOTM on aquatic organisms

Organism	Test duration	Result (mg/L)	Reference
Algae Selenastrum capricornutum ATCC22662	72 hr	EC <sub>50</sub> >100 NOEC>100	EA, Japan
Invertebrates			
Daphnia magna	24 hr	EC <sub>50</sub> >180	EA, Japan
	48 hr	$EC_{50} > 180$	
		NOEC>180	
	48hr	EC <sub>50</sub> >1	ICI 1990
	21 day	EC <sub>50</sub> ≃89.1 NOEC=55.6	EA, Japan

	21 day	NOEC=0.082	CMA (1985)
Fish Oryzias latipes	96 hr	LC <sub>50</sub> >100	EA, Japan
	14 day	LC <sub>m</sub> >75 NOEC>75	EA, Japan

As the acute toxicity data,  $EC_{\infty}$  (>100 mg/L, 72hr) of Selenastrum capricornutum ATCC22662 and  $EC_{50}$  (180 mg/L, 48hr) of Daphnia magna were adopted, respectively. As the chronic toxicity data of Daphnia magna and the prolonged toxicity data of fish Oryzias latipes), NOEC =0.082mg/L (21days) [CMA; 1985] and NOEC=75mg/L (14days) [EA Japan] were adopted, respectively. All those data in supersaturated solution, which was considered to be homogeneous substantially, was obtained with the aid of solubilizer (HCO-40). Though the observed concentration data was less reliable, one chronic toxicity data (NOEC >0.082mg/L) was reported in a lower concentration than saturation point.

Two other acute (ICI 1990) and chronic(EA Japan) data would be helpful for evaluation of the toxicity for Daphnia magna. These tests were conducted in a supersaturated solution.

Assessment factor of 100 was chosen to determine the lowest PNEC. Thus, calculated PNEC (=0.00082 mg/L) of TOTM is closely to the value of one hundredths (assessment factor) of saturation point. From these toxicity data, it is difficult to decide the exact PNEC, but we are sure that TOTM is practically non-toxic against aquatic organisms.

4.2 Terrestrial effects

There is no available information.

#### 4.3 Initial assessment for the Environment

Hydrolysis may be an important environmental fate process based orestimated hydrolysis half-lives of 17.5 and 11.9 days at pH 7 and 9, respectively. The substance is not readily biodegradable. Measured BCF values of this chemical is reported as less than 1 to 2.7 in carp for 6 weeks, which suggest that bioconcentration in aquatic organisms is much lower than the value estimated from logPow(=5.94). If released into surface water water, TOTM is expected to adsorb to suspended solids and sediment based upon the fugacity model calculation. The sediment toxicity data was not available, and will need to assess when obtained.

#### 5. Conclusions and recommendations

#### 5.1 Conclusions

# Exposure (Physical/chemical property, production, use and distribution)

TOTM is manufactured as the plasticizer of PVC application.

The production volume of TOTM in Japan isapproximately 20,000 tonnes/year and also, there are 5 manufacturers in Japan. Estimated global production is 40,000 – 100,000 tonnes/year. TOTM is produced in closed system andmainly used as plasticizer for PVC electrical cable and wire. And so, this substance has been already blended to the compound aplasticizer, so it is not expected that downstream users or consumers of electric wire industry may expose to this substance.

Occupational exposure may occur through dermal contact and inhalation of vapor. The process

is constructed by closed system and workers wear protective gloves and gggles during the operation, so significant exposure is not expected.

In case of disposal, this substance would be incinerated with following all regulations. Therefore, it is not significant released to the environment

#### Human health

Acute toxicity of TOTM is low, LD<sub>50</sub> >2,000 mg/kg in rats. In the irritation-test for animals, this substance is slightly irritating to the skin and the eyes. Sensitization test on guinea pig showed "no sensitization". Oral study in rats conducted for 28 days at doses of 0(0), 0.2(184), 0.67(650). 2.0(1826) % (mg/kg bw/day) of TOTM. There were no statistically significant differences in body weights between control and TOTM treated groups. There was a significant difference between control and treated groups in thefollowing: hemoglobin concentration (lower in both sexes, 0.67 or 2.0% TOTM), leucocyte counts (higher in males at 0.67 or 2.0%), absolute and relative liver weights (higher in both sexes at all levels except 0 or 0.2%), serum albumin (higher in both sexes at 0.67 or 20%), serum cholesterol levels (higher in males at 0.67 or 2.0%), serum urea (higher in males at 2.0%), serum lipids (decreased in females a0.2%). Liver biochemistry revealed statistically significant differences between treated and control groupses indicated by palmitoyl CoA oxidation (increased in both sexes at 20% and males at all dose levels), and catalase activity (increased in males at 2.0%). Therefore, the NOAEL and the LOAEL for repeated oral toxicity were considered to be100 and 300 mg/kg/day for male rats. The NOELs for reproductive/developmental toxicity were considered to be 1,000 mg/kg/day for female rats and for offspring.

TOTM is not genetoxic/mutagenic in bacterial and mammalian cell tests in vitro tests. The carcinogenecity tests revealed that the chemical is negative but test result was invalid.

#### Environment

The Mackay levelll fugacity model was employed to estimate the environmental distribution of TOTM in air, water, soil and sediment. If released to air, TOTM will exist solely in the particulate-phase in the ambient atmosphere. If released to soil, TOTM is not expected to have mobility. If released into water, TOTM is expected to adsorb to suspended solids and sediment in water.

Measured BCF of values of less than 1 to 2.7 in carp suggest that bioconcentration in aquatic organisms is low.

As the lowest acute and chronic toxicity data, EC<sub>50</sub> (>100 mg/L, 72hr) of Seletiastrum capricornutum ATCC22662 and NOEC (0.082mg/L, 21day) of Daphnia magna were adopted, respectively. Assessment factor of 100 was chosen to both acute and chronic toxicity data to determine PNEC Thus, PNEC of TOTM is 0.00082mg/L.

## 5.2 Recommendations

The chemical is currently of low priority for further work.

#### 6. References

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JOHN R. HODGSON. Results of peroxisome induction studies ontri(2-ethylhexyl)rimellitate and 2-ethylhexanol. Toxic.Ind. Hlth, Vol.3, No.2, 1987

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Nuodex Inc. Acute dermal toxicity test of Tenneco Chemicals, Inc. compound Nuoplaz 6959 in rabbits. Doc ID878214467. (1982a)

Nuodex Inc. Acute inhalation toxicity test in sprague-dawley rats using compound Nuoplaz 6959 Doc ID878214466 (1982b)

Nuodex Inc. Acute oral toxicity—Rats Doc ID878214469 (1981)

The British Industrial biological Research Association; A 28-day Toxicity Study with TOTM in the Rat with Cover Letter Dated 111885. (1985)

Yoshimasa Kurata. Subchronic Toxicity of Di(2-ethylhexyl)phthalate in Common Marmosets: Lack of Hepatic Peroxisome Proliferation, Testicular Atrophy, or Pancreatic Acinar Cell Hyperplasia. Toxicological Sciences 42, 49-56 (1998) PROPOSED ROBUST SUMMARY for Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate CAS No. 3319-31-1

Sponsor Country: Japan

D' sug 24, 2001

## PHYSICAL/CHEMICAL ELEMENTS

#### MELTING POINT

### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

· Remarks:

Source: Tokyo Kasei Kogyo Co., Ltd. Lot No. AX01

Purity: 98.5%

**METHOD** 

Method/guideline:

OECD TG 102

• GLP:

Yes

· Year:

1998

· Remarks:

Not stated.

RESULTS

Melting point value:

<-50 °C (223 K)

Decomposition:

Not stated,

Sublimation:

Not stated.

Remarks:

Not stated.

## CONCLUSIONS

Melting point is <-50°C (223 K).

### DATA QUALITY

Reliabilities:

Key study

· Remarks:

Well conducted study, carried out by Chemicals Evaluation and

Research Institute (Kurume, Japan).

#### REFERENCES

Ministry of International Trade and Industry (1998)

- Last changed:
- Order number for sorting
- Remarks:

## BOILING POINT (a)

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Unavailable.

#### **METHOD**

Method:

Not specified.

GLP:

Not stated.

Year:

Not stated.

Remarks:

Not stated.

### RESULTS

Boiling point value:

283°C

· Pressure:

4

Pressure unit:

hPa

Decomposition:

Not stated,

Remarks:

Not stated.

#### CONCLUSIONS

Boiling point is 283°C at 4 hPa.

### DATA QUALITY

Reliabilities:

Key study

Remarks:

Not stated.

### REFERENCES

Midwest Research Institute; Thomas W. Lapp, Charles EMumma Joseph Chaszar: A Survey of Plasticizers: Epoxies, Linear Polyesters and Trimellitates Chemical Technology and Economics in Environmental Perspective, Task IV, Environmental Protection Agency (Nov. 1981)

- Last changed:
- Order number for sorting
- Remarks:

## BOILING POINT (b)

### TEST SUBSTANCE

· Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

· Remarks:

Source: Unavailable.

### METHOD

• Method:

Not specified.

GLP:

Not stated.

· Year:

Not stated.

Remarks:

Not stated.

### **RESULTS**

Boiling point value:

414°C (687K)

Pressure:

1,013

· Pressure unit:

hPa

Decomposition:

Not stated.

Remarks:

Not stated.

### CONCLUSIONS

Boiling point is 414°C at 1,013hPa.

### DATA QUALITY

• Reliabilities:

Key study

· Remarks:

The Sigma-Aldrich Library of Regulatory and Safety Data.

## REFERENCES

Ministry of International Trade and Industry (1998)

- Last changed:
- Order number for sorting
- Remarks:

### DENSITY

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Unavailable.

#### METHOD

Method:

Not specified.

• GLP:

Not stated.

· Year:

Not stated.

· Remarks:

Not stated.

#### RESULTS

Density:

 $0.987 - 0.990 \text{ g/cm}^3$ 

Temperature

20°C

Remarks:

Not stated.

#### CONCLUSIONS

Density is 0.987-0.990 g/cm<sup>2</sup> at 20°C.

### DATA QUALITY

Reliabilities:

Key study

Remarks:

Not stated.

## REFERENCES

Midwest Research Institute; Thomas W. Lapp, Charles EMumma Joseph Chaszar: A Survey of Plasticizers: Epoxies, Linear Polyesters and Trimellitates Chemical Technology and Economics in Environmental Perspective, Task IV, Environmental Protection Agency (Nov. 1981)

- Last changed:
- Order number for sorting
- Remarks:

## VAPOR PRESSURE (a)

### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Tokyo Kasei Kogyo Co., Ltd. Lot No. AX01

Purity: 98.5%

**METHOD** 

Method/guideline:

**OECD TG 104** 

• GLP:

Yes

· Year:

1998

Remarks:

Not stated.

RESULTS

Vapour Pressure value:

 $< 2.8 \times 10^{-4} \text{ Pa}$ 

• Temperature:

100°C

Decomposition:

Not stated.

Remarks:

Not stated.

**CONCLUSIONS** 

Vapour pressure is < 2.8 x 10<sup>-4</sup> Pa at 100°C.

DATA QUALITY

· Reliabilities:

Key study

Remarks:

Well conducted study, carried out by Chemicals Evaluation and

Research Institute (Kurume, Japan).

REFERENCES

Ministry of International Trade and Industry (1998)

- Last changed:
- · Order number for sorting
- Remarks:

### VAPOR PRESSURE (b)

#### TEST SUBSTANCE

· Identity:

Tris(2-ethylhexyl)benzene 1,2,4 tricarboxylate

· Remarks:

Source: Unavailable,

### **METHOD**

Method/guideline:

Not stated

• GLP:

Not stated

Year:

Not stated

· Remarks:

Not stated.

#### RESULTS

Vapour Pressure value:

0.27 - 6.7 hPa

Temperature:

250 -260 °C

· Decomposition:

Not stated.

· Remarks:

Not stated.

### CONCLUSIONS

Vapour pressure is 0.27- 6.7 hPa at 250 - 260 °C.

## DATA QUALITY

Reliabilities:

Key study

\* Remarks:

Not stated.

### REFERENCES

Midwest Research Institute; Thomas W. Lapp, Charles EMumma Joseph Chaszar: A Survey of Plasticizers: Epoxies, Linear Polyesters and Trimellitates Chemical Technology and Economics in Environmental Perspective, Task IV, Environmental Protection Agency (Nov. 1981)

- Last changed:
- Order number for sorting
- Remarks:

### PARTITION COEFFICIENT

### TEST SUBSTANCE

· Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Tokyo Kasei Kogyo Co., Ltd. Lot No. AX01

Purity: 98.5%

### **METHOD**

Method/guideline:

Yes

OECD TG 107 (Shake Flask Method, 1995)

• GLP:

GLP: Year:

1998

Remarks:

Not stated.

#### RESULTS

· Log P...:

5.94

Temperature:

25°C ±1°C

Remarks:

Test condition: Test was conducted in duplicate under the following

three conditions. Test chemical was analyzed by HPLC.

Condition-1	Condition-2	Condition-3
$10  \mathrm{mL}$	20 mL	40 mL
240 m.L.	230 mL	210 mL
ed with water (	52.2 mg)	
	10 mL 240 mL	10 mL 20 mL

10 ml. 10 ml.

Test reults	Log		
•	*	Ъ	Mean
Condition-1	5.99	5.99	
Condition-2	5.95	5.87	5.94
Condition-3	5.92	5.93	

**CONCLUSIONS** 

log P is 5.94.

### DATA QUALITY

Reliabilities:

Key study

\* Remarks:

Well conducted study, carried out by Chemicals Evaluation and

Research Institute (Kurume, Japan).

### REFERENCES

Ministry of International Trade and Industry (1998)

#### OTHER

Last changed:

Order number for sorting

Remarks:	 <del></del>	

#### WATER SOLUBILITY

### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

· Remarks:

Source: Tokyo Kasei Kogyo Co., Ltd. Lot No. AX01

Purity: 98.5%

METHOD

• Method:

OECD TG 195 (flask method).

• GLP:

Yes

· Year:

1998.

Remarks:

Not stated,

RESULTS

• Value:

0.13 mg/L at 25 °C±1°C

Description of solubility:

Of very low solubility

• pH value;

No dissociation group.

pKa value;

There is no pertinent functional group.

Remarks:

Not stated.

### CONCLUSIONS

This chemical is very low solubility in water.

### DATA QUALITY

· Reliabilities:

Key study

· Remarks:

Well conducted study, carried out by Chemicals Evaluation and

Research Institute (Kurume, Japan).

#### REFERENCES

Ministry of International Trade and Industry (1998)

- Last changed:
- Order number for sorting
- Remarks:

## ENVIRONMENTAL FATE AND PATHWAYS ELEMENTS

#### STABILITY IN WATER

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Tokyo Kasei Kogyo Co., Ltd. Lot No. AX01

Purity: 98.5%

#### METHOD

Method/guideline:

OECD TG 111

Type:

Hydrolysis as a function of pH

• GLP:

Yes

Year:

1998

Remarks:

No hydrolysis of test chemical was observed at pH 4 at 50°G-1°C for 5 days. Hydrolysis rates at pH 7 were determined at 60, 70 and 80 °C, and at pH 9 at 50, 60, and 70°C. They were extrapolated to 25 °C using Arrhenius relationship. Half life at 25 °C was calculated from the rate

constant.

#### RESULTS

Nominal:

ca. 0.2 mg/L

Measured value:

Not stated.

Degradation:

No hydrolysis occurred in 5 days, at 50 °C pH 4. At pH 7 and pH 9,

test chemicals were hydrolysed at all temperatures studied.

Half-life (t<sub>(1/2)</sub>):

Rate Constant (hr<sup>-1</sup>) 1.65 x 10<sup>-9</sup> Half-life(day)

pH7 pH9

2.44 x 10<sup>-3</sup>

17.5 11.9

Breakdown products:

Not stated.

Remarks:

Not stated.

### CONCLUSIONS

This chemical is stable in aqueous water at pH 4 under the condition studied, but it is hydrolysed at pH 7 and pH 9 at 25 °C with half-life of 17.5 and 11.9 days.

## DATA QUALITY

Reliabilities:

Key study

Remarks:

Well conducted study, carried out by Chemicals Evaluation and

Research Institute (Kurume, Japan).

#### REFERENCES

Ministry of International Trade and Industry (1998)

# DRAFT ENV/JM/EXCH(99)13

- Last changed: Order number for sorting
- Remarks:

# TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS (FUGACITY)

## TEST SUBSTANCE

Identity:

Tris(2-cthylhexyl)benzene-1,2,4-tricarboxylate

\* Remarks:

Source: Not applicable.

#### METHOD

Test:

Calculation

· Method:

Fugacity level III

· Year:

2001

• Remarks

The parameters used are shown in Appendix.

#### RESULTS

Media :

Estimated Distribution under three emissionscenarios:

Compartment	Release100 % to air	Release 100 % to water	Release 100 % to soil
Air	19.6 %	0.0%	0.0 %
Water	4.7 %	32.7 %	0.0%
Soil	66.2 %	0.1 %	100.0%
Sediment	9.5 %	67.2 %	0.0 %

Remarks

### CONCLUSIONS

If this chemical is released into water the majority of this chemical is expected to stay in sediment but if it is released into air or soil, this chemical is expected to stay in soil

### DATA QUALITY

\* Reliabilities:

Key study.

Remarks:

Not stated.

#### REFERENCES

Dainippon lak and Chemicals, Incorporated (2001), unpublished report.

- Last changed;
- Order number for sorting
- Remarks:

#### BIODEGRADATION

## TEST SUBSTANCE

· Identity:

Tris(2-ethylliexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Unavailable

METHOD

· Method:

OECD TG 302C "Inherent Biodegradability: Modified MITHest(II)"

· Test Type:

Acrobic

· GLP:

No

· Years

1977

Contact time:

28 days

Inoculum:

The supernatant (500ml) of activated sewage sludge obtained from a sampling sites and Sliters of supernatant removed from a previously established culture are transferred to a culture vessel. The pH of the culture mixture was adjusted to 7.0±1.0 and constantly acrated. Thirty minutes after stopping acration, discard about 1/3 of the whole volume of the supernatant, and add an equal volume of 0.1% synthetic sewage and the persistence of carried Parent this precedure care a day.

and the acration re-started. Repeat this procedure once a day.

· Remarks:

During the aeration, appearance of supernatant and the formation of activated sewage was observed. The sludge was found to form a clear supernatant on settling and formed cloudyfloes when on aeration. Operating temperature, pH and a dissolved oxygen concentration were recorded. The protozoa of sludge were observed under an optical microscope.

\*Incubation apparatus: Respirometry(Closed bottle) Ohkura Electric Co.

\*CO2 absorbent: Soda lime No.1 (Wako pure chemicals Inc.)

\*Stirrer : Magnetic stirrer \*Temperature : 25±1°C

Concentration of test chemical: 30mg/L, 100mg/L.

\*Reference substance: Aniline

#### RESULTS

Degradation:

Results:

4.2% after 28days

Kinetic:

The percentage degradation in term of oxygen consumption was

calculated as follows:

% degradation = (BOD-B)/IOD x 100

BOD: Biological Oxygen Demand of the test material

B : Oxygen consumption in basal culture medium to which

inoculum is added (control)

TOD: Theoretical oxygen demand to completely oxidize the test

Material

Breakdown products:

Not stated.

Remarks:

At the end of incubation, measure the residual dissolved organicarbon and test material concentration. The reference substance, aniline attained more than 40% and 60% degradation after 7 and 14days confirming the suitability of the inoculum and culture conditions.

### CONCLUSIONS

This chemical islow biodegradable.

### DATA QUALITY

· Reliabilities:

Key study

Remarks:

Well conducted study, carried out by Chemical Inspection and Testing

Institute.

### REFERENCES

Biodegradation and Bioaccumulation Data of Existing Chemicals Based on the CSCL Japan (1992)

Ministry of International Trade and Industry

- Last changed:
- · Order number for sorting
- · Remarks:

### BIOACCUMULATION

### TEST SUBSTANCE

ldentity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Unavailable

METHOD

Method:

OECD TG 305C

Species:

Cyprinus Carpio (Obtained from Nakajima hatchery in Kumamoto,

Japan)

GLP:

No

**Уеаг:** 

1978

Exposure Period:

42 days

Remarks:

Test fish:

Acclimated for ca. 8 weeksbefore testing at 25±2°C. Fish with ca.10cm

In length and ca.30g in weight were selected at random. Lipid content

was 2-6%.

Test condition

Concentrations: 0.2 and 2 mg/L, solubilizer controlled

Type of test: flow-through (200-800mL/min), 100L glass tank.

Dissolved oxygen concentration: 6-8mg/L.

Temperature: 25 ±2°C

Water chemistry was tested in the control and two concentrations every

2 times in a week.

Test was conducted in duplicate every 2 weeks for two concentrations.

(The control was done before and after testing.)

RESULTS

Resuits:

BCF=1-2.7 (concentration: 0.2mg/L)

BCF=0.1-0.23(concentration: 2mg/L)

Kinetic:

BCF=C1/C2

C1: Concentration of this chemical in Fish...

C2: Concentration of this chemical in water

Breakdown products:

Not stated.

#### CONCLUSIONS

This chemical is low bioaccumulation.

DATA QUALITY

Reliabilities:

Key study

Remarks:

Well conducted study, carried out by Chemical Inspection and

Testing Institute

## REFERENCES

Biodegradation and Bioaccumulation Data of Existing Chemicals Based on the CSCI, Japan(1992)

Ministry of International Trade and Industry

- · Last changed:
- · Order number for sorting
- · Remarks:

## ECOTOXICITY ELEMENTS

#### ACUTE TOXICITY TO FISH

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Tokyo Kasei Kogyo Co., Ltd. Lot No. AX01

Purity: >95.0%

#### METHOD

Method:

OECD TG 203

Type:

Semi-static

GLP:

Yes

Year:

1998.

Species/Strain/Supplier: Oryzias latipes (Medaka): Obtained from commercial domestic

hatcheries.

Analytical monitoring

Yes. Test solutions were measured by IIPLC before and after 24 hours.

exposure period. Test solutions were replaced every 24 hours to new ones.

Exposure period (h):

Statistical methods:

Not applicable because of nomortality.

Remarks:

Test fish:

Acclimated for more than 12 days before testing; any groups showing no mortality for 7 days before test started. Fish with 22.1 mm (18.3-23.8) mm) in length were selected at random. Average body weight of fishers 0.1462g (n=10).

Test conditions

Details of test: Semi-static (water changedevery 24 hours)

Dilution water source: Tap water after dechlorinated by passing through

activated carbon.

Dilution water chemistry: Hardness: 25 mg/L as CaCO3; pH: 6.7 Slock and test solution and how they are prepared: Pipene or pour the appropriate amount of the solution (0.3 wt% of test chemicalwith solubilizer hydrogenated caster oil HCO-40 3000mg/L into the test waters.

Concentrations dosing rate, flow-through rate, in what medium: Concentrations of 0, 100 mg/L and dispersant control were tested. Vehicle/solvent and concentrations: Hydrogenated caster oil HCO-40,

100mg/L

Stability of the test chemical solutions: Stable, measured concentration was 101-103%.

Exposure vessel type: 10 fish per group in 3L glass beaker without acration under room light

Number of replicates, fish per replicate: One replicatewas donce Water chemistry in test (O 2, pH) in the control and all concentration where effects were observed: Dissolved oxygen readings and pH values were taken daily during 96 h exposure period.

Dissolved oxygen concentration: 5.0-9.2 mg/L.

pH values: 6.7-6.8.

Test temperature range: Water temperature at 23.5-24.1°C.

Method of calculating mean measured concentrations: Geometric mean.

#### RESULTS

Nominal concentrations: 0, 100 (mg/L)

Mensured concentrations: <1, 103 (0hr), <1, 102 (24hr)</li>

• Unit :

 $mg/L_{\star}$ 

Element value

LC<sub>sp</sub> at 96 hours >100.0 mg/L based on nominal concentrations.

Statistical results as appropriate: Not applied.

Remarks field for Results:

Biological observations

Not described.

Table showing cumulative mortality:

Percent mortality of Oryzias latipes exposed to the test chemical Nominal concentration (mg/L) Cumulative number of dead fish (% mortality) 24 hour 48 hour 72 hour 96 hour O(0)Control 0(0)Q(U)1(10) $\Omega(0)$ 0(0)(0)0O(D)Dispersant Control O(0)1(10) 1(10) 1(10)100

Lowest test substance concentration causing 100% mortality:

Not obtained under the test conditions studied.

Mortality of controls:

1 fish was dead at 96h.

Abnormal responses:

At 24 hr, one fish showed abnormal breathing behaviour at 100mg/L.

Reference substances:

Copper(II) sulfate pentahydrate. LC<sub>50</sub> at 96h was 0.43 mg/L.

Any observations, such as precipitation that might cause a difference between measuredne nominal values:

It became clouded in 100mg/L concentration, but not precipitation.

### CONCLUSIONS

LC50 (96h) > 100mg/L for fish.

### DATA QUALITY

Reliabilities: Klimisch Code: 1=reliable without restrictions.

Remarks field for Data Reliability:

Experimental design and analytical procedure were well documented.

Carried out by Toray Research Center (Japan).

### REFERENCES

Environment Agency of Japan (1998).

- · Last changed:
- · Order number forsorting:

• Remarks field for GeneralRemarks:

#### PROLONGED TOXICITY TO FISH

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricatboxylate

Remarks:

Source: Tokyo Kasei Kogyo Co., Ltd. Lot No. AX01

Purity: >95.0%

#### METHOD

Method:

OECD TG 204

Type:

Flow-through.

GLP:

Yes.

Year:

1998.

Species/Strain/Supplier:

Organias latipes (Medaka): Obtained from commercial domestic hatcheries.

Analytical monitoring:

Yes. Test solutions were measured by HPLC before and after 7, 14days

exposure period.

Exposure period:

14 day.

Statistical methods

Binomial method (TOXDAT MULTI-METHOD PROGRAM, USEPA) Dunnet method was used for LG, and for fish body weight difference,

respectively.

Remarks field for Test Conditions:

Text fish:

Acclimated for more than 12 days before testing; any groups showing 9% mortality for 7 days before test started. Fish with 20.0 mm (18.5-21.6 mm) in length were selected at random. Average body weight of fish was 0.484g (0.1182-0.2014g)(n=10). Fish were starved for 24 hours before the test

Test conditions

Details of test: Flow-through,

Dilution water source: Tap water after dechlorinated by passing through

activated carbon.

Dilution water chemistry: Hardness: 15.3mg/L as CaCO<sub>5</sub>; pH: 7.0 Stock and test solution and how they are prepared: The working solution (4.8wt% of test chemical with solibitizer HCO-40 controlled was prepared with the dilution water. The test solution was supplied continuously by

mixing the working solution and the dilution water with the help of a mechanically operated quantitative water-pump.

Concentrations dosing rate, flow-through rate, in what medium: Nominal concentrations of 0, 18.8, 37.5 and 75.0 mg/L and Dispersant control were tested.

Vehicle/solvent and concentrations Hydrogenated caster oil HCO-40, Max. 75.0 mg/L

Stability of the test chemical solutions: It became clouded in high concentration, but not precipitation.

Exposure vessel type: 10 fish per group in 3L glass beaker without aeration

under room light

Number of replicates, fish per replicate: One replicate was done.

Water chemistry in test (O2 pH) in the control and one concentration where

effects were observed: Dissolved oxygen readings and pH values were

taken every 3 days during the exposure period. Dissolved oxygen concentration: 6.6-7.7 mg/L.

pH values: 6.9~7.2.

Test temperature range:

Water temperature at 23.5-24.1 °C ( $24\pm2$  °C).

Method of calculating mean measured Geometric mean.

### RESULTS

- Nominal concentrations: 0, 18.8, 37.5, 75.0 (mg/L) and dispersant control
- Measured concentrations :

Measured concentration of the test chemical during a 14-day exposure of orange killifish (Oryzias latipes) under flow-through test conditions

Nominal concentration (mg/L)	Measured concentration (mg/L) (percent of nominal)							
	O day	7 day	14 day	Mean				
Control	<1.0	< 1.Ô	< 1.0	•				
Dispersant Control	< 1.0	< 1.0	< 1.0	<b>-</b> -				
18.8	17.7(94.1)	15.8(84.0)	15.5(82,4)	16.3(86.9)				
<b>3</b> 7. <i>5</i>	35.7(95.2)	33.2(88.5)						
75.0	70.6(94.1)	68.8(91.7)	71.2(94.9)					
ma/l	•	, ,	• /	4,				

Unit:

mg/L

Element value:

 $LC_{50}$  (7 days) > 75.0mg/L (nominal concentration)

 $LC_{so}$  (14 days) > 75.0 mg/L (nominal concentration)

NOEC (14 days) > 75.0 mg/L (nominal concentration)

Statistical results, as appropriate:

The mean body weight of fish exposed toall concentration of the test chemical was not significantly different from controls during the test periodalfa=0.05, Dunnet).

Remarks field for Results:

Biological observations: Not described.

Cumulative mortality:

Percent mortality of *Orygins latipes* exposed to the test chemical under flow-through test Conditions

Nominal conc. (mg/L)		Cum	ulativ	e num	ber of	dead	lish	(% m	ortalit	y)				(da	ys)	
-	0	1	2	3	4	5	б	7	3	9	10	11 ·	12	13	14	
Control	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	1(10)	1(10)	1(10)	
Disp. Cont.	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	c(o)	0(0)	
18.8	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	
37.5	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	
75.0	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	

Fish weight:

Numinal cone. (mg/L)	•	Fish v	veight (	<b>g</b> )							
	No.1	No.2	No. 3	No.4	No.5	Nv.6	No.7	No.8	No.9	No.10	Ave.
Control	0.1879	0.2526	0.1273	0.2239	0.1139	0.1434	0.1708	0.1789	0.1558	-a	0.1727
Disp. Cont.	0.2205	0.1827	0.1192	0.1884	0.1438	0.1823	0.1563	0.2120	0.1635	0.1580	0.1727
18.8	0.1731	0.1513	0.1593	0.1472	0.2150	0.1548	0.1547	0.1306	0.2104	0.1020	0.1598
37.5	0.1264	0.1495	0.1872	0.1237	0.2055	0.1396	0.1805	0.2101	0.1577	0.1303	0.1611
75.0	0.1746	0.1848	0.1804	0.1625	0.1494	0.1633	0.2103	0.1454	0.1600	0.1818	0.1713
	- 1	: No me	as uremen	t was mad	de becaus	e the Orai	ige Killif	ish was de	ead.		

Lowest test substance concentration causing 100% mortality>75.0 mg/mL (nominal).

Mortality of controls:10 % mortality observed during the test period (12 through 14 days).

Food intake:

Fish was fed with TetraMin<sup>®</sup> fish food (2% of fish body weight).

Abnormal responses: No abnormal response showed through 14 days.

Reference substances (if used)- results: Copper (II) sulfate pentahydrate. LC<sub>50</sub> at 96h was 0.30 mg/L.

Any observations, such as precipitation that might cause a difference between measured and nominal values: It became clouded high concentration, butnot precipitation.

#### CONCLUSIONS

 $LC_{50}$  (7 days) > 75.0 mg/L (nominal concentration)  $LC_{50}$  (14 days) > 75.0 mg/L (nominal concentration) NOEC (14 days) > 75.0 mg/L (nominal concentration)

## **DATA QUALITY**

Reliabilities:

Klimisch Code: 1=reliable without restrictions.

Remarks field for Data Reliability:

Experimental design and analytical procedure were well documented.

Carried out by Toray Research Center (Japan).

### REFERENCES

Environment Agency of Japan (1998).

- Last changed:
- Order number for sorting:
- Remarks and for GeneralRemarks:

## ACUTE TOXICITY TO AQUATIC INVERTEBRATES (e.g., Daphnia)

#### TEST SUBSTANCE

· Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Tokyo Kasei Kogyo Co., Ltd. Lot No. AX01

Purity: >95.0%

METHOD

Method:

OECD TG 202

Type:

Static

• GLP:

Yes

· Year:

1998

Species/Strain/Supplier:

Daphnia magna

Analytical monitoring

Yes. Test solutions were measured by HPLC before and after 48 hours

exposure period.

Exposure period (h):

48

Statistical methods:

Not applicable.

Remarks field for Test Conditions:

Test organisms:

Source, supplier, any pre-treatment, breeding method: Supplied by NIES

(Japan).

Age at study initiation: Juveniles within 24h old.

Control group: Yes.

Test conditions

Stock solutions preparation and stability: No solvent used. Test chemical

was diluted to 1800mg/L(with solubilizer HCO-40 1000mg/L controlled)

with diluting water (Elendt M4) before use.

Test temperature range:

19.9-20.2 °C (average temperature 20°C).

Exposure vessel type: 100mL test solution in a 100 mL glass beaker; 4

beakers per treatment

Dilution water source: Elendt M4(OECD guideline No.211 Annex 2)

Dilution water chemistry: Hardness: 228mg/L as CaCO<sub>3</sub> Lighting: room light 16h:8h light-darkness cycle Water chemistry in test: DO= 8.0-8.6mg/L; pH=7.3-7.8.

Feeding: none

Test design:

Number of replicates=20

Concentrations: 0, 17.1, 30.9, 55.6, 100 and 180 mg/L, because 48h-EiC<sub>50</sub> for parent Daphnia (Acute immobilization test) was>1000mg/L. Dispersant

control was also tested.

Method of calculating mean measured concentrations. Geometric mean.

Exposure period:

48 h

Analytical monitoring:

By HPLC analysis. 95.1-99.6% of the nominal concentration at

preparation; 90.1-97.7% after 48hr.

## ACUTE TOXICITY TO AQUATIC INVERTEBRATES (e.g., Daphnia)

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Tokyo Kasei Kogyo Co., Ltd. Lot No. AX01

Purity: >95.0%

#### METHOD

Method:

**OECD TG 202** 

• Type:

Static

• GLP:

Yes

Year:

1998

Species/Strain/Supplier:

Daphnia magna

Analytical monitoring

Yes. Test solutions were measured by HPLC before and after 48 hours

exposure period.

Exposure period (h):

48

Statistical methods:

Not applicable.

#### Remarks field for Test Conditions:

Test organisms:

Source, supplier, any pre-treatment, breeding method: Supplied by NIES

(Japan).

Age at study initiation: Juveniles within 24h old.

Control group: Yes.

Test conditions

Stock solutions preparation and stability: No solvent used. Test chemical

was diluted to 1800mg/L (with solubilizer HCO-40 1000mg/L controlled)

with diluting water (Elendt M4) before use.

Test temperature range:

19.9-20.2 °C (average temperature 20°C).

Exposure vessel type: 100mL test solution in a 100 mL glass beaker; 4

beakers per treatment

Dilution water source: Elendt M4(OECD guideline No.211 Annex 2)

Dilution water chemistry: Hardness: 228mg/L as CaCO<sub>3</sub> Lighting: room light 16h:8h light-darkness cycle Water chemistry in test: DO= 8.0-8.6mg/L; pH=7.3-7.8.

Feeding: none

Test design:

Number of replicates=20

Concentrations: 0, 17.1, 30.9, 55.6, 100 and 180 mg/L, because 48h-EiC<sub>50</sub> for parent Daphnia (Acute immobilization test) was>1000mg/L. Dispersant

control was also tested.

Method of calculating mean measured concentrations. Geometric mean.

Exposure period:

48 h

Analytical monitoring:

By HPLC analysis. 95.1-99.6% of the nominal concentration at

preparation; 90.1-97.7% after 48hr.

### RESULTS

Nominal concentrations: 17.1, 30.9, 55.6, 100.0, 180.0 (mg/L) (Solubilizer controlled)

Measured concentrations:

Measure Concentrations of test chemicals during a 48hr.

Nominal Concentration	Measur	ed concent	Percent of nominal		
( mg/L)	Ohr	48hr	Mean	Ohr	48hr
Control	< 1.0	< 1.0	-	-	-
Disp.Cont.	< 1.0	< 1.0	•	•	_
17.1	16.3	15.4	15.8	95.3	90.1
<b>30.</b> 9	29.4	28.5	28.9	95.1	92.2
55.6	53.0	52.1	52.5	95.3	93.7
100.0	98.4	96.3	97.3	98.4	96.3
180.0	179.2	175.8	177.5	99.6	97.7

Unit:

mg/L.

Nominal concentration (mg/L)

Element value

EC<sub>50</sub> at 24 hours >180.0 mg/L

EC<sub>10</sub> at 48 hours > 180.0 mg/L.

NOEC > 180.0 mg/L LOEC > 180.0 mg/L

- Statistical results as appropriate:Not applied.
- Remarks field for Results:

Biological observations

Not described.

Table showing mortality or immobility

Mortality or immobility of Daphnia magna to the test chemical

	(Percent Mort	(Percent Mortality or Immobility)					
	24 hour	48 hour					
Control	0(0)	0(0)					
Dissersant Control	0(0)	1(5)					
17.1	0(0)	1(5)					
30.9	0(0)	0(0)					
<b>55.6</b>	0(0)	0(0)					
100.0	0(0)	0(0)					
180.0	0(0)	0(0)					

Lowest test substance concentration causing 100% mortality:

Not obtained under the test conditions studied.

Mortality of controls:

No mortality observed during test period.

Abnormal responses:

No abnormal responses observed during test period

Reference substances:

Potassium dichromate EC<sub>sp</sub> at 48h was 0.87 mg/L.

Any observations, such as precipitation that might cause a difference between measuredand nominal

values:

It became clouded in high concentration, butnot precipitation.

Cumulative number of dead or immobilizes Daphnia

#### CONCLUSIONS

 $EC_{50}$  (48h) > 180mg/L and NOEC (48h) > 180mg/L for Daphnia magna.

## DATA QUALITY

Reliabilities:

Klimisch Code: 1=reliable without restrictions.

Remarks field for Data Reliability:

Experimental design and analytical procedure were well documented.

Carried out by Toray Research Center (Japan).

### REFERENCES

Environment Agency of Japan (1998).

- Last changed:
- · Order number forsorting:
- Remarks field for GeneralRemarks:

## TOXICITY TO AQUATIC PLANTS (E.G., ALGAE)

#### TEST SUBSTANCE

· Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Tokyo Kasei Kogyo Co., Ltd. Lot No. AX01

Purity: >95.0%

**METHOD** 

Method/guideline followed: OECD TG 201

Test type :

Static.

GLP:

Yes

· Year:

1998

Species/strain # and source: Selenastrum capricornutum ATCC22662 (purchased from ATCC)

Element basis:

Area under the growth curve.

Exposure period:

72 h.

Analytical monitoring:

Yes, measured by HPLC at start and end of the test (72hr).

Statistical methods:

Bartlett test for homogeneity in variances and One-wayAnova (EcoTox-Statistics Ver. 1.0 beta-edition R1.4) were used for EC<sub>50</sub>, LC<sub>50</sub> and NOEC.

determination (p=0.05).

#### Remarks field for Test Conditions:

Test organisms

Laboratory culture: OECD medium

Method of cultivation: Shaking at 100rpm

Controls: OECD medium. EC<sub>50</sub> of potassium dichromate was 0.41 mg/L.

Test Conditions

Test temperature range: 23±2 °C

Growth/test medium: OECD medium.

Shaking: 100 rpm

Dilution water source: OECD medium.

Exposure vessel type: 100 mL OECD medium in a 300 mL Erlenmeyer

flask with a silicon cap which allows ventilation.

Water chemistry in test (pH) in at least one replicate of each concentration (at start and end of the test): pH=7.3-7.4 at start and 8.3-8.8 at end of the

test (72 h).

Stock solutions preparation: No stock solution was prepared. Test chemical was diluted to 100 mg/L (solubilizer, HCO-40 100 mg/L) with

OECD medium and sterilised with filter before use.

Light levels and quality during exposure: 4,756-4,822 lux, continuous

illumination.

Test design

Number of replicates: Triplicate

Concentrations: 0, 100 mg/L and dispersant control were tested.

Initial cell number in cells/mL: 1x104

Method of calculating mean measured concentrations Geometric mean.

#### RESULTS

Nominal concentrations:

0, 100 (mg/L) and dispersant control.

Measured concentrations:

At start of the test (0 hr), <1.0, 80.6, <1.0 (mg/L)

At end of the test (72 hr), <1.0, 68.7, <1.0 (mg/L)

• Unit :

mg/L

Results:

(calculated based on nominal concentrations)

(1) Growth inhibition (comparison of area under growth curve)

 $EC_{5c}$  (0-72 h) > 100 mg/L NOEC (0-72 h)> 100 mg/L

(2) Growth inhibition (comparison of growth rates)

 $EC_{50}$  (24-48) > 100 mg/L  $EC_{50}$  (24-72) > 100 mg/L NOEC (24-72) > 100 mg/L

Was control response satisfactory:

Yes: Mean cell density increased to 270x10<sup>6</sup> cells/mL (270-fold increase) after 72 hr for control. Mean cell density increased to 275x10<sup>6</sup> cells/mL (275-fold increase) after 72 hr for Dispersant control.

Statistical results as appropriate:

Significant difference in the growth curve was not observed between values at 100 mg/L and in each control.

#### Remarks field for Results:

Biological observations

Cell density at each flask at each measuring point:

Nominal Concentration (mg/L)	0 br	Cell Density	(x10 <sup>4</sup> cells/mL) 48 hr	72 hr
Control	1.0±0.00	6.5±0.50	50.5± 3.48	270.5±23.50
Dapersant Control	1.0±0.00	9.3±1.66	57.5± 9.39	275.3±17.22
1 00	1.0±0.00	16.1±7.82	65.1±12.82	283.3± 7.98
	(Each value	e represents the	mean of three :	ample counts.)

Growth curves: Logarithmic growth until end of the test (72 h).

Percent biomass/growth rate inhibition per concentration: Not described.

Observations: Test group (100 mg/L) showed normal and similar growth to that of control (283 fold increase after 72 hr).

#### CONCLUSIONS

(1) Growth inhibition comparison of area under growth curve)  $EC_{50}$  (0.72 h) > 100 mg/L NOEC (0.72 h) > 100 mg/L (2) Growth inhibition (comparison of growth rates)  $EC_{50}$  (24-48) > 100 mg/L  $EC_{50}$  (24-72) > 100 mg/L NOEC (24-72) > 100 mg/L

# DATA QUALITY

· Reliabilities:

Klimisch Code: 1=reliable without restrictions.

Remarks field for Data Reliability:

Experimental design and analytical procedure were well documented.

Carried out by Toray Research Center (Japan).

### REFERENCES

Environment Agency of Japan (1998).

- Last changed:
- Order number forsorting:
- · Remarks field for GeneralRemarks:

Nominal concentration

(mg/L)

## CHRONIC TOXICITY TO AQUATIC INVERTEBRATES (e.g., DAPHNIA) (1)

#### TEST SUBSTANCE Identity: Tris(2-cthylhexyl)benzene-1,2,4-tricarboxylate Remarks: Source: Nuoplaz 6965 **METHOD** ASTM and USEPA Method: Flow-through condition Test type: GLP: Yes 1984 Year: Yes. Measured by GLC, on 0,4,7,14,21day) Analytical procedures: Species/Strain: Daphnia magna Test details: Dynamic flow-through Statistical methods: ANOVA, 2WANOVA, arcsin transformation and Fisher's protected Least Significant Difference (LSD) Remarks field for Test Conditions: Test organisms: Source; in house culture Age at study initiation: Juveniles within 24h old. Control group: Yes (control and solvent control) Test conditions Dilution Solvent for Concentrated stock standards: Acetone (1.049mg/mL) A proportional diluter system was used for the intermittent introduction of test material and dilution water into the test chambers. Test temperature range: 18-22 °C (average temperature 20°C). Well water was delivered to the chambers as a minimum rate of 2.0mL/min. Exposure vessel type: 900mL test solution in a 1000 mL glass beaker, 4 beakers per treatment Dilution water chemistry: Hardness and other characteristics are reported. Dilution water pH in test: pH=8.3-8.4. Lighting: 37-74 footcandles, 16h:8h light-darkness cycle Feeding: Algae (Selenastrum capricornutum) three times a day Supplemented with a trout chow suspension at least twice a week Mean cumulative numbers of juveniles produced per adult (reproduction) Element (unit) basis: Growth (length) of parental Daphnia Long-term survival Number of replicates=4; individuals per replicate=10; Test design: Method of calculating mean measured concentrations Geometric mean. 21 d Exposure period: By GLC analysis, 33-101% of the nominal concentration at Preparation Analytical monitoring: RESULTS Nominal concentrations: 0, 0.0074, 0.012, 0.027, 0.048, 0.100 mg/L Measured concentrations:

0

Measured concentration of test chemical during 21-day exposure

Measured concentration (day, mg/L)

14

21

mean

Control			מא	ND		ND	ND		ND	NI
Solvent Cont.			ND	NE	)	ND	ND		ND	ND
0.0074		0	.00328	0.00	0366	0.00558	0.00	246	0.00482	0.004
0.012		0	.00748	0.00	0626	0.00843	0.00	478	0.00747	0.006
0.027		0	.0172	0.0	150	0.0204	0.01	10	0.0157	0.015
0.048		0	.0305	0.03	252	0.0371	0.01	7 <b>6</b>	0.0348	0.029
0.100		0.	.0824	0.07	766	0.0870	0.06	30	0.1011	80.0
Cumulative Nun	ber of	Dead	Parent	aDaph	nia.					
Nominal conc.	Days			_						
(mg/L)	0	3	5	7	10	12	14	17	19	21
Control	0	0	0	0	0	Ó	0	1	1	2
Solvent Cont.	0	0	0	0	0	1	1	2	3	4
0.0074	0	Û	0	0	0	1	1	1	1	1
0.012	0	Û	0	0	0	0	0	0	0	0
0.027	0	0	0	0	0	0	0	0	0	0
0.048	0	0	0	0	1	1	1	1	. 1	1
0.100	0	0	0	0	0	0	0	0	0	0
Control Solvent Cont.		59.1	(n=9) (n=7)	59	3.4 (n=9) 2.0 (n=10	59.	8 (n=10) 0 (n=9)	5	58.5 (n=1) 59.3 (n=1)	0)
0.0074		59.5	(n=10)		3.5 (a=1(	•	1 (n=9)		59.5 (u=1	-
0.012			(p=10)		9.4(n=10	•	.5 (n=10)		9.8 (n=10	•
0.027			(01=a)		3.4 (n=10	*	.9 (n=10)		50.3 (n=1	-
0.048			(n=10)		9.6 (n-10	•	1.7 (n=9)		8.6 (n=10	-
0.100		58.7	(n=10)	60	0.0 (n=10	0) 58.	.8 (n=10)	} <b>.</b>	59.0 (v=1	0)
Mean number		ar pr	oduced	durin	g 21-d.					
Nominal conc.	Days		_							
(mg/L)	0	3	5	7	10	12	14	17	19	21
Control	-	-	•	•	109	196	317	86	179	170
	-	•	•	16	164	178		240	<b>75</b>	156
Solvent Cont.	-		-	3	141	202	302	261	75	274
0.0074			_	3.5	122	206	373	221	96	265
0.0074 0.012	•				_		446	~4 D		7117
0.0074 0.012 <b>0.02</b> 7		-	-	8.3	150	189	317	218	138	313
0.0074 0.012	-	-	-		150 113 135	189 203 186	242 223	120 180	233 93	214 269

## Statistical results as appropriate:

Calculated LC<sub>50</sub> Value for Parental Daphnia: LC<sub>50</sub>(21day) > 0.082(mg/L) Calculated EC<sub>50</sub> value for Inhibition of Reproduction: EC<sub>50</sub>(21day) > 0.082(mg/L)

### Remarks field for Results:

Biological observations

Cumulative numbers of dead parental Daphnia: Control: 2 (mortality: 5%), Solv. Cont.: 4 (mortality: 10%) 0.0074 mg/L: 1 (mortality: 2.5%) 0.012 mg/L: 0 (mortality: 0%) 0.027 mg/L: 0 (mortality: 0%)

0.048 mg/L: 1 (mortality: 2.5%)

0.100 mg/ L: 0 (mortality: 0%)

Time of the first production of juveniles:Control: 7-10d

Solvent control: 5-7d 0.0074 mg/L: 5-7d 0.012 mg/L: 5-7d 0.027 mg/L: 5-7d

0.048 mg/L: 7-10d 0.100 mg/ L: 5-7d

Mean cumulative numbers of juveniles produced per adult alive for 21days:

Control: 112.7
Solvent control: 168.5
0.0074mg/L: 119.6
0.012 mg/L: 139.3
0.027 mg/L: 133.3
0.048 mg/L: 116.0
0.100 mg/L 112.9

Was control response satisfactory: Yes.

### CONCLUSIONS

·NOEC (21-d, reproduction): 0.082 mg/L, ·LOEC (21-d, reproduction): >0.082 mg/L, ·ECso (21-d, reproduction): >0.082 mg/L; ·LCso for parental Daphnia (21-d): >0.082 mg/L

### DATA QUALITY

- Reliabilities:
- Remarks field for Data Reliability:

Experimental design and analytical procedure were well documented. Carried out by Analytical Biochemistry Laboratories, Inc.,

### REFERENCES

CMA Doc. I.D. 40-8565036 (1985).

- · Last changed:
- · Order number for sorting :
- Remarks field for GeneralRemarks:

## CHRONIC TOXICITY TO AQUATIC INVERTEBRATES (e.g., DAPHNIA) (2)

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Tokyo Kasei Kogyo Co., Ltd. Lot No. AX01

Purity: >95.0%

METHOD

Method:

OECD TG 211 (revised edition of No.202).

Test type:

Semi-static.

• GLP:

Yes

Year:

1998

Analytical procedures:

Yes. Measured by HPLC 2-3 times a week (before and afterthe

replacement of the test water)

Species/Strain:

Daphnia magna

Test details:

Semi-static (water renewal: 3 times a week), open-system.

Statistical methods:

Eco-Statics (Version 1.0 beta-edition R1.4)

### Remarks field for Test Conditions:

Test organisms:

Source, supplier, any pre-treatment, breeding method: Supplied by NIES

(Japan).

Age at study initiation: Juveniles within 24h old.

Control group: Yes.

Test conditions

Stock solutions preparation and stability: No solvent used. Test chemical was diluted to 1.0wt.% (with solubilizer HCO-40 1.0wt.% controlled)

with diluting water (Elendt M4) before use. Solubilizer concentration was controlled 100mg/L with working solution (HCO-40 1.0wt,%), Test temperature range: 19. 9-20.8 °C (average temperature 20°C).

beakers per treatment

Dilution water source: Elendt M4(OECD guideline No.211 Annex 2)

Exposure vessel type: 80mL test solution in a 100 mL glass beaker; 10

Dilution water chemistry: Hardness: 251mg/L as CaCO<sub>3</sub>

Lighting: <1,200 lx, 16h:8h light-darkness cycle

Water chemistry in test: DO= 7.0-9.2mg/L; pH=7.4-7.9. Feeding: Chlorella regularis, 0.1-0.2 mgC/day/individual

Element (unit) basis:

Mean cumulative numbers of juveniles produced per adult (reproduction)

Test design:

Number of replicates=10; individuals per replicate=10;

Concentrations: 0, 55.6, and 100 mg/L, because 48h-EiC<sub>50</sub> for parent Daphnia (Acute immobilization test) was>180 mg/L. Dispersant control

was also tested.

Method of calculating mean measured concentrations:Geometric mean.

Exposure period:

**21** d

Analytical monitoring:

By HPLC analysis. 99.7-101.3% of the nominal concentration at

preparation; 94.7-99.3% just before the renewal of the test water (after 2

days exposure).

#### RESULTS

- Nominal concentrations: 0, 55.6, 100 mg/L
- Measured concentrations: Time-weighted measured concentrations of test chemical during a 21-day exposure were 54.8 and 98.7 mg/L.

## Measured concentration of test chemical during 21-day exposure

Nominal concentration	Measured concentration (day, mg/L)							
(mg/L)	0(new)	2 (old)	7(new)	9(old)	16(new)	19(old)		
Control	< 1.0	< 1.0	< 1.0	< 1.0	< 1.0	< 1.0 ´		
Disp.Cont.	< 1.0	<1.0	< 1.0	< 1.0	< 1.0	<b>&lt;</b> 1.0		
55.6	56.3	54.4	55.4	53.9	56.3	52.6		
100	100.4	99.3	100.0	98.5	99.8	95.2		

new: freshly prepared test solutions. old: test solution after 2 days exposure.

Unit:

nig/L

- ·NOEC (21-d, reproduction): 55.6 mg/L,
- ·LOEC (21-d, reproduction): >100 mg/L,
- -ECse (21-d, reproduction): 89.1 mg/L;
- LCso for parental Daphnia (21-d): >100 mg/L; calculated based on nominal concentrations.

### Cumulative Number of Dead Parental Daphnia.

Nominal conc.	Day	<b>73</b>					_														
(mg/L)	1	2	3	4	5	б	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21
Control	0	0	0	0	0	0	0	0	0	0	0	Ø	0	Ď.	0	0	0	0	0	O	0
Disp.Cont.	0	0	0	0	0	0	0	0	. 0	0	0	0	0	0	0	0	0	0	0	0	0
<b>55.</b> 6	0	0	Û	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
100	0	Ô	0	0	0	0	0	0	0	0	0	0	0	0	0	0	1	2	2	2	2

### Mean cumulative numbers of juveniles produced per adult during 21-d.

Nominal conc.	Days																			
(mg/L)	1 2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21
Control	0.0 0.0	0.0	0.0	0.0	0.0	0.0	1.8	2.2	7.1	7.7	8.2	19.6	20.4	23.2	43.	8 48.0	61.6	83.0	88.0	88.7
Disp.Cont.	0.0 0.0	0.0	0.0	0.0	0.0	0.0	0.3	0.3	8.2	8.2	8.7	29.2	31.9	33.0	55.	8 61.5	64.8	72.0	73.8	3 73.8
55.6	0.0 0.0	0.0	0.0	0.0	0.0	0.0	0.2	1.0	2.0	2.7	5.1	9.3	13.6	26.6	34.	4 43.9	51.4	66.2	74.	79.9
100	0.0 0.0	0.0	0.0	0.0	0.0	0.0	0.0	0.0	1.6	2.6	3.6	7.8	9.3	11.0	15.	1 17.5	20.3	30.3	33.C	0.88 C

### Cumulative Number of Juveniles produced per Adult Alive for 21-d.

Vessel No.	Cont.	Disp.Cont.	Nominal 55.6	Concentration(mg/L) 100.0
1	74	74	68	37
2	57	71	70	25
3	126	92	65	•
4	127	78	96	<u>.</u>
5	90	73	89	36
6	84	70	116	29
7	71	76	78	35
8	94	84	93	28
9	<i>7</i> 8	75	87	34
10	86	45	37	40
Mean (S.D)	88.7(22.524)	73.8(12.072)	79.9(21.533	33.0(5.127)
Inhibition rate(9	6)	0.832	0.901	0.372

Significant difference\*1

-: were not calculated because the parental Daphnia was dead during a 21-days testing period,

1\*: Indicates a zignificant difference by Dunnet multiple comparison procedure, Two-sides test.

\*\*:Indicates a significant difference (alpha=0.01) from the control.

Statistical results as appropriate:

Calculated LC<sub>so</sub> Value for Parental Daphnia: LC<sub>so</sub>(21day) >100(mg/L)

Calculated EC<sub>so</sub> value for Inhibition of Reproduction:  $EC_{so}(21\text{day}) = 89.1(\text{mg/L})$ 

(Statistical method: Logit)

#### Remarks field for Results:

Biological observations

Cumulative numbers of dead parental Daphnia: Control: 0 (mortality: 0%),

Disp.Cont.: 0(mortality: 0%) 55.6 mg/L: 0(mortality: 0%)

100 mg/L: 2 (mortality: 20%)

Time of the first production of juveniles: 8-13d for control

8-12d for dispersant control

8-13d for 55.6 mg/L

10-14d for 100 mg/L

Mean cumulative numbers of juveniles produced per adult alive for 21days:

Control: 88.7, Dispersant control: 73.8

55.6 mg/L: 79.9, 100 mg/L: 33.0

Was control response satisfactory: Yes. Mean cumulative numbers of uveniles produced per adult was 88.7 and 73.8 > 60.

#### CONCLUSIONS

-NOEC (21-d, reproduction): 55.6 mg/L,

·LOEC (21-d, reproduction): >100 mg/L,

·ECs (21-d, reproduction): 89.1 mg/L:

·LCso for parental Daphnia (21-d): >100 mg/L; calculated based on nominal concentrations.

### DATA QUALITY

Reliabilities:

Klimisch Code: 1=reliable without restrictions.

Remarks field for Data Reliability:

Experimental design and analytical procedure were well documented.

Carried out by Toray Research Center (Japan).

#### REFERENCES

Environment Agency of Japan (1998).

- Last changed:
- Order number forsorting:

Remarks field for GeneralRemarks:

# HEALTH ELEMENTS

#### ACUTE ORAL TOXICITY

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Daihachi Kagaku Kogyo Co., Ltd. Lot. No. N-60601

Purity: >99.0%

Kept at room temperature in a dark place until use. Stability of mixture of

dose was confirmed for 7 days under 4C.

METHOD

Method:

OECD TG 401

Test type:

Single Dose Oral Toxicity Test

GLP:

Yes

Year:

1996

Species:

Rat

Strain:

Crj: CD(SD)

Route of administration:

Oral (by single-dosegavage)

Doses/concentration levels: 0(vehicle) and 2,000 mg/kg

Sex:

Male & Female

Vehicle

Com oil

Post exposure observation period: Two weeks.

Statistical methods:

Not applicable because of no fatality.

REMARKS FIELD FOR TEST CONDITIONS

Test Subjects:

Age at study initiation: 6 weeks old for both sexes.

Weight at study initiation: 149-163 g for male.

126-140 g for female

No. of animals per sex per dose: 5 per sex per dose group

Study Design:

Vehicle: Com oil. 40.0w/v% for 2000 mg/kg.

Satellite groups and reasons they were added: None

Clinical observations performed and frequency:

Each rat was weighed immediately prior to treatment,7 and 14 days after post-treatment observation period. The rats were observed each hour to 6hr, after that, 2 times for one day during this time for signs of toxicity.

RESULTS

LD<sub>50</sub>:

Malc :> 2.000 mg/kg

Female: > 2,000 mg/kg

REMARKS FIELD FOR RESULTS.

# DRAFT ENV/JM/EXCH(99)13

Body weight:

The test substance did not cause any changes in body weight.

No detailed body weight data available.

Food/water consumption: No detailed data available.

Clinical signs :

Loosening erring of the stool attributable to the treatment with corn oil was observed for 3 hours from the administration for both sexes in the groups given 0 and 2000 mg/kg. However, no deaths occurred of either male or

female animals.

Haematology:

Not done

Biochem:

Not done.

Ophthalmologic findings: Not examined.

Mortality and time to death: No deaths were recorded in treated and control group.

Gross pathology incidence and severity: No macroscopic abnormalities that could be attributes to

treatment with the test substance were seen on pathological examination.

Organ weight changes:

Not done.

Histopathology (incidence and severity): Not done.

## CONCLUSIONS

 $LD_{40}$  was established at > 2,000 mg/kg for both sexes.

# DATA QUALITY

Reliabilities:

Klimisch Code: 1=reliable without restrictions.

Remarks field for Data Reliability:

Well conducted study, carried out by the Biosafety Research Center, Food, Drugs and Pesticides (An-pyo Center), Japan

#### REFERENCES

Toxicity Testing Reports of Environmental Chemicals, vol.4(1996)

Ministry of Health & Welfare, Japan

# ACUTE INHALATION TOXICITY

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Nouplaz 6959, Batch No. 39049

Purity: 98.95%

#### METHOD

Method;

Not specified

• GLP:

Yes

· Year:

1982

Species:

Rat

Strain:

Crj: CD(SD)

Doses/concentration levels: 2,600 mg/m³

Sex:

Male & Female

Post exposure observation period: Two weeks.

Statistical methods:

Not applicable because of no fatality.

# REMARKS FIELD FOR TEST CONDITIONS

Test Subjects:

Age at study initiation: Not stated.

Weight at study initiation: 210-275 g for both sexes.

No. of animals per sex per dose: 5 per sex per dose group

Study Design:

Inhalation Chamber: A 0.5m3 stainless steel inhalation chamber was used.

( Youg and Bertke, Cincinnati, Ohio)

The test compound atmosphere was generated directly into the chamber by

means of Jet Nebulizer Mechanism. Chamberconcentrations were

monitored by a filter paper/gravimetric techniqueapproximately every 30

min during the exposure period.

The HEPA filtered chamber air-flow was maintained between 10 to 20 air changes per hour during the exposure period with the chamber under

slightly negative pressure.

The temperature in the chamber was maintained at 69-75 degree F with

relative humidity of 30-50%

Satellite groups and reasons they were added: None

Clinical observations performed and frequency:

After the exposure, all animals were observed daily for 14 days for clinical signs of toxicity. Body weights were recorded prior to exposure and weekly thereafter. All animals were subjected to necropsy at termination of the

study.

#### RESULTS

• LD<sub>a</sub>:

Male : > 2,600 mg/m<sup>3</sup>

Female: > 2,600 mg/m $^{3}$ 

#### REMARKS FIELD FOR RESULTS.

Body weight:

The test substance did not cause any changes in hody weight.

Mean body weight(g) of rats exposed to this chemical

Males

Initial weight

265.1(8.40)

First week

297.8(14.02)

Second week

329.7(15.27) 213.9(2.66)

Females

Initial weight

First week

223.2(3.96)

Second week

238.1(4.82)

Food/water consumption:

No detailed data available.

Clinical signs:

All animals (male and female) had matted, drenched coats for the first 2

Mean(S.D.)

days, otherwise no visible signs.

Haematology:

Not done.

Biochem:

Not done.

Ophthalmologic findings:

Not examined.

Mortality and time to death: No deaths were recorded.

Organ weight changes:

Not done.

General necropsy observations: All males and 3/5 females exhibited reddening patches on lungs.

# CONCLUSIONS

LD<sub>0</sub> was 2,600 mg/m<sup>3</sup> for both sexes.

## DATA QUALITY

Reliabilities:

Klimisch Code: 1=reliable without restrictions.

Remarks field for Data Reliability:

Well conducted study, carried out by Midwest Research Institute.

#### REFERENCES

Nuodex Inc. Acute inhalation toxicity test in SpragueDawley rats using compoundNouplaz 6959

Environmental Protection Agency (1983)

#### ACUTE DERMAL TOXICITY

## TEST SUBSTANCE

· Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Nouplaz 6959, Batch No. 39049

Purity: 98.95%

**METHOD** 

Method:

Procedure set forth in the Federal Insecticide, Fungicide, and Rodenticide

Act (FIFRA)

• GLP:

Yes

Year:

1981

Species:

Rabbits

Strain:

New Zealand albino white rabbits

Doses/concentration levels: 2.0 mL/kg

Sex:

Male & Female

Post exposure observation period: Two weeks.

Statistical methods:

Not applicable because of no fatality.

# REMARKS FIELD FOR TEST CONDITIONS

Test Subjects:

Age at study initiation: Not stated.

Weight at study initiation: 2.3-3.2 kg for both sexes.

No. of animais per sex per dose: 3 per sex per dose group and 2 per sex

for control.

Study Design:

Procedure: 24 hours prior to treatment the hair on the back of each rabbit was clipped so as to expose approximately 10% of the body surface area. Before dosing, epidermal abrasions were made longitudinally over the exposure area. The abrasions were sufficiently deep to penetrate the

stratum comeum but not so deep as to cause bleeding.

A dosage was applied to the exposure area. A 2 x 2-inch

A dosage was applied to the exposure area. A 2 x 2-inch gauze pad was placed on the exposure area to prevent seepage of the compound from the area. Each animal was then wrapped with a rubber dam. After 24 hour of exposure, the rubber dam and gauze pad were removed, and the exposure

area was wiped to remove any remaining test material. Satellite groups and reasons they were added: None Clinical observations performed and frequency:

After the exposure, all animals were observed daily for 14 days for clinical signs of toxicity. A gross necropsy was performed on all animals at the end

of the 14 day observation period.

# RESULTS

LD,;

Male : > 2.0 mL/kg Female: > 2.0 mL/kg

#### REMARKS FIELD FOR RESULTS.

Body weight:

The test substance did not cause any changes in body weight.

Individual Animal Boy Weights

	Sex	Body	weight (kg)	
Control		day 1	day 7	day 14
	male	3.2	3.4	3.6
	•	3.2	3.4	3.6
	Temale	2.7	3.0	3.1
		2,9	<b>3</b> .1	3.3
2.0 mL/kg	maje	2.3	2.3	2.5
•		2.4	2.4	2.5
		2.3	2.2	2.4
	female	2,3	2.5	2.7
		2.4	2.6	2.7
		2.4	2.5	2.6

Food/water consumption: No detailed data available.

Clinical signs:

No toxic sign.

Haematology:

Not done

Biochem:

Not done.

Ophthalmologic findings: Not examined.

Mortality and time to death: No deaths were recorded.

Organ weight changes:

Not done.

Gross Pathology:

Nothing noted.

## CONCLUSIONS

LD<sub>50</sub> was 2.0 mL/kg for both sexes.

# DATA QUALITY

Reliabilities:

Klimisch Code: 1=reliable without restrictions.

Remarks field for Data Reliability:

Well conducted study, carried out by Midwest Research Institute.

## REFERENCES

Nuodex Inc. Acute dermal toxicity test of Tenneco Chemicals Inc. compoundNouplaz 6959 in

rabbit.

Environmental Protection Agency (1981)

#### SKIN IRRITATION

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Nouplaz TOTM(Tenneco Chemicals, Inc.)

Purity: 98.95%

#### METHOD

The test method was similar to Section 1500.41. Federal Hazardous Method: Substances Act Regulations - 16 CFR

GLP:

Yes

Year:

1981

Species:

Rabbits

Strain:

New Zealand albino white rabbits

- Doses/concentration levels: 0.5 mL
- Sex:
- Post exposure observation period:24, 72 hours
- Statistical methods:

Not applicable because of no fatality.

## REMARKS FIELD FOR TEST CONDITIONS

Husbandry Conditions Temperature - 70 ± 2 degree F Relative Humidity - 45% ± 5%

Light - 12 hour light/dark cycle

Diet - Wayne 15% Rabbit Ration and tap water are provided ad

libitum. Based on our current knowledge no contaminants are known to be in this diet or water that might be expected to

interfere with the objectives of the study.

Caging - Stainless steel with elevated wire mesh flooring 1 rabbit/cage

Bedding - Techbord

Shepherd Products Company Kalamazoo, Michigan 49005

Test method:

A 0.5 mL portion of material was applied to an abraded and an intact akin site on the same rabbit. Gauze patches were then placed over the treated areas and an impervious material was wrapped snugly around the trunks of

the animals to hold the patches in place.

The wrapping was removed at the end of the twenty-four (seventy two) hour period and the treated area were examined. The Draize method of

scoring was employed.

Evaluation: Draize Scale For Scoring Reactions

Erythema and Eschar Formation:	Value
No erythema	0
Very slight erythema(barely perceptible)	1
Well defined erythem	2
Moderate to severe erythema	3
Severe erythema (beet redness) to slighteschar formation	
(injuries in depth)	4

Edema Formation	Value
No edema	0
Very slight edema(barely perceptible)	1
Slight edema(edges of area well defined by definite raising)	2
Moderate edema (raised approximately 1 millimeter)	3
Severe edema (raised more than 1 millimater and extending	
beyond the area of exposure)	4

# RESULTS

Primary Irritation Score: 4.16/4 = 1.04

## REMARKS FIELD FOR RESULTS.

	Reading			Rabb	it Nu	wper		
Erythema and Eschar Formation	(Hours)	1	2_	3	4	5_	6	Average
Intact skin	24	2	1	2	1	2	1	1.50
]ntaet skin	72	6	0	1	0	()	0	0.17
Abraded skin	24	2	1	2	1	2	1	1.50
Abraded skin	72	•	0	1	1	0	0	0.33
						Subte	ial	3.50
Edema Formation								
Intact skin	24	1	0	0	0	1	0	0.33
Intact skin	72	Ð	•	0	0	0	Ω	0.00
Abraded skin	24	1	Ð	Ď	0	1	•	0.33
Abraded skin	72	D	Ð	0	0	0	0	0.00
• •					:	Subtoi	lal	0.66
						To	taj	4.16

## CONCLUSIONS

Slightly irritating

This report concluded that TOTM was not a primary skin irritant in rabbit. It is not possible to assign a classification.

# DATA QUALITY

- Reliabilities: Klimisch Code: 1= reliable without restrictions.
- Remarks field for Data Reliability:
   Well conducted study, carried out by Biosearch Inc.

## REFERENCES

Nuodex Inc. Primary Skin Irritation - Rabbits. OTS 2065758. Doc ID 878214470,1981

#### EYE IRRITATION

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)bcnzene-1,2,4-tricarboxylate

Remarks:

Source: Nouplaz TOTM(Tenneco Chemicals, Inc.)

Purity: 98.95%

#### METHOD

Method:

The test method was similar to Section 1500.42. Federal Hazardous Substances Act

Regulations - 16 CFR.

GLP:

Yes

Year:

1981

Species:

Rabbits

Strain:

New Zealand albino white rabbits

Numbers of animals

Doses/concentration levels: 0.1 mL

Sex:

Post exposure observation period: 1,2,3,4,7 days

Statistical methods:

Not applicable because of no fatality.

# REMARKS FIELD FOR TEST CONDITIONS

Husbandry Conditions Temperature - 70 ± 2 degree F

Relative Humidity - 45% ± 5% Light - 12 hour light/dark cycle

Diet - Wayne 15% Rabbit Ration and tap water are provided ad libitum. Based on our current knowledge no contaminants are known to be in this diet or water that might be expected to

interfere with the objectives of the study.

Caging - Stainless steel with elevated wire mesh flooring 1 rabbit/cage

Bedding - Techbord

Shepherd Products Company Kalamazoo, Michigan 49005

Test method:

0.1 mL of the experimental material was instilled into the right eyes of the test animals while the other eyes remained untreated to severe as controls. The treated eyes were examined at one, two, three, four and seven days

Following instillation of the test materials into the eyes.

Evaluation:

Interpretation of the results was made in accordance with theDraize Scale

of Scoring Ocular Lesions.

Scale of Scoring Ocular Lesions

(1) CORNEA

Value range

A. Opacity - Degree of Density(area most dense taken for reading)

B. Area of Cornea Involved

0 - 4 1 - 4

Score equals  $A \times B \times 5$  (Total Maximum = 80)

(2) IRIS

A. Values	0 - 2
Score equals A x 5 (Total Maximum = 10)	
(3) CONJUNCTIVAE	
A. Redness (refers to palpebral and bulbar conjunctivae	
excluding cornea and iris)	0 - 3
B. Chemosis	0 - 4
C. Discharge	0 - 3
Score equals (A+B+C) x 2 (Total Maximum = 20	)

# RESULTS

Average Ocular Irritation Score: 2.3(1 day), 1.7(2day), 0(3,4,7day)

# REMARKS FIELD FOR RESULTS.

Rabbit	number Tissue	I day	2 day	3 day	4 dar	7day
1	(1) Cornea total	0	0	0	0	Ö
	(2) Iris total	0	0	0	0	Ð
	(3) Conjunctivae total	2	2	0	0	0
	Total Ocular Irritation Score	2	2	0	0	0
2	(1) Cornea total	0	0	0	0	0
	(2) Iris total	0	0	0	0	D
	(3) Conjunctivae total	4	2	Ð	0	0
	Total Ocular Irritation Score	4	2	0	0	0
3	(1) Cornea total	0	0	0	Ð	0
	(2) Iris total	0	0	0	0	0
	(3) Conjunctivae total	2	2	0	O	0
	Total Ocular Irritation Score	2	2	0	Û	0
4	(1) Cornea total	0	0	0	Ð	0
	(2) Iris total	0	Ð	Q	0	0
	(3) Conjunctivae total	2	2	Ð	0	0
	Total Ocular Irritation Score	2	2	0	0	0
5	(1) Cornen total	Ġ	0	Ō	0	0
	(2) Iris total	Ü	Ð	9	6	0
	(3) Conjunctivae total	2	2	0	0	0
	Total Ocular Irritation Score	2	2	Ò	0	0
6	(1) Cornea total	Ð	0	0	0	Û
	(2) Iris total	0	0	0	0	0
	(3) Conjunctivae total	2	0	Đ	0	ð
	Total Ocular Irritation Score	2	0	0	0	0
	Average Ocular Irritation Score	2.3	1.7	0.0	0.0	0.0

# CONCLUSIONS

Slightly irritating

This report concluded that TOTM was not a primary skin irritant in rabbit. It is not possible to assign a classification.

# **DATA QUALITY**

- Reliabilities: Klimisch Code: 1=reliable without restrictions.
- Remarks field for Data Reliability:

Well conducted study, carried out by Biosearch Inc

# REFERENCES

Nuodex Inc. Primary Eye Irritation - Rabbits. OTS 2065758. Doc ID 878214471,1983

# DRAFT ENV/JM/EXCH(99)13

#### SENSITIZATION

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Nouplaz TOTM(Tenneco Chemicals, Inc.)

Purity: 98.95%

#### **METHOD**

Method:

Buehler test

GLP:

Yes

Year:

1981

Species:

Guinea pig

Strains

Albino guinea pig

Numbers of animals

10

Doses/concentration levels: 0.5 mL

Post exposure observation period:10 application

Statistical methods:

Not applicable because of no fatality.

## REMARKS FIELD FOR TEST CONDITIONS

Husbandry Conditions Temperature - 70 ± 2 degree F

Relative Humidity-45% ± 5%

Light - 12 hour light/dark cycle

Diet - Charless River Guinea Pig Furmula and tap water are provided ad Libitum, Based on our current knowledge no contaminants were Known to be in this diet or water that might be expected to

Interfere with the objectives of the study.

Caging - Stainless steel with elevated wire mesh flooring 5 guinea pigs/cage

Bedding - Deotized Animal CageBoard(DACB)

Shepherd Products Company Kalamazoo, Michigan 49005

Test method:

A 0.5 mL portion of material was applied to the intact akin test site on the guinea pigs. A gauze patch was placed over the treated area and an impervious material was wrapped snugly around the trunks of the animals to hold the patches in place. After a 24 hour contact period the patch was removed and the animals were allowed to rest for one day. Following this rest period another application was applied to the same skin site using a fresh sample. After the tenth application the animals were rested for a two week period. At the termination of the rest period a challenge application was put on skin sites differing from the original test sites. The challenge application remained on for 24 hours.

The sites were examined for reaction using the Draize method of scoring to grade reactions.

Evaluation: Draize Scale For Scoring Reactions

Ervihema and Eschar Formation:

Value

No erythema

eptible) 1
2
3
to slighteschar formation(injuries in depth) 4
Value
0
tible) 1
defined by definite raising) 2
mately 1 millimeter) 3
1 millimater and extending
- 4
֡֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜֜

#### RESULTS

## No sensitization

# REMARKS FIELD FOR RESULTS.

	Reading After Application number							Challenge					
Guinea pig No.			2	_3	4	5	6	7	8	9	10	24hou	ra 48hours
1	Erythema	0	0	0	0	0	0	0	O	Ð	0	0	•
	Edema	0	0	Û	0	0	0	0	<b>a</b>	0	D	0	0
2	Erythema	0	0	Ü	0	0	Ð		0	D	0	•	0
	Edema	•	0	D	0	0	0	0	0	0	0	0	. 0
3	Erythema	0	0	Ð	0	Đ	0	0	ø	0	0	0	0
	Edema	0	Ð	D	0	0	0	D	0	ũ	0	0.	0
4	Erythema	0	0	D	0	0	0	0	0	0	0	0	0
	Edema	0	0	0	Ð	0	0	D	Ð	0	0	Ð	0
5	Erythema	0	0	Ð	Φ	0	0	Û	0	0	O	G	0
	Edema	0	0	0	D	Ô	0	0	0	0	Ð.	0	0
6	Erythema	Ð	0	0	0	0	0	0	0	0	0	0	O
	Edema	0	0	0	0	0	•	0	0	0	Ð	0	0
7	Erythema	0	0	0	0	0	Ð	0	0	0	0	0	0
	Edema	0	œ	G	0	0	0	0	ø	Q	D	0	•
8	Erythema	Ď	0	0	0	Ü	0	0	0	0	0	Ð	0
	Edema	9	0	0	0	0	0	0	0	0	0	0	0
9	Erythema	D	0	Ð	0	0	0	0	0	0	0	Ð	Ð
-	Edema	0	0	0	0	0	0	0	Ð	0	0	0	0
10	Erythema	0	0	0	0	0	0	0	0	0	G	0	0
	Edema	0	0	. 0	0	0	Ð	0	0	Ð	0	0	0

## CONCLUSIONS

No senstization

# DATA QUALITY

• Reliabilities:

Klimisch Code: 1=reliable without restrictions.

Remarks field for Data Reliability:

Well conducted study, carried out by Biosearch Inc.

## REFERENCES

Nuodex Inc. Guinea Pig Contact Dermal Irritatiom/Sensitization-Modified Buehler Method OTS 206574. Doc ID 878214475,1981

# REPEATED DOSE TOXICITY (a)

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Nuoplaz 6959

Purity: 98.2% (GC/FID) 97.9% (HPLC)

Impurities were detected at level than 0.1-0.5%, one being di(2-ethylhexyl)

phthalate (DEHP).

METHOD

Method:

BIBRA Standard Operating Procedures

Test type:

Repeat Dose Toxicity

GLP:

Yes

Year:

1984 Rat

Species:

Strain:

Fischer 344

Route of administration

Oral

Doses/concentration levels: 0(0), 0.2(184), 0.67(650) and 2(1826) % (mg/kg bw/day) Vehicle:

Rodent diet

Sex:

Male & Female

Exposure period:

28 days

Frequency of treatment:

Once daily

Control group and treatment: Dietary level 0% and reference compound DEHP 0.67%.

Post exposure observation period: None

Duration of test:

Males and females; for 28 days

Statistical methods:

The control and TOTM treated groups were subject to analysis of

variance, and if this was significant the treated groups were compared with

the controls using the Least Significant Difference test.

The controls and DEHP groups were compared using a two-tailed pooled student t test with Welch's correction. In all cased a probability level of

P<0.05 was taken to indicate statistical significance.

#### REMARKS FIELD FOR TEST CONDITIONS

Test Subjects:

Age at study initiation: 48-51 days old for males and females

Weight at study initiation: 137-154g for male.

111-132g for female.

No. of animals per sex per dose: 5 Rats per sex per dose group

Study Design:

Vehicle: Dict

Satellite groups and reasons they were added: None Clinical observations performed and frequency:

Body wt. was recorded immediately prior to the first exposure and again for each

animal 1, 3, 7, 10, 14, 17, 21, 24, 27 th days.

Twice each day the animals were observed in their cages forvariations in behaviour or condition, and once weekly a more detailed examination was made at the time of

a weighing.

Food intakes were measured over the period day-3 to 0 and continuos intakes were measured at twice-weekly intervals until the day preceding autopsy. The intakes of test article or reference compound for each animal were calculated twice weekly using the analysed dietary concentrations of TOTM or DEHP, and the individual valued for bodyweight and food intake.

Hematologic parameters were evaluated for each animal. On the day preceding the start of the autopsies a sample of blood was collected from a caudal vein of each

Autopsy: At the end of the 28th day treatment period the rats were deprived of food overnight, with water available. On the day of autopsy each animal was weighted and then killed. The blood was used to provide serum for clinical chemistry. During the autopsy any abnormalities of the external condition and of the thoracic or abdominal viscera were noted.

Organs: The weight of the following organs were recorded: adrenal glands, lungs. brain, ovaries, heart, spleen, kidneys, testes, liver, thyroid.

Electron microscopy: Two thin slices of liver, one from the left lobe, the other from the median lobe, were fixed for analysis. (The remainder of the liver was used for biochemical analysis.)

Biochemical analysis of the liver: Whole homogenates were prepared and assayed for protein and cyanide-insensitivepalmitoyl-CoA.

## RESULTS

NOAEL

184 mg/kg bw

LOAEL

650 mg/kg bw

#### REMARKS FIELD FOR RESULTS.

Body weight:

No statistically significant differences of bodyweight between the control and TOTM or DEHP treated groups of either sex. There was a trend for the male rats from all the TOTM treated groups to be lighter than the controls. In the females, this trend was only evident in the 2.0% TOTM group.

Food/water consumption: Female rats fed 2.0% TOTM consumed significantly less diet than the controls during first seven days of treatment after which their intakes increased but remained lower than those of the controls. In the males there were no statistically significant differences between the control and TOTM fed groups during the treatment period.

Haematology: In both sexes haemoglobin concentration of the rats given diet containing 0.67 or 2.0% TOTM were statistically significantly lower than the control. In the males there was a small lowering of erythrocyte count in all groups given TOTM but this was not reproduced in the females.

> Both sexes given the two higher dietary concentrations of TOTM had higherleucocyte counts than the control, but the differences were statistically significant only nthe males. These male groups also had lower proportions of the leucocytes axosinophils and monocytes.

Significantly lower values for haemotocrit and mean cell volume were limited to females given the two lower dose levels of TOTM.

Organ weights: In both sexes the liver weights, and liver weights relative to bodyweight, were increased in the TOTM and DEHP treated animals compared to the controls. These differences were small and not statistically significant in the 0.2% TOTM group. The increase seen in the rats given 2.0% TOTM was less than that in those given DEHP. In the males fed TOTM the higher values for brain weights relative to body weight, in the absence of any significant differences in the recorded weight probably reflect the lower bodyweights in the groups concerned. In the females there were statistically significant higher lung weights in the rats fed 0.2 or 0.67% TOTM when compered to the controls. In the case of the TOTM treatedanimals this difference was not dose related and not statistically significant when expressed relative to bodyweight.

Serum analyses: Analysis of serum from the males and females showed statistically significantly increased levels of albumin in the groups given 0.67 or 2.0% TOTM. In the males there were statistically significantly higher cholesterol levels in the 0.67 and 2.0% TOTM groups.

> Concentrations of serum urea were statistically significantly increased in the male 2.0% TOTM group to the control values. In the females there was also an isolated statistically significantly lower value for lipid concentration in the 0.2% TOTM group.

Liver Biochemistry: Neither TOTM or DEHP treatment influenced to a statistically significant degree the concentration of hepatic protein. After TOTM treatmentPCoA activity was statistically significantly higher than controls in both sexes at the highest dose and in the males at the lower two doses. In the groups given TOTM only the highest dose level males had statistically significant increases of enzyme level. Both sexes given 0.67 or 2.0% TOTM had statistically significantly increased carnitine acetyltransferase activity with little difference between the two sexes.

Histology:

No abnormalities were detected in themajority of the animals. The only lesions occurring with any frequency were focal interstitial pneumonitis and nephrocal cinosis in the females. The observations were not firmly dose related. Thoneumonitis was of limited extent, often only a single focus. Two female rats fed 2.0% TOTM showed reductions in sytoplasmic basophilia in liver although it was only marginal.

Electron Microscopy: In the hepatocytes from the control rats theperoxisomes varied in size from small to moderately large. They had uniformly electron dense contents and some possessed a lattice core. They were ubiquitously distributed throughout the cytoplasm. Feeding diet containing 2% TOTM produced a slight increase in the numbers of peroxisomes, which varied between cells. No difference was seen between the centrilobular and periportal areas.

#### CONCLUSIONS

The NOAEL for repeated dose toxicity is considered to be 184 mg/kg and the LOAEL is Considered 650 mg/kg for both sexes.

# DATA QUALITY

Klimisch Code: 1=reliable without restrictions. Reliabilities:

Remarks field for Data Reliability:

Well conducted study, carried out by the British Industrial Biological Research Associations

# DRAFT ENV/JM/EXCH(99)13

# REFERENCES

Chemical Manufacturers Association, Project No. 3.0496. Report No. 0496/1/85

CMA Reference. TM-3.0-BT-BIB

# REPEATED DOSE TOXICITY (b)

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Daihachi Kagaku Kogyo Co., Ltd. Lot. No. N-60601

Purity: >99.0% Kept at room temperature in a dark place until usc.

METHOD

Method:

Guidelines for 28-day Repeated Dose Toxicity Testing of Chemicals

(Japan)

Test type:

Repeat Dose Toxicity

GLP:

Yes

Vest:

1996

Species:

Rat

Strain:

Crj:CD(SD)

Route of administration

Oral

Doses/concentration levels: O(vehicle) 100, 300 and 1,000 mg/kg/day

Vehicle:

Sex

Male & Female

Exposure period:

28 days

Frequency of treatment:

Once daily

Control group and treatment: Vehicle (corn oil)

Post exposure observation period:2 weeks for 0 and 1,000 mg/kg/day dose.

Duration of test:

Males and females; for 28 days

Statistical methods:

Bartlett's test, Dunnett's test or Kruskal-Wallis test depending on whether

or not the data were nonhomogeneous or homogeneous.

Fisher 's test for the pathological result. Jonckheere's test for the

correlation of dosage

# REMARKS FIELD FOR TEST CONDITIONS

Test Subjects:

Age at study initiation: 6 weeks old for males and females

Weight at study initiation: 130-151g for male.

110-121g for female.

No. of animals per sex per dose: 5 Rats per sex per dose group

Study Design:

Vehicle: Com oil

Satellite groups and reasons they were added: None Clinical observations performed and frequency:

Body wt. was recorded immediately prior to the first exposure and again for each

animal every week

Hematologic parameters were evaluated for each animal. Bloodsamples for the hematologic determinations were taken fromabdominal artery in rats after 16 hr fast. Clinical chemistry analyses were performed on serum samples from each animal. Urinalyses were performed for each rat. Urine samples were collected from each rat

on the day prior to scheduled termination.

Organs examined at necropsy:

Organ weight: brain, liver, kidney, spleen, adrenal, spermary (male) and ovary

(females) for each animal.

Microscopic: heart, liver, kidneys, spleen, adrenal and bone marrow from rats in the

control and high-exposure groups and kidney from all dosage male.

#### RESULTS

NOAEL

Male: >1,000 mg/kg/day Female: >1,000 mg/kg/day

#### REMARKS FIELD FOR RESULTS.

Body weight: The mean body weight of treatment groups of rats for males and females ot

Significantly different from controls at any time during the course of the study.

Food/water consumption: No significantly different from controls at any time during dosing and

recovering period for both sexes.

Clinical signs: No unusual clinical observations during the study.

Males:

No dose-related change in general clinical signs. No dose-related change in general clinical signs.

Females: Haematology:

at the end of dosing

Males and females: No dose-related significant changes inhematology.

In the blood clotting test, prothrombin times for males were slightly extended, but they were considered within the physiological change. For females, no significant

changes in all test.

after recovering period

Males:

In hematology, hemoglobin amounts for males at 1000mg/kg dosing were slightly increased, but they were considered within the physiological change. In the blood

clotting test, no significant changes in all tests.

Females:

No significant change in alltests.

Biochem:

at the end of dosing

Males:

No dose-related significant adverse treatment-related effect in clinical chemistry.

Females:

At 300, and 1,000 mg/kg dosing, chlorine contents were low.

after recovering period

Males:

At 1,000 mg/kg dosing, potassium contents were slightly high.

Females:

At 1,000 mg/kg dosing, GOT were slightly high.

But both changes were considered to be no meaning, because at the end of treatment these changes were not recognised

Urinalysis:

at the end of dosing

Males and Female: At 1,000 mg/kg dosing, some of rats (both sexes), amounts of urinary increased,

but the mean urinary specific gravity values in the 1,000 mg/kg dosing group

was not significant change from control group.

after recovering period:

Males and Females: No dose-related significant change in all tests.

Mortality and time to death: No deaths prior to scheduled termination.

Organ weight changes:

at the end of dosing

Male:

No dose-related change in all testedorgans.

Female:

Relative liver weight were slightly increased at 100 mg/kg dosing, but no

dose-related change. Other organs, no significant change.

after recovering period:

Males:

At 1,000 mg/kg dosing, relative kidneyweight were slightly low.

Female:

At 1,000 mg/kg dosing, absolute and relative adrenal weight were lightly

high.

But both changes were considered no related to dosing and recovering of this chemical.

Gross pathlogy and histopathlogy:

at the end of dosing:

Males:

Coloured patch/zone of lungs were observed 1 of 100 mg/kg, 2 of 300 mg/kg and 3 animals of 1,000 mg/kg dosing group. Also hypertrophy of the kidney.

hypertrophy of parathyroid, and etc. were observed.

Amounts of eosinophilic body in the kidney were slightly increased in dosing

group. But all these changes were considered no related the dosing and

recovering of this chemical, because the degree and rate of changes were same

of all the group included control.

Females:

Red patch/zone of thymus dilated lumen of the uterus and etc. were observed. But all these changes were considered no related the dosing and recovering of this chemical, because the degree and rate of changes were same of all the

group included control.

after recovering period:

Males and Females: No dosc-related significant change in all tests.

#### CONCLUSIONS

No test substance related changes were noted in terms of clinical signs, body weight, food consumption, and hematology, blood chemical examination, urinally sis, and pathological

The NOEL for repeated dose toxicity is considered to be 1,000 mg/kg/day for both sexes.

## DATA QUALITY

Reliabilities:

Klimisch Code: 1=reliable without restrictions.

Remarks field for Data Reliability:

Well conducted study, carried out by the Biosafety Research Center, Food, Drugs and Pesticides (An-pyo Center), Japan

#### REFERENCES

Toxicity Testing Reports of Environmental Chemicals, vol.4(1996) Ministry of Health & Welfare, Japan

## TOXICITY TO REPRODUCTION

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Daihachi Kagaku Kogyo Co., Ltd. Lot. No. N-80301

Purity: >99.0% Kept at room temperature in a dark place until use.

METHOD

Method:

OECD Preliminary reproductive toxicity screening test

Test type:

Preliminary reproduction toxicity screening test.

GLP:

Yes

Yeart

1998

Species:

Rat

Strain:

Crj;CD (SD)

Route of administration:

Oral (by gavage)

Doses/concentration levels: O(vehicle) 100, 300, 1,000 mg/kg/day

Vehicle:

Corn oil

Sex:

Male & Female

Administration period:

Male: for 46 days from 2 weeks prior to mating

Female; from 2 weeks prior to mating to day 3of lactation

Frequency of treatment:

Once daily.

Control group and treatment: Vehicle (com oil)

Post exposure observation period:None.

Terminal kill

Male: day 47

Female: day 4 of lactation

Statistical methods:

Chi square test for 1 grade positive data and Fisher's test for another.

Bartlett's test or Kruskal-Wallis' test for 2 or more grade positive data. And used Dunnett's test of Mann-Whitney's U-test for examination

#### REMARKS FIELD FOR TEST CONDITIONS

Test Subjects:

Age at study initiation: 10 week old for both sexes.

Weight at study initiation: 373-435 g for males, 217-257 g for females

No. of animals per sex per dose: 12 per sex per dose group

Study Design:

The animals were sacrificed on the day 4 of lactation for females. Males and females with nomated were killed 1 day after the mating period. Females with no

delivery killed 26° day of gestation period.

Vehicle: Corn oil

Satellite groups and reasons they were added: None

Mating procedures: Male/female per cage; 1/1, length of cohabitation; with in the limit of 14 days until proof of pregnancy (formation sperm detection in vagina)

was observed.

Clinical observations performed and frequency:

Parent: General appearance once a day

Foetus: General appearance once a day after birth

Organs examined at necropsy:

Parent: Males and females: Grosspathlogy of all organs were tested. Males: Organ weight: Testis and epididymis of all animals.

Female: Organ weight: Ovary of all animals.

Count: Implantation sites and corpusluteum of ovary of all

Microscopic: Males: Testis and epididymis. Count of sertoli sells, spermatocytes, round spermatids and elongatespermatids in seminiferous tubules of 5animals of all dosing groups. (Stage I-VI, VII-VIII, IX-XI, XII-XIV of spermatozoon formative cycle.)

Females: Ovary

Pup: Gross pathlogy of all organs were tested. Dead pups and abnormal organs were tested histopathogy.

Parameters assessed during study:

Body weight. Males: Prior to the first dosing and 2, 5, 7, 10, 14 day. After that once a week, the daysacrificed. Females: Prior to the first dosing and 2, 5, 7, 10, 14 day. During gestation period, 0, 1,3, 5, 7, 10, 17 and 20 day. During lactation period, 0, 1, and 4. During cohabitation period, the same day with male. Pups: Day 0 and 4

Food/water consumption. The same day when body wt. determined except lactation period and the day sacrificed for males, also, 0 day of gestation and lactation for female.

No. of pairs with successful copulation, copulation index (No.of pairs with Successful copulation/No. of pairs mated) x 100, duration of mating No. of pregnant females, fertility index = (No.of pregnant animals/No. of pairs with successful copulation x 100, No. of corpora lutea, No. of implantation sites, implantation index (No. of implantation sites/No. of corpora lutea) x 100, No. of pups born, delivery index (No.of pups born/No. of implantation sates)x 100, No. of love pups born, live birth index (No.of love pups born/No. of pups born) x 100, sex ratio of pups, No. of dead pups born, gestation length, gestation index (No. of females with live pops delivered/ Noof pregnant females) x 100, nursing index (No. of females nursing live pups/No. of females with normal delivery) x 100, No. of live pups on day 4, viability index (No.of live pups on day 4/No. of live pups born) x 100,

#### RESULTS

Repeat dose toxicity: NOEL 100 mg/kg/day for males

1,000 mg/kg/day for female

Reproductive and developmental toxicity: NOEL100 mg/kg/day for males

1,000 mg/kg/day for female 1,000 mg/kg/day for offspring

### REMARKS FIELD FOR RESULTS.

Mortality and day of death: None.

No statistical significant difference from controls. Body weight: No statistical significant difference from controls. Food/water consumption:

Reproductive data:

No statistical significant difference from controls.

Pups data:

Body weight and weight gain of 300 mg/kg dosing group for both sexes were slightly low. But all pups of 100 and 1000 mg/kg dosing group were not statistical significant difference from controls.

At the other tests, no statistical significant difference from controls.

Grossly visible abnormalities, external, soft tissue and skeletal abnormalities: For males:

Slightly decrease of spermatocytes and spermatids: 2 animals of 300 mg/kg dosing group.

11 of 1000 mg/kg dosing group.

Moderate decrease of spermatocytes and spermatids: 1 of 1000 mg/kg/dosing group.

At this animal, a few multinucleate giant cell were appeared and slightly-acuolization of sertoli sells were observed. Also, at the epididymis, moderate amount of cell debris moderate decrease of spermatids and slightly granuloms of spermatic were observed. For the control group, atrophy of seminiferous tubule were observed 2 animals. At these animals, slightly amount of cell debris were observed one of these animals, slight decrease of spermatids was also observed.

Number of cells in seminiferous tubules:

Group 1(Stage I-VI) : Low value of spermatids at 300 mg/kg dosing group.

Low values of spermatocytes and spermatids at 1000 mg/kg dosing group. Group 2(Stage VII-VIII):Low values of round spermatids and ratio of sertoli cells at 1000 mg/kg. Group 3(stage IX-XI) :Low values of elongatespermatids and ratio of sertoli cells at 1000 mg/kg. Group 4(stage XII-XIV) :Low values of spermatocytes, elongatespermatids, and ratio of sertoli cells at 1000 mg/kg dosing group.

#### For females:

Cyst of corpus luteum of ovary was observed 2 animals of 300 mg/kg dosing group.

No abnormal ovary observed at the female of 100 mg/kg dosing without successful copulation, females of control and 100 mg/kg dosing without pregnant.

#### Histopathological finding in rats

•		dosc (mg/kg)						
Items		0	100	300	1,000			
No. of male animals examined		12	12	12	12			
Organ: Findings								
_	Grade							
Testis:								
Decrease, spermatocyte and spermatid	Total	0	0	2	12**			
	+	0	0	2	11			
	+ +	· 0	0	0	1			
Multinuclear glant cell, seminiferous tubi	ile +	0	0	0	1			
Vacuolozation, Sertoli cell	+	0	0	0	1			
Atrophy, seminiferous tubule	+	2	0	0	0			
Epididymis:								
Cell debris, lumen	Total	2	0	0	1			
•	+	2	0	0	O			
	++	0	0	0	1			
Decrease, sperm	Total	1	0	0	1			
• •	+	1	0	0	0			
	++	0	0	0	1			
Granuloma, spermatic	+	D	Ô	0	1			
_								

No. of female animals examined		12 1	2 12	12
Ovary:				
Cyst, corpus luteum	<+>	٥	0 2	0
Values are no, of animals with fine				
Grade: +=slight, ++=moderate cha	-			
Significantly different from 0 mg/	kg group: **: $p \le 0.0$	01.		
Number of cells in seminiferous tubule	s of male rats.			
Items	0	dose (n 100		1.000
No. of animals examined	5	100 5	300 5	1,000
Group 1 (Stage I-VI)	ر	,	j	5
No. of Sertoll cells	20.12(3.18)	19.08(1.49)	18 59/1 455	19 00/1 40
Spermatogonia	20.12(3.10)	13.00(1.43)	18.52(1.45)	18.08(1.45)
Брегимиводы No.	16.80(5.65)	20.52(2.58)	18.48(3.17)	1 <i>ፍ ማሪያ</i> ግ ረቀን
raijo <sup>s)</sup>	0.85(0.29)	1.08(0.19)	1.01(0.21)	15.76(2.61)
Spermatneyles	0.00(0.23)	1.00(0.13)	*****(0.51)	0.87(0.11)
No.	50.80(7.44)	51.80(4.84)	42.64(2.63)	40.84(5.63)*
ratio	2.53(0.13)	2.72(0.26)	2.37(0.24)	2.25(0.16)
Round spermatids	Lucy (U.L.)	#172(U.2U)	2.3 ((0.24)	4.23(U.10)
No.	138.36(17.20)	128.00(8.89)	117.68(5.59)*	112.60(3.11)*
ratio	6.91(0.35)	6.75(0.84)	6.39(0.70)	6.26(0.48)
Elongate spermatids			4.47(4.74)	was(0.40)
No.	130.00(21.71)	132.32(11.17)	103.28(12.34)	* 95.36(8.44)*
ratio	6.53(1.15)	6.98(0.88)	5.62(0.90)	5.30(0.69)
	<b>\</b>	V3	<b>(</b> = <b>- y</b>	
Group 2 (Stage VII-VIII)				•
No. of Sertoli cells	16.96(2.63)	17.04(2.17)	16.64(2.73)	16.52(2.23)
Spermatogonia	· · · · ·	<u>-</u>	-	` ,
No.	2.92(1.06)	2.40(0.93)	2.04(0.68)	2.60(1.10)
ratio	0.18(0.09)	0.14(0.05)	0.12(0.03)	0.16(0.06)
Spermatocytes				•
No.	91.68(10.37)	94.68(6.55)	84.44(6.99)	82.32(6.70)
ratio	5.45 (0.56)	5.60(0.51)	5.16(0.79)	5.03(0.54)
Round spermatids		•		-
No.	142.08(13.39)	, ,	. ,	, ,
ratio	8.45(0.62)	7.75(0.39)	7.66(1.66)	7.25(0.62)*
Elongate spermatids				
No.	129.24(17.37)			105.65(13.47)
ratio	7.78(1.54)	7.56(0.72)	7.09(1.62)	6.46(1.05)
Group 3 (Stage VII-VIII)				
No. of Sertali cells	19.28(1.92)	20.52(1.55)	19.20(1.58)	19.32(2.18)
Spermatogonia	• •	` ,	• •	
No.	4.52(1.32)	4.20(1.50)	4.92(1.63)	3.32(1.02)
ratio	0.23(0.05)	0.21(0.08)	0.26(0.11)	0.18(0.05)
Spermalocytes	= ()		( <b>-</b> )	-( <u>)</u>
No.	102.52(10.83)	99.08(8.42)	97.56(4.50)	89.04(9.00)
ratio	5.34(0.56))	4.85(0.50)	5.10(0.36)	4.62(0.32)
Elongate spermatids	(0.00))		2,22(0,2)	(uium)
No.	145.24(11.01)	130.64(9.90))	131.68(19.71)	119.24(15.90
ratio	7.56(0.61)	6.37(0.23)	6.88(1.04)	6.21(0.83)*

C 1 (C4 3/7) 3/73/7)		· · · · · · · · · · · · · · · · · · ·		
Group 4 (Stage VII-VIII)  No. of Sertoli cells	19.16(2.81)	20.92(1.73)	10 64(1 77)	1770/0.00
Spermatogonia	19.10(2.01)	20.92(1.73)	18.64(1.72)	16.72(0.92)
No.	4.04(0.89)	2 72/0 725	2 (4/0 30)	0 (1 (0 = 1)
	• •	3.72(0.72)	3.64(0.48)	3.64(0.71)
ratio	0.21(0.05)	0.18(0.03)	0.20(0.02)	0.22(0.05)
Spermatocytes				, ,
No.	109.80(13.15)	110.36(9.22)	99.44(4.54)	88.76(4.33)**
ratio	5.76 (0.29)	5.28(0.12)	5.36(0.34)	5.32(0.46)
Elongate spermatids	•	,	` ,	_(,
No.	159.76(15.91)	150.28(18.99)	137.08(17.70)	105.16(18.34)**
ratio	8.39(0.63)	7.19(0.71)	7.35(0.62)	6.33(1.31)**
Values are appropried as Mann(S	T) )	•	, ,	Ç

Values are expressed as Mean(S.D.)

Significantly different from 0 mg/kg group; \*  $p \le 0.05$ , \*\*  $p \le 0.01$ 

a): (No. of spermatogenic cells/no. of sertoli cells in a seminiferous tubule)

# Influence on reproductive performances of rats

	dose (mg/kg)						
Items	0	100	300	1,000			
No. of male animals examined	12	12	12	12			
No. of pairs with successful copulation	12	12	12	12			
Duration of mating (day, Mean, (SD))	2.1(1.2)	2.3(1.3)	2.7(1.2)	2.7(1.1)			
Copulation index(%)*	100.0	91.7	100.0	100.0			
No. of pregnant animals	11	10	12	12			
Fertility index(%)**	91.7	90.9	100.0	100.0			

<sup>\*(</sup>No.of pairs with successful copulation/no.of pairs mated) x 100

# Influence on developmental performances of rais

•		dose (n	rg/kg)	• *
Items	0	100	300	1,000
No. of male animals examined	12	12	12	12
No. of corpora lutea	16.8(1.5)	17.3(1.3)	17.0(2.3)	17.9(2.2)
No. of implantation sites	15.5(1.7)	16.6(1.3)	16.0(2.0)	16.3(2.3)
Implantation index(%) "	92.5(7.2)	96.2(6.6)	94.5(8.4)	91.3(8.8)
No. of pups born(%)	13.7(3.1)	15.0(1.7)	15.0(1.8)	15.1(2.7)
Delivery index(%) 4	87.6(15.4)	90.3(6.8)	94.1(7.2)	92.2(9.6)
Live pups born				
No.	13.3(2.9)	14.7(2.0)	14.9(2.0)	15.0(2.7)
Live birth index(%)	97.1(5.6)	97.8(3.6)	99.2(2.6)	99.4(2.1)
Sex ratio(M/F)	1.09(0.69)	1.05(0.50)	1.17(0.75)	0.76(0.44)
Dead pups born	-			
No.	0.5(0.9)	0.3(0.5)	0.1(0.3)	0.1(0.3)
Gestation length(day)	22.7(0.5)	22.7(0.5)	22.5(0.5)	11.6(0.5)
Gestation index(%)	100.0	100.0	100.0	100.0
Nursing index(%)	100.0	100.0	100.0	100.0
Live pups on day 4				
No.	13.2(2.8)	14.6(2.1)	14.4(2.9)	14.5(2.9)
Viability Index(%) 6	99.5(1.8)	99.3(2.3)	95.6(11.5)	96.7(6.7)
Body weight of pups(g)	•		,	•
Male Day 0	7.32(0.77)	7.13(0.52)	6.69(0.55)	6.87(0.84)
Day 4	11.71(1.76)	11.09(0.93)	10.23(0.98)*	10.60(1.47)
Day 0-4, gain(g)	4.39(1.04)	3.96(0.53)	3.54(0.77)*	3.73(0.80)
Body weight gain (%) "	59.41(8.87)	55.54(6.16)	53.19(11.91)	54.39(9.50)

<sup>\*\*(</sup>No. of pregnant animals/no. of pairs with successful copulation) x 100

Female	Day 0	6.93(0.83)	6.63(0.64)	6.33(0.58)	6.58(0.62)
	Day 4	11.08(1.71)	10.28(1.01)	9.84(1.01)*	10.03(1.46)
	Day 0-4, gain(g)	4.16(1.00)	3.65(0.56)	3.14(0.79)*	3.46(0.96)
	Body weight gain(%)	59.63(10.42)	55.24(8.07)	49.95(13.09)	52.17(11.10)

Values are expressed as Mean (S.D.)

Significantly difference from 0 mg/kg group; p ≤ 0.05

- a): (No. of implantation sites/no. of corpora lutea) x 100
- b): (No. of pups born/no. of implantation sites) x 100
- c): (No. of live pups born/no. of pups born) x 100
- d): (No. of females with live pups delivered/ no. of pregnant remales) x 100
- e): (No. of females nursing live pups/no. of females with normal delivery) x 100
- f): (No. of live pups on day 4/ no. of live pups born) x 100
- g): (Body weight gain/body weight on day 0) x 100

## CONCLUSIONS

## Repeat dose toxicity

Histopathological examination of the testes, demonstrated decrease of permatocytes and spermatids in males of the 300 and 1000 mg/kg group. No effects of this chemical on general appearance, body weight, food consumption, autopsy findings, weights of the reproductive organs of both sexes, or histopathlogical features of the ovary were detected.

The NOELs are considered to be 100 mg/kg/day for males, and 1,000 mg/kg/day for females.

# Reproductive and developmental toxicity

Except for the effects in males observed onhistopathological examination, no influence of this chemical was detected regarding reproductive ability, organ weight ohistopathological feature of the ovary, delivery or maternal behaviour of dams. No effects of this chemical were detected on iability, general appearance, body weights or autopsy findings for offspring.

The NOELs are considered to be 100 mg/kg/day for males, 1,000 mg/kg/day for females, and 1,000 mg/kg/day for offspring.

#### DATA QUALITY

- Reliabilities: Klimisch Code: 1=reliable without restrictions.
- · Remarks field for Data Reliability:

Well conducted study, carried out by the Safety Research Institute for Chemical Compounds Co., Ltd.(Japan)

#### REFERENCES

Toxicity Testing Reports of Environmental Chemicals, vol. 6(1998)

Ministry of Health & Welfare, Japan

## GENETIC TOXICITY IN VITRO (BACTERIAL TEST)

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

· Remarks:

Source: Daihachi Kagaku Kogyo Co., Ltd. Lot. No. N-60601

Purity: >99.0% Kept at room temperature in a dark place until use.

**METHOD** 

Method:

Guideline for ScreeningMutagenicity Testing of Chemicals (Japan) and

OECD TG 471 and 472

Test type:

Reverse mutation assay

• GLP:

Yes

Year:

1996

Species/Strain:

Salmonella typhimurium TA100, TA1535, TA98, TA1537

Escherichia coli WP2 uvrA

Positive controls:

-S9 mix, 2-(2-Furyl)-3-(5-nitro-2-furyl)acrylamide (TA100, WP2, TA98)

Sodium azide (TA1535)
9-Aminoacridine (TA 1537)

+S9 mix, 20Aminoanthracene (five strains)

S9:

Rat liver, induced with phenobarbital and 5,6-benzoflavone

Statistical methods

No statistical analysis was done.

# REMARKS FIELD FOR TEST CONDITIONS

Study Design:

Concentration: -S9: 0, 313, 625, 1,250, 2,500, 5,000 ug/plate (five strains) +S9: 0, 313, 625, 1,250, 2,500, 5,000 ug/plate (five strains)

Number of replicates: 2

Plates/test: 3

Procedure: Plate incorporation method

Solvent: Acetone Positive controls:

-S9 mix, 2-(2-Furyl)-3-(5-nitro-2-furyl)acrylamide (TA100, WP2, TA98)

Sodium azide (TA1535)
9-Aminoacridine (TA 1537)

+S9 mix, 20Aminoanthracene (five strains)

#### RESULTS

Cytotoxic concentration:

Toxicity was not observed up to 5,000 ug/plate in five strains with and without metabolic activation (S9 mix).

Genotoxic effects:
+ ? -
With metabolic activation: [ ] [ x ]
Without metabolic activation: [ ] [ x ]
REMARKS FIELD FOR RESULTS.
CONCLUSIONS
Bacterial gene mutation is negative with and without metabolic activation.
·
DATA QUALITY
Reliabilities: Valid without restriction.
Remarks field for Data Reliability
Well conducted study, carried out by Hatano Research Institute, Food and Drug Safety Center
(Hadano, Japan).
REFERENCES
Toxicity Testing Reports of Environmental Chemicals, vol.4(1996)
Ministry of Health & Welfare, Japan
•
GENERAL REMARKS
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REMARKS FIELD FOR RESULTS.

# GENETIC TOXICITY IN VITRO (NON-BACTERIAL IN VITRO TEST) TEST SUBSTANCE Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate Identity: Source: Daihachi Kagaku Kogyo Co., Ltd. Lot, No. N-60601 Remarks: Purity: >99.0% Kept at room temperature in a dark place until use METHOD Method: Guideline for Screening Toxicity Testing of Chemicals (Japan) Chromosomal aberration test Test type: GLP: Yes Year: 1996 Species/Strain: CHL/IU cell Metabolic activation: with and without S9 from rat liver, induced withphenobarbital and 5.6-benzoflavone. Statistical methods Fisher's exact analysis REMARKS FIELD FOR TEST CONDITIONS Study Design: For continuous treatment, cells were treated for 24 or 48 hrs without S9. For short-term treatment, cells were treated for 6 hrs with and without S9 and cultivated with fresh media for 18 hrs. Concentration: -S9 (continuous treatment): 0, 1.3, 2.5, 5.0 mg/mL -S9 (short-term treatment): 0, 1.3, 2.5, 5.0 mg/ml. +S9 (short-term treatment): 0, 1.3, 2.5, 5.0 mg/mL Plotes/test: 2 Solvent: Acetone Positive controls: Mitomycin C for continuous treatment Cyclophosphamide for short-term treatment RESULTS Cytotoxic concentration: Toxicity was not observed up to 5.0 mg/ml in continuous and short-term treatment with or without S9 mix. Genotoxic effects: Clastogenicity polyploidy [ ] [ ] [ x ] With metabolic activation: Without metabolic activation: [ ] [ ] [ x ]

# GENETIC TOXICITY IN VITRO (NON-BACTERIAL IN VITRO TEST) TEST SUBSTANCE Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate Identity: Remarks: Source: Daihachi Kagaku Kogyo Co., Ltd. Lot. No. N-60601 Purity: >99.0% Kept at room temperature in a dark place until use METHOD Method: Guideline for Screening Toxicity Testing of Chemicals (Japan) Chromosomal aberration test Test type: GLP: Yes 1996 Year: Species/Strain: CHL/IU cell Metabolic activation: with and without S9 from rat liver, induced withphenobarbital and 5.6-benzoflavone. Statistical methods Fisher's exact analysis REMARKS FIELD FOR TEST CONDITIONS Study Design: For continuous treatment, cells were treated for 24 or 48 hrs without S9. For short-term treatment, cells were treated for 6 hrs with and without S9 and cultivated with fresh media for 18 hrs. Concentration: -S9 (continuous treatment): 0, 1.3, 2.5, 5.0 mg/ml. -S9 (short-term treatment): 0, 1.3, 2.5, 5.0 mg/ml. +S9 (short-term treatment): 0, 1.3, 2.5, 5.0 mg/mL Plotes/test: 2 Solvent: Acetone Positive controls: Mitomycin C for continuous treatment Cyclophosphamide for short-term treatment RESULTS Cytotoxic concentration: Toxicity was not observed up to 5.0 mg/ml in continuous and short-term treatment with or without 59 mix. Genotoxic effects: Clastogenicity polyploidy

# REMARKS FIELD FOR RESULTS.

With metabolic activation: [ ] [ ] [ x ] Without metabolic activation: [ ] [ ] [ x ]

Appendix I
Parameters used in caluculation of distribution by Mackay level III fugacity model.

Physico-chemical Parameter for TOTM

		44
ar weight	546.79	Measured
oint [°C]	-50	Measured
essure [Pa]	2.80E-04	Estimated
bility [g/m²]	0.13	Measured
Kow	5.94	Measured
in air	12	Estimated
in water	288	Estimated
in soil	288	Estimated
in sediment	864	Estimated
	in water in soil	bility [g/m³] 0.13  Kow 5.94  in air 12  in water 288  in soil 288

[emp. [°C] 25

# Environmental Parameter

		Volume	depth	агеа	organic	lipid content	density	residence
		$[m^3]$	[m]	[m²]	carbon[-]	[-]	[kg/m³]	time {h}
	air	1.0E+13					1.2	100
bulk air	particles	2.0E+03						
	total	1.0E+13	1000	1E+10				
,	water	2.0E+10					1000	1000
bulk water	particles	1.0E+06			0.04		1500	
	Fish	2.0E+05				0.05	1000	
	Total	2.0E+10	10	2E+09				
	Air	3.2E+08			·		1.2	
bulk soil	Water	4.8E+08					1000	•
	Solid	8.0E+08			0.04		2400	
	Total	1.6E+09	0.2	8E+09				,
bulk	Water	8.0E+07					1000	
sediment	Solid	2.0E+07			0.06		2400	50000
	Total	1.0E+08	0.05	2E+09			•	

# Intermadea Transport Parameter (m/h)

air side air-water MTC	5	soil air boundary layer MTC	5
water side air water MTC	0.05	sediment-water MTC	1E-04
rain rate	1E-04	sediment deposition	5E-07
aerosol deposition	6E-10	sediment resuspension	2E-07
soil air phase diffusion MTC	0.02	soil water runoff	5E-05
soil water phase diffusion MTC	1E-05	soil solid runoff	1E-08

# Theoretical Distribution of TOTM

# scenario 1

	emission rate	conc.	amount	percent	Transformation rate [kg/h]				
	[kg/h]	[g/m³]	[kg]	[%]	Reaction	advection			
Air	1,000	1.3.E-07	1.3.E+04	19.6	7.5E+02	1.3.E+02			
Water	O	1.6.E-05	3.10.E+03	4.7	7.6E+00	3.1.E+00			
Soil	0	2.5.E-03	4.4,E+04	66.2	1.1E+02				
Sediment	•	1.3.E-02	6.3.E+03	9.5	5.1E+00	1.2.E-01			
		total amount	6.7.E+04						

# scenario 2

	Emission rate	conc.	Amount	percent	Transformation rate [kg/h]				
	[kg/h]	[g/m³]	[kg]	[%]	Reaction	advection			
air	0	1.8.E-09	1.8.E+02	0.0	1.0.E+01	1.8.E+00			
water	1000	9.7.E-04	1.9.E+05	32.7	4.7.E+02	1.9. <b>E</b> +02			
soil	0	3.4.E-05	6.2.E+02	0.1	1.5.E+00				
sediment		7.9.E-01	3.9.E+05	67.2	3.2.E+02	7.9.E+00			
		total amount	5.9.E+05						

# DRAFT ENV/JM/EXCH(99)13

# scenario 3

	emission rate	. 1	Amount	percent [%]	Transformation rate [kg/h]				
	[kg/h]	[g/m³]	[kg]		Reaction	advection			
аіг	0	7.0.E-13	7.0.E-02	0.0	4.1.E-03	7.0.E-04			
water	0	5.2.E-08	1.0.E+01	0.0	2.5.E-02	1.0.E-02			
soil	1000	2.3.E-02	4.2.E+05	100.0	1.0.E+03				
sediment		4.2.E-05	2.1.E+01	0.0	1.7.E-02	4.2.E-04			
	:	total amount	4.2.E+05						

scenario 4

	emission rate	conc.	Amount	percent	Transformati	on rate [kg/h	
	[kg/h]	[g/m³]	[kg]	[%]	Reaction	advection	
air	600	7.8.E-08	7.8.E+03	3.0	4.5.E+02	7.8.E+01	
water	300	3.0.E-04	6.0.E+04	23.5	1.5.E+02	6.0.E+01	
soil	100	3.8.E-03	6.8.E+04	26.6	1.6.E+02		
sediment		2.4.E+01	1.2.E+05	46.9	9.8.E+01	2.4.E+00	
		total amount	2.6.E+05				

# Summary of SIDS Information on Trimellitates A. Physical/Chemical Properties of Trimellitates

(R)					-		Water	Photodeg	Hydrolysis		Transpo	ort (%) c	
Carbon Chain Length	CAS Number	AS Chemical MP* BP** VP PC Solubility Half-		Half-life (days)	-life Half-life		Air	Water	Sediment				
C8	3319-31-1	tris 2-ethylhexyl (TOTM)	-46 97 c	>300 541 c	<0.0001*** 5.25E-11 c	5.94 11.59 c	3.9E-04 4.51E-08 c	0.33 с	0.05 0.32 c	97.8	3.6E - 6	2.8E - 7	2.17
C8	27251-75-8	triisooctyl ester	<0 197 c	541 c	5.25E-11 c	11.59 с	4.51E-08 c	0.35 с	0.43 с	97.8	3.64E - 6	2.8E - 7	2.17
C9	53894-23-8	triisononyl ester	<0. 224 c	>300 575 c	3.17E-12 c	13.06 c	1.32E-09 c	0.31 с	0.86 с	97.8	2.74E - 7	9.61E -9	2.17
C8,C10	67989-23-5	decyl, octyl ester	<0 234 c	585 c	1.37E-12 c	12.79 с	2.78E-09 c	0.32 c	0.98 c	97.8	1.02E - 7	1.79E - 8	2.17

c = calculated data using EPWIN; all other values are derived from measurements

<sup>\* =</sup> All of these trimellitates are liquids at zero degrees C. Modeled data do not accurately reflect melting points for these substances

<sup>\*\* =</sup> Measured boiling points were determined to be >300°C at 0.66 kPa

<sup>\*\*\* =</sup> vapor pressure of TOTM 13 Pa @ 200°C

# **Summary of SIDS Information on Trimellitates** B. Toxicology Data on Trimellitates

(R) Carbon Chain Length	CAS Number	Chemical Name	Acute Oral LD50	Acute Dermal LD50	Acute Inhalation LC50	Repeated Dose Toxicity	GeneTox (Ames)	GeneTox (Chrom. Abs.)	Toxicity to Reproduction	Developmental Toxicity / Teratogenicity	Acute Fish (A) mg/L	Daphnia (B) mg/L	Algal (C) mg/L	Biodegradation %
C8	3319-31-1	tris 2-ethylhexyl (TOTM)	> 3.2 g/kg (rat, mouse)	>20 ml/kg (guinea pig) >2.0 ml/kg (rabbit)	<2.64 mg/L (rat, nominal)	NOAEL (rat, dietary) 654 mg/kg/day	Negative	Negative (CHL/IU cells)	NOAEL (rat, oral) 1000 mg/kg/day	NOAEL (rat, oral) 1000 mg/kg/day (3)	>100	>180	>100	68-71 (1) 4.2 (2)
C8	27251-75-8	Triisooctyl ester												
C9	53894-23-8	Triisononyl ester	> 10 g/kg (rat)											
C8, C10	67989-23-5	decyl, octyl ester						·						

- Footnotes: A) Japanese Medaka (Oryzias latipes), 96 hr LC50 & NOEC
  - B) Daphnia magna, 48-hr EC50
  - C) Selenastrum capricornutum, 72-hr EC50 & NOEC
  - (1) Inherent biodegradation by Shake Flask Method
  - (2) Ready biodegradation by MITI method (OECD 301C)
  - (3) OECD Preliminary reproduction toxicity screening test; indirect measure of develomental effects

# SIDS INITIAL ASSESSMENT PROFILE

CAS NO.	3319-31-1
CHEMICAL NAME	Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate
Structural formula	CH <sub>2</sub> CH <sub>3</sub> COOCH <sub>2</sub> CHCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub> CH <sub>3</sub> CH <sub>2</sub> CH <sub>3</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub> CH <sub>2</sub> CH <sub>3</sub>

# RECOMMENDATION

The chemical is currently of low priority for further work.

### SUMMARY CONCLUSIONS OF THE SIAR

#### Human health

Acute toxicity of TOTM is low, LD<sub>50</sub> >2,000 mg/kg in rats. In the irritation-test for animals, this substance is slightly irritating to the skin and the eyes. Sensitization test on guinea pig showed no sensitization. Oral study in rats conducted for 28 days at doses of 0(0), 0.2(184), 0.67(650), 2.0(1826) % (mg/kg bw/day) of TOTM. There were no statistically significant differences in body weights between control and TOTM treated groups. There was a significant difference between control and treated groups in the following: hemoglobin concentration (lower in both sexes, 0.67 or 2.0% TOTM), leucocyte counts (higher in males at 0.67 or 2.0%), absolute and relative liver weights (higher in both sexes at all levels except 0 or 0.2%), serum albumin (higher in both sexes at 0.67 or 2.0%), serum cholesterol levels (higher in males at 0.67 or 2.0%), serum urea (higher in males at 2.0%), serum lipids (decreased in females at 0.2%). Liver biochemistry revealed statistically significant differences between treated and control groups indicated by palmitoyl CoA oxidation (increased in both sexes at 2.0% and males at all dose levels), and catalase activity (increased in males at 2.0%).

Preliminary reproductive toxicity screening test reveals moderate decrease of of permatocytes and spermatids in males at 100mg/kg/day. From these two test results, he NOAELs for repeated or all toxicity were considered to be 100 mg/kg/day for male rats. The NOAELs for reproductive/ developmental toxicity were considered to be 1,000 mg/kg/day for female rats and for offspring. TOTM was evaluated its genotoxicity by many assay systems. It was neithermutagenic in bacteria nor clastogenic in mammalian cells in vitro. All other in vitro and in vivo assays gave negative results. It is concluded that TOTM is not genotoxic in vitro and in vivo. The reported results of carcinogenecity was invalid.

Absorption and metabolism were studied for 14°C labeled TOTM and about 75% of the dose was excreted unchanged in the feces, 16% in the urine as metabolites and 1.9% was expired as 14°CO<sub>2</sub>.

#### Environment

The Mackay level III fugacity Model was employed to estimate the environmental distribution of TOTM in air, water, soil and sediment. If released to air, TOTM will exist solely in the particulate phase in the ambient atmosphere. If released to soil, TOTM is not expected to have mobility. If released into water, TOTM is expected to adsorb to suspended solids and sediment in water.

TOTM has to be considered as weakly toxic against aquatic organisms. The substance is not readily biodegradable. Measured BCF of this chemical E reported as less than 1 to 2.7 in carp for 6 weeks, which suggest that bioconcentration in aquatic organisms is much lower than the value estimated from logPow(=5.94). The toxicity data to aquatic plants (algae; Selenastrum capricornutum) was >100 mg/L for EC<sub>50</sub> (72hr) and NOEC (72hr). The acute toxicity data in fish (medaka, Oryzias latipes) were >100 mg/L (96h, LC<sub>50</sub> and NOEC) and >75 mg/L (14d, LC<sub>50</sub> and NOEC). In Daphnia magna, acute toxicity was >180mg/L (48hr: EC<sub>50</sub>) and chronic toxicity was 55.6mg/L (21d, reproduction NOEC). All these data were obtained in supersaturated solution with the aid ofsolubilizer (HCO-40). The test solution was considered to be homogeneous substantially. Another chronic toxicity data inDaphnia magna (NOEC >0.082mg/L) was reported. Though this value is lower than the saturation point, the observed concentration data was less reliable. Assessment factor of 100was chosen to determine the lowest PNEC. Thus, calculated PNEC (=0.00082 mg/L) of TOTM is closely to the value of one hundredths (assessment factor) of saturation point. From these toxicity data, it is difficult to decide the exact PNEC, but we are sure of the practical safety of TOTM against aquatic organisms.

#### Exposure

TOTM is manufactured as the plasticizer of PVC applications.

The production volume of TOTM in Japan is approximately 20,000 tonnes/year and also, there are 5 manufacturers in Japan. Estimated global production is 40,000-100,000 tonnes/year. This substance is produced in closed system and mainly used asplasticizer for PVC electrical cable and wire. And so, this substance has been already blended to the compound asplasticizer, so it is not expected that downstream users or consumers of electric wire industry may expose to this substance.

Occupational exposure may occur through dermal contact and inhalation of mist. The process is constructed by closed system and workers wear protective gloves and goggles during the operation, so significant exposure is not expected.

# NATURE OF FURTHER WORK RECOMMENDED

No recommendation

**FULL SIDS SUMMARY** 

CAS NO: 3319-31-1   SPECIES   PROTOCOL   RESULTS		IDS SUMMARY			
2.1   Meiting Point	CASN	O: 3319-31-1	SPECIES	PROTOCOL	RESULTS
2.2   Boiling Point   Diher (unknown)   Diher (unknown)   Doesity   Diher (unknown)   Doesity   Diher (unknown)   Doesity   Decoration   Doesity   Decoration	PHYSI	CAL-CHEMICAL			
2.2 Boiling Point 2.3 Density 2.4 Vapour Pressure 2.5 Pertition Coefficient (Log P <sub>om</sub> ) 2.6A. Water Solubility B. pH pKa 2.12 Oxidation: Reduction Potential  ENVIRONMENTAL FATE AND PATHWAY 3.1.1 Photodegradation 3.1.2 Stability in Water  3.2 Monitoring Data 3.3 Transport and Distribution  Calculated (Level III Fugacity Model)  Distribution  3.5 Biodegradation 3.7 Biosecumulation  ECOTOXICOLOGY 4.1 A Acute Toxicity to Fish Prolonged Toxicity to Fish  CECD TG 202  OECD TG 203  Other (unknown) ODER (107 OECD TG 107  CALC III CALC I	2.1	Melting Point		OECD TG 102	<-50 °C (223 K)
2.4   Vapour Pressure   Olther (unknown)   0.987-0.990 g/cm² at 20 °C	2.2				l , , , , , , , , , , , , , , , , , , ,
2.4   Vapour Pressure   Partition   Coefficient   Coeffi	2.3	_		•	\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \
2.5 Partition Coefficient (Log P <sub>em</sub> ) 2.6A. Water Solubility B. pH pXa 2.12 Oxidation: Reduction Potential ENVIRONMENTAL FATE AND PATHWAY 3.1.1 Photodegradation 3.1.2 Stability in Water  3.2 Monitoring Data Transport and Distribution  Calculated (Level III Fugacity Model)  Awater Soil Sediment 19.6% 4.7% 66.2% 9.5% (Release 100% to sir) Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil) Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil) Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil) Air Water Soil Sediment 0.0% 0.0% 100% 0.0% PECD TG 305C  ECOTOXICOLOGY  4.1 A Acute Toxicity to Fish 4.1 B Prolonged Toxicity to Fish  A courte Toxicity to Fish	2.4			•	1
Calculated   Cal	2.5	_		OECD TG 107	-
2.6A. Water Solubility B. pH pKa 2.12 Oxidation: Reduction Potential  ENVIRONMENTAL FATE AND PATHWAY  3.1.1 Photodegradation 3.1.2 Stability in Water  3.2 Monitoring Data 3.3 Transport and Distribution  Calculated (Level III Fugacity Model) Figacity Model  Distribution  Calculated (Level III Fugacity Model)  (Iocal exposure) Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to sair) Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil) Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil) Air Water Soil Sediment 0.0% 10.0% 0.0% 10.0% PCD TG 302C OECD TG 305C  ECOTOXICOLOGY  4.1 A Acute Toxicity to Fish Fish A.1 B Prolonged Toxicity to Fish  OECD TG 204  CECD TG 204  CECD TG 205  CECD TG 204  CECD TG 204  CECD TG 205  CECD TG 206  CECD TG 207  CECD TG 207  CECD TG 208  CECD TG 208  CECD TG 209  CECD TG 209  CECD TG 200  CECD TG 200  CECD TG 200  CECD TG 200  CECD TG 201  CECD TG 201  CECD TG 202  CECD TG 203  CECD TG 204  CECD TG 204  CECD TG 204  CECD TG 207  CECD TG 208  CECD TG 209  CECD TG 209  CECD TG 209  CECD TG 200  CECD		Coefficient			
2.6A. Water Solubility B. pH pKa 2.12 Oxidation: Reduction Potential  ENVIRONMENTAL FATE AND PATHWAY  3.1.1 Photodegradation 3.1.2 Stability in Water  3.2 Monitoring Data 3.3 Transport and Distribution  Calculated (Level III Fugacity Model) Figacity Model  Distribution  Calculated (Level III Fugacity Model)  (Iocal exposure) Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to sair) Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil) Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil) Air Water Soil Sediment 0.0% 10.0% 0.0% 10.0% PCD TG 302C OECD TG 305C  ECOTOXICOLOGY  4.1 A Acute Toxicity to Fish Fish A.1 B Prolonged Toxicity to Fish  OECD TG 204  CECD TG 204  CECD TG 205  CECD TG 204  CECD TG 204  CECD TG 205  CECD TG 206  CECD TG 207  CECD TG 207  CECD TG 208  CECD TG 208  CECD TG 209  CECD TG 209  CECD TG 200  CECD TG 200  CECD TG 200  CECD TG 200  CECD TG 201  CECD TG 201  CECD TG 202  CECD TG 203  CECD TG 204  CECD TG 204  CECD TG 204  CECD TG 207  CECD TG 208  CECD TG 209  CECD TG 209  CECD TG 209  CECD TG 200  CECD		(Log Park)	·		
B. pH pKs 2.12 Oxidation: Reduction Potential  ENVIRONMENTAL FATE AND PATHWAY  3.1.1 Photodegradation 3.1.2 Stability in Water  OECD TG 111 Stable at pH 4 at 50°C  T <sub>1/2</sub> =17.5 days at pH 7 at 25°C  T <sub>1/2</sub> =11.9 days at pH 9 at 25°C None  Calculated (Level III Fugacity Model)  Calculated (Level III Fugacity Model)  Calculated (Level III Fugacity Model)  For the stable at pH 4 at 50°C  T <sub>1/2</sub> =17.5 days at pH 7 at 25°C None  (Release 100% to air) Air Water Soil Sediment 0.0% 32.7% 0.1% 66.2% 9.5% (Release 100% to soil) Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil) Air Water Soil Sediment 0.0% 0.0% 100% 0.0%  PEC <sub>local</sub> = None  GECOTOXICOLOGY  4.1 A Acute Toxicity to Fish Fish  COECD TG 203  COECD TG 203  LC <sub>50</sub> (96 hr) > 100 mg/L NOEC(14 day) > 75 mg/L	2.6A.	• • •		OECD TG 105	0.13 mg/L at 25 °C
DKa   Oxidation: Reduction   Potential				!	
2.12		pKa	•		None
Potential	2.12				None
ENVIRONMENTAL FATE AND PATHWAY  3.1.1 Photodegradation 3.1.2 Stability in Water  3.2 Monitoring Data 3.3 Transport and Distribution  Calculated (Level III Fugacity Model)  Fugacity Model)  Distribution  Calculated (Level III Fugacity Model)  Fugacity Model)  Fugacity Model)  Calculated (Release 100% to air) Air Water Soil Sediment 19.6% 4.7% 66.2% 9.5% (Release 100% to water) Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil) Air Water Soil Sediment 0.0% 0.0% 100% 0.0%  Fugacity Model)  Fugacity Model		Reduction			
FATE AND PATHWAY  3.1.1 Photodegradation 3.1.2 Stability in Water  3.2 Monitoring Data 3.3 Transport and Distribution  Calculated (Level III Fugacity Model)  Fugacity Model)  Calculated (Level III Fugacity Model)  Fugacity Model)  Calculated (Level III Fugacity Model)  Air Water Soil Sediment 19.6% 4.7% 66.2% 9.5% (Release 100% to sail) Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil) Air Water Soil Sediment 0.0% 0.0% 100% 0.0%  Calculated (Level III Fugacity Model)  Fugacity Model)  Calculated (Level III Fugacity Model)  Fugacity Model)  Calculated (Level III Fugacity Model)  Calculated (Level III Fugacity Model)  Calculated (Release 100% to sail) Air Water Soil Sediment 0.0% 0.0% 0.0% 100% 0.0%  Calculated (Calculated		Potential			
3.1.1 Photodegradation 3.1.2 Stability in Water  3.1.2 Stability in Water  3.2 Monitoring Data 3.3 Transport and Distribution  Calculated (Level III Fugacity Model)  Calculated (Level III Fugacity Model)  Air Water Soil Sediment 19.6% 4.7% 66.2% 9.5% (Release 100% to sair) Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil) Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil) Air Water Soil Sediment 0.0% 0.0% 100% 0.0% PECD TG 302C 3.5 Biodegradation 3.7 Bioaccumulation  ECOTOXICOLOGY 4.1 A Acute Toxicity to Fish 4.1 B Prolonged Toxicity to Fish  OECD TG 203  CECD TG 203  LC <sub>50</sub> (96 hr) > 100 mg/L  NOEC(14 day) > 75 mg/L NOEC(14 day) > 75 mg/L NOEC(14 day) > 75 mg/L NOEC(14 day) > 75 mg/L NOEC(14 day) > 75 mg/L NOEC(14 day) > 75 mg/L NOEC(14 day) > 75 mg/L NOEC(14 day) > 75 mg/L NOEC(14 day) > 75 mg/L NOEC(14 day) > 75 mg/L	ENVIE	RONMENTAL			
3.1.2 Stability in Water  3.1.2 Stability in Water  3.2 Monitoring Data 3.3 Transport and Distribution  Calculated (Level III Fugacity Model)  Calculated (Release 100% to air)  Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil)  Air Water Soil Sediment 0.0% 0.0% 100% 0.0%	FATE	AND PATHWAY			
3.1.2 Stability in Water  3.1.2 Stability in Water  3.2 Monitoring Data 3.3 Transport and Distribution  Calculated (Level III Fugacity Model)  Calculated (Release 100% to air)  Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil)  Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil)  Calculated (Release 100% to air)  Calculated (Level III Fugacity Model)  Calculated (Release 100% to air)	3.1.1	Photodegradation			None
3.2 Monitoring Data 3.3 Transport and Distribution  Calculated (Level III Fugacity Model)  Calculated (Level III Fugacity Model)  Calculated (Level III Fugacity Model)  Calculated (Release 100% to air)  Air Water Soil Sediment 19.6% 4.7% 66.2% 9.5% (Release 100% to water)  Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil)  Air Water Soil Sediment 0.0% 0.0% 100% 0.0% 100% 0.0% PEC 1000 P	3.1.2	1 -		OECD TG 111	Stable at pH 4 at 50°C
3.2 Monitoring Data 3.3 Transport and Distribution  Calculated (Level III Fugacity Model)  Fugacity Model)  Fugacity Model)  19.6% 4.7% 66.2% 9.5% (Release 100% to water)  Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil)  Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil)  Air Water Soil Sediment 0.0% 0.0% 100% 0.0%				,	T <sub>1/2</sub> =17.5 days at pH 7 at 25°C
Calculated (Level III Fugacity Model)  Calculated (Level III Fugacity Model)  Calculated (Level III Fugacity Model)  Fugacity Model)  Calculated (Level III Fugacity Model)  Air Water Soil Sediment 19.6% 4.7% 66.2% 9.5% (Release 100% to water)  Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil)  Air Water Soil Sediment 0.0% 0.0% 100% 0.0		'			T <sub>1/2</sub> =11.9 days at pH 9 at 25°C
Distribution  (Level III Fugacity Model)  (Level III Fugacity Model)  (Release 100% to water)  Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil)  Air Water Soil Sediment 0.0% 0.0% 100% 0.0% 0.0% 0.0% 100% 0.0% 0.	3.2	Monitoring Data			
Fugacity Model)  19.6% 4.7% 66.2% 9.5% (Release 100% to water)  Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil)  Air Water Soil Sediment 0.0% 0.0% 100% 0.0% 0.0% 0.0% 0.0% 0.0%	3.3				, ,
(Release 100% to water)  Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil)  Air Water Soil Sediment 0.0% 0.0% 100% 0.0%  PEClocal = None 0ECD TG 302C 4.2 % after 28 days 0ECD TG 305C  ECOTOXICOLOGY  4.1 A Acute Toxicity to Fish Frolonged Toxicity to Fish  OFCD TG 203  CECD TG 204  OECD TG 205  OECD TG 206  OECD TG 207  OECD TG 207  OECD TG 208  OEC		Distribution		1 7	
Air Water Soil Sediment 0.0% 32.7% 0.1% 67.2% (Release 100% to soil) Air Water Soil Sediment 0.0% 0.0% 100% 0.0%  [local exposure] OECD TG 302C 3.7 Bioaccumulation  ECOTOXICOLOGY  4.1 A Acute Toxicity to Fish Frolonged Toxicity to Fish  OECD TG 203  LC <sub>50</sub> (96 hr) > 100 mg/L  LC <sub>50</sub> (96 hr) > 100 mg/L  CECD TG 204  CECD TG 205  CECD TG 206  CECD TG 207  CECD TG 207  CECD TG 208				Fugacity Model)	h .
0.0% 32.7% 0.1% 67.2%     (Release 100% to soil)   Air   Water   Soil   Sediment     0.0% 0.0% 100% 0.0%     0.0% 0.0% 100% 0.0%     PEC_{local} = None     4.2					
(Release 100% to soil)  Air Water Soil Sediment 0.0% 0.0% 100% 0.0%  PEClocal = None 4.2 % after 28 days  OECD TG 305C  OECD TG 305C  OECD TG 305C  OECD TG 305C  OECD TG 203  LC <sub>50</sub> (96 hr) > 100 mg/L  OECD TG 204  LC <sub>50</sub> (14 day) > 75 mg/L  NOEC(14 day) > 75 mg/L  OECD TG 202  OECD TG 202  OECD TG 203  CECD TG 204  OECD TG 204  OECD TG 204  OECD TG 207  CECD TG 207  OECD TG 208  OECD TG 208					l
Air Water Soil Sediment 0.0% 0.0% 100% 0.0%    Occidence of the process of the pr					1
(local exposure)  (local exposure)  (local exposure)  (local exposure)  (local exposure)  (PEC <sub>local</sub> = None  (A.2 Mafter 28 days  OECD TG 302C  OECD TG 305C  (DECD TG 305C  OECD TG 305C  OECD TG 203  (DECD TG 203  (DECD TG 203  (DECD TG 204  (DECD TG 204  (DECD TG 207  (DECD TG 207  (DECD TG 207  (DECD TG 208  (DECD TG 20					'
Comparison   Com				1	I .
3.5 Biodegradation 3.7 Bioaccumulation  CECD TG 302C  4.2 % after 28 days  BCF=1-2.7(Conc. 0.2 mg/L)  ECOTOXICOLOGY  4.1 A Acute Toxicity to Fish  Fish  Latipes  Oryzias Latipes  Oryzias latipes  Oryzias latipes  OECD TG 203  LC <sub>50</sub> (96 hr) > 100 mg/L  NOEC(14 day) > 75 mg/L  LOEC(14 day) > 75 mg/L  LOEC(14 day) > 75 mg/L  COECD TG 202  COECD TG 203  COECD TG 204  COECD TG 204  COECD TG 204  COECD TG 205  COECD TG 206  COECD TG 207  COECD TG 207  COECD TG 208				(lamal arrangement)	
3.7 Bioaccumulation  OECD TG 305C  BCF=1-2.7(Conc. 0.2 mg/L)  ECOTOXICOLOGY  4.1 A Acute Toxicity to Fish  OFCD TG 203  OFCD TG 203  LC <sub>50</sub> (96 hr) > 100 mg/L  Latipes  OFCD TG 204  CFCD TG 204  CFCD TG 207  CFCD TG 208  DECD TG 208  CFCD		W1 - 1 1 - 1	ļ		A 2 % often 20 dans
ECOTOXICOLOGY  4.1 A Acute Toxicity to Fish  4.1 B Prolonged Toxicity to Fish  Comparison of the first Toxicity to Fish  Acute Toxicity to Fish  Comparison of the first Toxicity Toxicity Toxicity Toxicity Toxicity Toxicity Toxicity Toxicity	13	, =		L .	
4.1 A Acute Toxicity to Fish  4.1 B Prolonged Toxicity to Fish  Acute Toxicity to Fish  OFCD TG 203  LC <sub>50</sub> (96 hr) > 100 mg/L  Latipes Oryzias latipes  OECD TG 203  LC <sub>50</sub> (14 day) > 75 mg/L  NOEC(14 day) > 75 mg/L  LOEC(14 day) > 75 mg/L  COECD TG 202  FC (24 hr) > 180 mg/L				CECD 16 303C	BCF=1-2.7(Conc. 0.2 mg/L)
Fish Prolonged Toxicity to Fish  Actives  Oryzias  Latipes Oryzias latipes  OECD TG 204  LC <sub>50</sub> (14 day) > 75 mg/L  NOEC(14 day) > 75 mg/L  LOEC(14 day) > 75 mg/L  OECD TG 202  FC., (24 br) > 180 mg/L			`		
4.1 B Prolonged Toxicity to Fish  Latipes Oryzias latipes OECD TG 204 LC <sub>50</sub> (14 day) > 75 mg/L NOEC(14 day) > 75 mg/L LOEC(14 day) > 75 mg/L CFCD TG 202  Fig. (24 br) > 180 mg/L	4.1 A		Oryzias	OECD TG 203	LC <sub>50</sub> (96 hr) > 100 mg/L
Toxicity to Fish  Oryzias latipes  OECD 16 204  LC <sub>50</sub> (14 day) > 75 mg/L  NOEC(14 day) > 75 mg/L  LOEC(14 day) > 75 mg/L  OECD 16 202  FC:: (24 hr) > 180 mg/L		]		OPOD TO SO!	T.C. (14 down) 5 25 2
Toxicity to Pish  LOEC(14 day) > 75 mg/L  LOEC(14 day) > 75 mg/L  A 2 Agusta Toxicity to	4.1 B	. –		OECD 16 204	
4.2 Agus Tayleitu to OFCD TG 202 FC (24 hr) > 180 mg/l		Toxicity to Fish	'		
4.2 Acute Toxicity to   Description of the CD 1G 202   EC50 (24 m) > 180 mg/L		4 4	!	<b>∆E¢n</b> #€ 600	
II I IIIIII I III III III III III III	4.2	-	Daphnia magna	OECD 1G 202	
Aquatic EC <sub>50</sub> (40 m) > 100 mg/L	ł				
Invertebrates NOEC > 180 mg/L		ļ -			
(Daphnia) LOEC > 180 mg/L				Aren me es:	- I
4.3 Toxicity to Selenastrum OECD TG 201 EC <sub>50</sub> (72 hr) > 100mg/L	4.3	7		OECD TG 201	
Aquatic Plants   Capricornutum   NOEC(72 hr) > 100mg/L		1			NOEC(72 hr) > 100mg/L
e.g. Algae ATCC22662			ATCC22662		
4.5.1 Chronic Toxicity None	4.5.1	Chronic Toxicity	1		None
to Fish	4	1 "	l .	1	

4.5.2	Chronic Toxicity to Aquatc Invertebrates (Daphnia)	Daphnia magna	OECD TG 211	NOEC(21d, reproduction) = 55.6 mg LOEC(21d, reproduction) > 100 mg/L EC <sub>50</sub> (21d, reproduction) > 89.1 mg/L LC <sub>50</sub> for parental <i>Daphnia</i> (21d) > 100 mg/L NOEC=0.0082 (21d, reproduction, parent <i>Daphnia</i> mortality)
4.6.1	Toxicity to Soil Dwelling Organisms			None
4.6.2	Toxicity to Terrestrial Plants		;	Νοπε
4.6.3	Toxicity to Other Non- Mammalian Terrestrial Species			<b>Р</b> Опе
	(Including Birds)			
TOXIO	COLOGY			
5.1.1	Acute Oral Toxicity	Rat	OECD TG 401	LD <sub>50</sub> > 2,000 mg/kg (for both sexes)
5.1.2	Acute Inhalation Toxicity	Rat	Other	2,600 mg/m³ (4hr)
5.1.3	Acute Dermal Toxicity	Rabbit	Other	$LD_0 > 2.0 \text{ mL/kg}$
5.2.1	Skin Irritation	Rabbit	Other	Slightly irritating
5.2.2	Eye Irritation	Rabbit	Other	Slightly irritating
5.3	Skin Sensitisation	Guinea pig	OECD TG 406	Not sensitizing
5.4	Repeated Dose Toxicity	Rat	OECD TG 421	NOAEL = 100 mg/kg bw LOAEL = 300 mg/kg bw
5.5	Genetic Toxicity In Vitro			
A.	Bacterial Test	S.typhimurium, E. coli	Japanese Guideline and OECD TG 471 & 472	- (With metabolic activation) - (Without metabolic activation)
В.	Non-Bacterial In Vitro Test	CHL/IU cells	Japanese Guideline	- (With metabolic activation) - (Without metabolic activation)
5.6	Genetic Toxicity In Vivo	Mouse	Other	No valid data
5.7	Carcinogenicity	Mouse	Other	No valid data
5.8	Toxicity to	Rat	OECD TG 421	NOAEL = 100 mg/L (male)
	Reproduction		Preliminary toxicity screening test	NOAEL = 1,000 mg/L (female) NOAEL = 1,000 mg/L (Offspring)
5.9	Developmental			None
1	Toxicity/			
	Teratogenicity			
5.11	Experience with			None
	Human Exposure			

[Note] Data beyond SIDS requirements can be added if the items are relevant to the assessment of the chemical, e.g. corrosiveness/irritation, carcinogenicity.

# SIDS Initial Assessment Report for 13th SIAM

(November 6-9, 2001)

Chemical Name: Tris(2-ethylhexyl) benzene-1,2,4-tricarboxylate

CAS No:

3319-31-1

Sponsor Country: Japan

National SIDS Contact Point in Sponsor Country: Mr. Koji Tomita, Ministry of Foreign Affairs, Japan

### HISTORY:

The original IUCLID documents were prepared by European Commission. Dainippon Ink and Chemicals Inc., Japan reviewed the documents after incorporation of Japanese testing results.

# COMMENTS:

ICCA Initiative work led by Dainippon Ink and Chemicals Inc., Japan

Deadline for circulation:

Date of Circulation:

# SIDS INITIAL ASSESSMENT REPORT (SIAR)

Tris(2-ethylhexyl) benzene-1,2,4-tricarboxylate

1. IDENTITY

IUPAC Name:

Tris(2-ethylhexyl) benzene-1,2,4-tricarboxylate

CAS Number:

3319-31-1

Molecular formula:

 $C_{33}H_{54}O_6$  (MW=546.79)

Structural formula:

$$\begin{array}{c} \text{CH}_2\text{CH}_2\\\\ \text{COOCH}_2\text{CHCH}_2\text{CH}_2\text{CH}_2\text{CH}_2\\\\ \text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\\\\ \text{CH}_2\text{CH}_3\\\\ \text{CH}_2\text{CH}_3\\\\ \end{array}$$

Synonym:

TOTM

Tris(2-ethylhexyl) trimcllitate

Benzene-1, 2, 4-tricarboxylic acid tris-(2-ethylhexyl) ester

Purity:

>99.5%

Impurity:

Di(2-ethylhexyl) phthalate (DEHP) < 0.1%

Water

Additives:

None

Table 1. Physical and Chemical Properties

Items	Protocol	Results
Melting Point	OECD TG 102	< -50°C
Boiling Point	Unknown	283°C (4 hPa)
Density	Unknown	0.987 - 0.990 g/cm <sup>3</sup> (20°C)
Vapor pressure	OECD TG 104	<2.8 x 10 <sup>-1</sup> Pa(100°C)
Partition Coefficient (LogP)	OECD TG 107	5.94 (25°C)
Water Solubility	OECD TG 105	0.13 mg/ L (25°C)

#### 2. GENERAL INFORMATION ON EXPOSURE

The production volume of TOTM in Japan is approximately 20,000 tonnes /year and also, there are 5 manufacturers in Japan. Estimated global production is 40,000 – 100,000 tonnes/year. TOTM is produced in closed system andmainly used as plasticizer for PVC electrical cable and wire especially for high temperature application. TOTM is no source of potential release to the environment except for samplingand maintenance of the production acidities.

#### 2.1 Environmental Fate

Based upon the biodegradation measurement, the substance is not readily biodegradable. TOTM achieved 4.2 percent of its theoretical BOD using an activated studge inoculum during a 4 weeks incubation in a single screening study.

The Mackay levelll fugacity model was employed to estimate the environmental distribution of TOTM in air, water, soil and sediment. The calculation results are shown in Table 2.If released to air, an estimated vapor pressure of less than 2.8 x 10<sup>-4</sup> Pa at 100°C indicates TOTM will exist solely in the particulate-phase in the ambient amosphere. Particulate-phase TOTM is removed from the atmosphere by wet and dry deposition. If released to soil, TOTM isnot expected to have mobility based upon the fugacity model calculation Volatilization from soil surfaces is not expected to be an important environmental fate process based on the stimated vapor pressure of this substance. If released into water, TOTM is expected to adsorb to suspended solids and sediment in water based upon the fugacity model calculation. [Dainippon Ink and Chemicals, Inc. (2001)]

Hydrolysis may be an important environmental fate process based orestimated hydrolysis half-lives of 17.5 and 11.9 days at pH 7 and 9, respectively. Measured BCF values of less than 1 to 2.7 in carp suggest that bioconcentration in aquatic organisms is low.

Table 2. Predicted distribution of TOTM using Fugacity level III (%)

Compartment	Release 100% w air	Release 100% to water	Release 100% to soil
Air	19.6	0.0	0.0
Water	4.7	32.7	0.0
Soil	66.2	0.1	100.0
Sediment	9.5	67.2	0.0

# 2.2 Human Exposure

# 2.2.1 Occupational exposure

The substance is produced and used in closed system. So, occupational exposure is limited in the case of sampling and maintenance at the production facilities. Moreover, the exposure time is very short. The major rout of occupational exposure is inhalation and ermal.

The atmospheric concentration was measured at two production sites in Japan. The monitoring data are shown in Table 3. The maximum exposure level is estimated according to working schedules as follows. From Table 3, if a single worker (Body weight; 70 kg, respiratory volume; 1.25 m³/hour) is assigned to implement all daily operation without protection, the daily intake (EHE inh) is calculated as 1.77 x 10³ mg/kg/day as the worst case. On the other hand, a single worker (surface area of exposed skin 840 cm² for hands) daily dermal dose (EHE der) is calculated as 2.47 mg/kg/day based on below calculation as the worst case without protection. Workers wear protective gloves and goggles during the operation, so significant exposure is not expected.

Table 3. Available workplace monitoring data for TOTM (EHE inh)

Occupation	Frequency Times/day	Duration Hr	Working hr/day	Max concentration mg/m³	EHE inh mg/kg/day	Reference
Sampling	5	0.017	0.085	0.210	3.19x10 <sup>-4</sup>	JISHA,
Analysis	5	0.067	0.335	0.053	3.17x10 <sup>-4</sup>	Japan
Charge to drum	1	0.833	0.833	0.076	$1.13 \times 10^{-3}$	(2001)
Total	11	-	1.253	-	1.77×10 <sup>-3</sup>	

EHE inh: Estimated Human Exposure for inhalation

Calculation: EHE der = (Cder \* T \* S \* t ) /W

EHE der. Estimated Human Exposure for dermal

Cder = 990 mg/cm<sup>3</sup> (Rate in product contacted by worker)

T = 0.01 cm (Thickness of substance)

 $S = 840 \text{ cm}^2$  (Surface area of exposed skin) for hand

t = 0.0208 day/day (Exposure time per day; 10min/8Hr, [1day = 8Hr] assumed)

W = 70 Kg (body weight)

# 2.2.2 Consumer exposure

Usually, this substance has been already blended to the compound asplasticizer, so it is not expected that downstream users or consumers of electric wire industry may expose to this substance.

#### 3.HUMAN HEALTH HAZARDS

#### 3.1 Effects on Human Health

#### 3.1.1 Toxicokinetics and metabolism

Absorption and metabolism were studied for TOTM(14C-labeled on the 2-carbon atom of 2-ethylhexyl group) mixed with corn oil and administered by gavage in a single dose of 100 mg/kg of body weight in 4 male SDrats. About 75% of the dose was exercted unchanged in the feces, 16% in the urine as metabolites and 1.9% was expired as <sup>14</sup>CO<sub>2</sub>. Radioactivity was excreted in the feces as unchanged TOTM (85% of the fecal radioactivity), mono- and di(2-ethylhexyl) trimellitate(MOTM and DOTM, respectively), and as unidentified polar metabolites. Metabolites in the urine were identified as MOTM and metabolites of 2-ethylhexanol less than 0.6% of the dose remained in the tissues. Elimination of CO<sub>2</sub> was biphasic with half-lives of 4.3 and 31 hrs, and excretion of radioactivity in the urinewas biphasic with half-lives of 3.4 hrs and 42 hrs. [Eastman Kodak Company]

# 3.1.2 Acute toxicity

Acute toxicity data are mainly reported for rat, mice and rabbits. We could find 12 acute toxicity data for animals (oral(6), inhalation(1), IP(2) and dermal(3)) test data, and one (oral) study (MHW, Japan (1996)) and two (oral and dermal) studies Nuodex Inc. (1981), Nuodex Inc. (1982c) ) were conducted by the method of OECD TG and similar method to OECD TG, respectively.

The data, which we feel informative to evaluate the acute toxicity, are listed in Table 4.

Table 4. Summary of effects of TOTM on animals (Acute Toxicity)

Route	Animals	Values	Type	References
Oral	Rat	>2000 mg/kg	$LD_{\mathbf{x}}$	MHW, Japan (1996)
	Rat	>5000 mg/kg bw	$LD_0$	Nuodex Inc. (1981)
Inhalation	Rat	>2600 mg/m <sup>3</sup>	LC <sub>0</sub>	Nuodex Inc.(1982b)
Dermal	Rabbit Rabbit	>2 ml/kg >1970 mg/kg bw	LD <sub>o</sub> LD <sub>o</sub>	Nuodex Inc(1982c) Tenneco Chemicals(1981))
I.P.	Rat Mouse	>3200 mg/kg bw >3200 mg/kg bw	LD <sub>50</sub>	Eastman Kodak (1983) Eastman Kodak (1983)

It can be concluded that acute toxicity (Oral) of TOTM is LD<sub>20</sub>>2000 mg/kg in rat.

# 3.1.3 Repeated dose toxicity

Among the eight available data, four were conducted under the GLP. Three studies were considered to be key study.

The first study was the oral study by CMA(1985). The subchronic toxicity of TOTM administered orally in the diet togroups of 5 male and 5 femaleFischer 344 rats at levels of 0(0), 0.2(184), 0.67(650), 2.0(1826) % (mg/kg bw/day) for 28 days was determined. There were no statistically significant differences in body weights between control and OTM treated groups. There was a significant difference between control and treated groups in theollowing: absolute and relative liver weights (higher in both sexes at all levels except 0 or 0.2%) serum albumin (higher in both sexes at 0.67 or 2.0%), serum cholesterol levels (higher in males at 0.67 or 2.0%). Liver biochemistry revealed statistically significant differences between treated and control groups as indicated by palmitoyl CoA oxidation (increased in both sexes at 20% and males at all dose levels), and catalase activity (increased inmales at 2.0%). So, the NOAEL for repeated dose toxicity is considered to be 184 mg/kg and the LOAELis 650 mg/kg for both sexes.

The second study was the oral study by MHWJapan(1996). No test substances related changes were noted in terms of clinical signs, body weight, food consumption, and hematology, blood examination, urinalysis, and pathological findings. So, the NOEL for repeat dose toxicity is considered to be 1,000 mg/kg/day for both sexes.

The third study was the OECD preliminary reproduction toxicity screening test by MHW lapan(1998). Gavage study in SD rats conducted at doses of 100, 300 and 1,000 mg/kg/day (Male; 46days, Female; from 14days before mating to day 3 of lactation) of TOTM. The decreases in spermatocytes and spermatids in males was observed for 300 and 1,000 mg/kg groups by histopathological examination. No effects on general appearance, body weight, food consumption, autopsy findings, weights of the reproductive organs of both sexes, or histopathlogical features of the ovary were detected. So, the NAOEL is considered to be 100 mg/kg/day for males, and 1,000 mg/kg/day for females.

There is no available information on human toxicity.

#### Conclusions:

The NOAEL and the LOAEL for repeated oral toxicity are considered to be 100 and 300 mg/kg/day for rats, respectively.

# 3.1.4 Genotoxicity / Mutagenicity

We can find five reports for Ames Tests. One (MHW, Japan: 1996) is conducted under GLP and others are not. The study of MHW is considered to be a key study.

TOTM has been investigated in vitro tests. This substance did not induce gene mutation in bacterial system (MHW, Japan: 1996), and chromosomal aberration in mammalian cultured cells (MHW, Japan: 1996), with and without an exogenous metabolic activation system. Among these studies, MHW study was identified to be a key study because it was well conducted and reported.

Reverse gene mutation assay was conducted by OECD TG 471 and 472, using pre-incubation method. TOTM was not mutagenic in Salmonella typhimurium TA100, TA1535, TA98, TA1537 and Escherichia coli WP2 uvrA at concentration of up to 5000 ug /plate, with or without an exogenous metabolic activation system (MHW, Japanit 996).

Chromosomal aberration test by OECD TG 473 was conducted in cultured Chinese hamsterlung (CHL/IU) cells. Structural chromosomal aberrations and polyploidy were not inducedup to a maximum concentration of 5.0 mg/mL on continuous treatment, and with Short-term treatment, with and without an exogenous metabolic activation system (MHW, Japan: 1996).

And all other test results (HGPRT assay, Unscheduled DNA synthesis, Dominant Lethal Assay for example) shows that TOTM is not genetoxic

### Conclusions:

This substance is considered to be not genotoxic with and without an exogenous metabolic activation system in bacterial test and chromosomal aberration testin vitro.

# 3.1.5 Carcinogenecity

One brief report states only that tests in mice, with a propensity to form pulmonary adenomas, were negative for TOTM, unlike those using urethane. The carcinogeneous tests revealed that the chemical is negative but test result was invalid.

# 3.1.6 Reproduction/developmental toxicity

The OECD Preliminary Reproduction Toxicity Screening Test was performed. [MHW, Japan: 1998]. This study was identified to be well conducted and eported.

Gavage study in SD rats conducted at doses of 100,300 and 1,000 mg/kg/day (Male; 46days, Female; from 14days before mating to day 3 of lactation) of TOTM.

Histopathological examination of the testes revealed decreases impermatocytes and spermatids in males of the 300 and 1,000 mg/kg groups. No effects of TOTM were detected on general appearance, body weight, food consumption, autopsy findings, and weight of reproductive organs of both sexes, or on histopathological examination of the ovary. On the basis of these findings, the NOELs of TOTM for repeat dose texicity are considered to be 100 mg/kg/day for males, and 1,000 mg/kg/day for females.

Except for the effects in males observed on histopathological examination, no influence of this substance was detected regarding reproductive ability, organ weights or histopathological appearance of the ovaries, delivery or maternal behavior of dams. No effect of TOTM were detected on viability, general appearance, body weight or autopsy findings of offspring. On the basis of these findings, the NOELs for reproductive developmental toxicity were considered to be 100mg/kg/day for male rats, 1,000 mg/kg/day for female rats, and 1,000 mg/kg/day for offspring.

### Conclusions:

The NOELs for reproductive/developmental toxicity were considered to be 100 mg/kg/day for male rats, 1,000 mg/kg/day for female rats, and 1,000 mg/kg/day for offspring, respectively.

#### 3.1.8 Other: Irritation and sensitization

Six and three results are reported for skin and eye irritation test, respectively. All these test results showed that TOTM is slightly irritating to the skin and the eye.

Sensitization test on guinea pig using OECD/TG 406 (Tenneco Chemicals, 1981) showed "no sensitization".

#### 3.2 Initial Assessment for Human Health

Acute toxicity of TOTM is considered to be LD<sub>0</sub> >2000 mg/kg in rat.

In the irritation-test for animals, TOTM is slightly irritating to the skin and the eye.

Sensitization test on guinea pig using OECD/TG 406 showed "no sensitization".

The NOAEL and the LOAEL for repeated oral toxicity are considered to be 100 and 300 mg/kg/day for rats, respectively

The NOELs for reproductive/developmental toxicity were considered to be 100 mg/kg/day for male rats, 1,000 mg/kg/day for female rats, and 1,000 mg/kg/day for offspring, respectively.

This substance is not genotoxic with and without an exogenous metabolic activation system in bacterial test and chromosomal aberration testin vitro.

TOTM produces the same spectrum of morphological and biochemical change in the rat liver as DEHP. TOTM, however, was much less potent in its action, with a dictary level of 2.0%, causing less peroxisome proliferation and peroxisome-associated enzyme induction than 0.67% DEHP. Also, the level of peroxisome induction in rats given TOTM is less than in those receiving a metabolically equivalent dose of 2-ethylhexanol. Furthermore, on a molar basis, effects were lower than with DEHP. An effect of MEHP, a metabolite of DEHP, was not seen with TOTM. [The British Industrial biological Research Association (1985), EPA OTS0510637(1985), JOHN R. HODGSON. (1987)]

In addition, recently studies have determined that rodents (rats) are susceptible toperoxisome proliferation. After all, these results suggest that the effect of DEHP on liver are markedly different between other species (marmosets) and rodents (rats). Yoshimasa Kurata et al. (1998)] Therefore, DEHP was downgraded from Group 2B to Group 3 by the IARC Monographs Working Group. (February 2000) Group 3 is "cannot be classified as to its carcinogenicity to humans".

#### 4. Hazards to the Environment

### 4.1 Aquatic Effects

TOTM has to be considered as weakly toxic against aquatic organisms. Aquatic effects were tested and results are summarized in Table 5. As the lowest acute and chronic toxicity data, EC<sub>50</sub> (>100 mg/L, 72hr) of Selenastrum capricornutum ATCC22662 and NOEC (55.6 mg/L, 21day) of Daphnia magna were adopted, respectively.

Table 5. Summary of effects of TOTM on aquatic organisms

Organism	Test duration	Result (mg/L)	Reference
Algae			,
Selenastrum	72 hr	EC <sub>50</sub> >100	EA, Japan
capricornutum		NOEC>100	
ATCC22662			
Invertebrates			
Daphnia magna	<b>24 h</b> r	EC <sub>50</sub> >180	EA, Japan
-	48 hr	EC <sub>50</sub> >180	
		NOEC>180	
	<b>48</b> hr	EC <sub>50</sub> >1	ICI 1990
	21 day	EC <sub>50</sub> ≈89.1	EA, Japan
	•	NOEC=55.6	

	21 day	NOEC=0.082	CMA (1985)
Fish Oryzias latipes	96 hr	LC <sub>so</sub> >100	EA, Japan
	14 day	LC <sub>∞</sub> >75 NOEC>75	EA, Japan

As the acute toxicity data,  $BC_{\infty}$  (>100 mg/L,72hr) of Selenastrum capricornutum ATCC22662 and  $EC_{50}$  (180 mg/L, 48hr) of Daphnia magna were adopted, respectively. As the chronic toxicity data of Daphnia magna and the prolonged toxicity data of fish Oryzias latipes), NOEC =0.082mg/L (21days) [CMA; 1985] and NOEC=75mg/L (14days) [EA Japan] were adopted, respectively. All those data in supersaturated solution, which was considered to be homogeneous substantially, was obtained with the aid of solubilizer (HCO-40). Though the observed concentration data was less reliable, one chronic toxicity data (NOEC >0.082mg/L) was reported in a lower concentration than saturation point.

Two other acute (ICI 1990) and chronic(EA Japan) data would be helpful for evaluation of the toxicity for Daphnia magna. These tests were conducted in a supersaturated solution.

Assessment factor of 100 was chosen to determine the lowest PNEC. Thus, calculated PNEC (=0.00082 mg/L) of TOTM is closely to the value of one hundredths (assessment factor) of saturation point. From these toxicity data, it is difficult to decide the exact PNEC, but we are sure that TOTM is practically non-toxic against aquatic organisms.

4.2 Terrestrial effects

There is no available information.

#### 4.3 Initial assessment for the Environment

Hydrolysis may be an important environmental fate process based orestimated hydrolysis half-lives of 17.5 and 11.9 days at pH 7 and 9, respectively. The substance is not readily biodegradable. Measured BCF values of this chemical is reported as less than 1 to 2.7 in carp for 6 weeks, which suggest that bioconcentration in aquatic organisms is much lower than the value estimated from logPow(=5.94). If released into surface water water, TOTM is expected to adsorb to suspended solids and sediment based upon the fugacity model calculation. The sediment toxicity data was not available, and will need to assess when obtained.

#### 5. Conclusions and recommendations

#### 5.1 Conclusions

#### Exposure (Physical/chemical property, production, use and distribution)

TOTM is manufactured as the plasticizer of PVC application.

The production volume of TOTM in Japan is approximately 20,000 tonnes /year and also, there are 5 manufacturers in Japan. Estimated global production is 40,000 – 100,000 tonnes/year. TOTM is produced in closed system andmainly used as plasticizer for PVC electrical cable and wire. And so, this substance has been already blended to the compound aplasticizer, so it is not expected that downstream users or consumers of electric wire industry may expose to this substance.

Occupational exposure may occur through dermal contact and inhalation of vapor. The process

is constructed by closed system and workers wear protective gloves and gogles during the operation, so significant exposure is not expected.

In case of disposal, this substance would be incinerated with following all regulations. Therefore, it is not significant released to theen vironment

#### Human health

Acute toxicity of TOTM is low, LD<sub>50</sub> >2,000 mg/kg in rats. In the irritation-test for animals, this substance is slightly irritating to the skin and the eyes. Sensitization test on guinea pig showed "no sensitization". Oral study in rats conducted for 28 days at doses of 0(0), 0.2(184), 0.67(650), 2.0(1826) % (mg/kg bw/day) of TOTM. There were no statistically significant differences in body weights between control and TOTM treated groups. There was a significant difference between control and treated groups in thefollowing: hemoglobin concentration (lower in both sexes, 0.67 or 2.0% TOTM), leucocyte counts (higher in males at 0.67 or 2.0%), absolute and relative liver weights (higher in both sexes at all levels except 0 or 0.2%), serum albumin (higher in both sexes at 0.67 or 20%), serum cholesterol levels (higher in males at 0.67 or 2.0%), serum urea (higher in males at 2.0%), serum lipids (decreased in females a0.2%). Liver biochemistry revealed statistically significant differences between treated and control groupsas indicated by palmitoy) CoA oxidation (increased in both sexes at 20% and males at all dose levels), and catalase activity (increased in males at 2.0%). Therefore, the NOAEL and the LOAEL for repeated oral toxicity were considered to be 100 and 300 mg/kg/day for male rats. The NOELs for reproductive/developmental toxicity were considered to be 1,000 mg/kg/day for female rats and for offspring.

TOTM is not genetoxic/mutagenic in bacterial and mammalian cell tests in vitro tests. The carcinogenecity tests revealed that the chemical is negative but test result was invalid.

### Environment

The Mackay levelll fugacity model was employed to estimate the environmental distribution of TOTM in air, water, soil and sediment. If released to air, TOTM will exist solely in the particulate-phase in the ambient atmosphere. If released to soil, TOTM is not expected to have mobility. If released into water, TOTM is expected to adsorb to suspended solids and sediment in water,

Measured BCF of values of less than 1 to 2.7 in carp suggest that bioconcentration in aquatic organisms is low.

As the lowest acute and chronic toxicity data, EC<sub>50</sub> (>100 mg/L, 72hr) of Seletiastrum capricornutum ATCC22662 and NOEC (0.082mg/L, 21day) of Daphnia magna were adopted, respectively. Assessment factor of 100 was chosen to both acute and chronic toxicity data to determine PNEC Thus, PNEC of TOTM is 0.00082mg/L.

#### 5.2 Recommendations

The chemical is currently of low priority for further work.

#### 6. References

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Yoshimasa Kurata. Subchronic Toxicity of Di(2-ethylhexyl)phthalate in Common Marmosets: Lack of Hepatic Peroxisome Proliferation, Testicular Atrophy, or Pancreatic Acinar Cell Hyperplasia. Toxicological Sciences 42, 49-56 (1998) PROPOSED ROBUST SUMMARY for Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate CAS No. 3319-31-1

Sponsor Country: Japan

D' aug 24, 2001

# PHYSICAL/CHEMICAL ELEMENTS

#### MELTING POINT

### TEST SUBSTANCE

· Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

· : Remarks:

Source: Tokyo Kasei Kogyo Co., Ltd. Lot No. AX01

Purity: 98.5%

**METHOD** 

Method/guideline:

OECD TG 102

• GLP:

Yes

· Year:

1998

Remarks:

Not stated.

RESULTS

Melting point value:

<-50 °C (223 K)

Decomposition:

Not stated.

Sublimation:

Not stated.

Remarks:

Not stated,

CONCLUSIONS

Melting point is <-50°C (223 K).

# DATA QUALITY

· Reliabilities:

Key study

Remarks:

Well conducted study, carried out by Chemicals Evaluation and

Research Institute (Kurume, Japan).

# REFERENCES

Ministry of International Trade and Industry (1998)

- Last changed:
- · Order number for sorting
- Remarks:

# BOILING POINT (a)

# TEST SUBSTANCE

· Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Unavailable.

### METHOD

Method:

Not specified.

• GLP:

Not stated.

· Year:

Not stated.

Remarks:

Not stated.

# RESULTS

Boiling point value:

283°C

• Pressure:

4

· Pressure unit:

hPa

• Decomposition:

Not stated,

Remarks:

Not stated.

# CONCLUSIONS

Boiling point is 283°C at 4 hPa.

# DATA QUALITY

Reliabilities:

Key study

Remarks:

Not stated.

#### REFERENCES

Midwest Research Institute; Thomas W. Lapp, Charles EMumma Joseph Chaszar: A Survey of Plasticizers: Epoxies, Linear Polyesters and Trimellitates Chemical Technology and Economics in Environmental Perspective, Task IV, Environmental Protection Agency (Nov. 1981)

- Last changed:
- · Order number for sorting
- Remarks:

# **BOILING POINT (b)**

#### TEST SUBSTANCE

· Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

· Remarks:

Source: Unavailable.

# METHOD

Method:

Not specified.

• GLP:

Not stated.

Year:

Not stated.

· Remarks:

Not stated.

### RESULTS

Boiling point value:

414°C (687K)

• Pressure:

1,013

· Pressure unit:

hPa

• Decomposition:

Not stated.

Remarks:

Not stated.

# CONCLUSIONS

Boiling point is 414°C at 1,013hPa.

# DATA QUALITY

Reliabilities:

Key study

· Remarks:

The Sigma-Aldrich Library of Regulatory and Safety Data.

# REFERENCES

Ministry of International Trade and Industry (1998)

- Last changed:
- Order number for sorting
- Remarks:

# DENSITY

### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

· Remarks:

Source: Unavailable.

METHOD

Method:

Not specified.

• GLP:

Not stated.

· Year:

Not stated.

Remarks:

Not stated.

RESULTS

Density:

 $0.987 - 0.990 \text{ g/cm}^3$ 

Temperature

20°C

Remarks:

Not stated.

CONCLUSIONS

Density is 0.987-0.990 g/cm<sup>3</sup> at 20°C.

# DATA QUALITY

Reliabilities:

Key study

Remarks:

Not stated.

#### REFERENCES

Midwest Research Institute; Thomas W. Lapp, Charles EMumma Joseph Chaszar: A Survey of Plasticizers: Epoxies, Linear Polyesters and Trimellitates Chemical Technology and Economics in Environmental Perspective, Task IV, Environmental Protection Agency (Nov. 1981)

- Last changed:
- Order number for sorting
- Remarks:

# VAPOR PRESSURE (a)

#### TEST SUBSTANCE

· Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

· Remarks:

Source: Tokyo Kasei Kogyo Co., Ltd. Lot No. AX01

Purity: 98.5%

**METHOD** 

Method/guideline:

Yes

OECD TG 104

GLP:

Year:

1998

Remarks:

Not stated.

RESULTS

Vapour Pressure value:

 $< 2.8 \times 10^{-4} \text{ Pa}$ 

• Temperature:

100°C

• Decomposition:

Not stated.

Remarks:

Not stated.

CONCLUSIONS

Vapour pressure is  $< 2.8 \times 10^{-4}$  Pa at 100 °C.

# DATA QUALITY

Reliabilities:

Key study

Remarks:

Well conducted study, carried out by Chemicals Evaluation and

Research Institute (Kurume, Japan).

#### REFERENCES

Ministry of International Trade and Industry (1998)

- Last changed:
- Order number for sorting
- · Remarks:

# VAPOR PRESSURE (b)

### TEST SUBSTANCE

· Identity:

Tris(2-ethylbexyl)benzene-1,2,4-tricarboxylate

· Remarks:

Source: Unavailable.

#### **METHOD**

Method/guideline:

Not stated

· GLP:

Not stated

· Year:

Not stated

· Remarks:

Not stated.

# RESULTS

Vapour Pressure value:

0.27 - 6.7 hPa

Temperature:

250 - 260 °C

· Decomposition:

Not stated.

Remarks:

Not stated.

### CONCLUSIONS

Vapour pressure is 0.27- 6.7 hPa at 250 - 260 °C.

# DATA QUALITY

Reliabilities:

Key study

Remarks:

Not stated.

# REFERENCES

Midwest Research Institute; Thomas W. Lapp, Charles EMumma Joseph Chaszar: A Survey of Plasticizers: Epoxies, Linear Polyesters and Trimellitates Chemical Technology and Economics in Environmental Perspective, Task IV, Environmental Protection Agency (Nov. 1981)

- Last changed:
- Order number for sorting
- Remarks:

### PARTITION COEFFICIENT

# TEST SUBSTANCE

Identity:

Tris(2-ethylnexyl)benzene 1,2,4-tricarboxylate

Remarks:

Source: Tokyo Kasei Kogyo Co., Ltd. Lot No. AX01

Purity: 98.5%

METHOD

\* Method/guideline:

OECD TG 107 (Shake Flask Method, 1995)

• GLP:

Yes

· Year:

1998

Remarks:

Not stated.

RESULTS

· Log P. :

5.94

· Temperature:

25°C =1°C

· Remarks:

Test condition: Test was conducted in duplicate under the following

three conditions. Test chemical was analyzed y HPLC.

Test condition	Condition-1	Condition-2	Condition-3
1-Octanol saturated with water	10 mL	20 mL	40 tnL
Water saturated with 1-octanol	240 mL	230 mL	210 mL
Test chemical in 1-octanol saturated	with water (	(52.2 mg)	
•	10 ml.	Jm Of	10 mL

Test results	Log	Pow	
· -	*	b	Mean
Condition-1	5.99	5.99	
Condition-2	5.95	5.87	5.94
Condition-1	5 0 2	₹ 03	

CONCLUSIONS

log P is 5.94.

# DATA QUALITY

Reliabilities:

Key study

· Remarks:

Well conducted study, carried out by Chemicals Evaluation and

Research Institute (Kurume, Japan).

# REFERENCES

Ministry of International Trade and Industry (1998)

- Last changed:
- Order number for sorting

Remarks:		

# WATER SOLUBILITY

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Tokyo Kasai Kogyo Co., Ltd. Lot No. AX01

Purity: 98.5%

METHOD

Method:

OECD TG 105 (flask method).

GLP:

Yes

Year:

1998.

Remarks:

Not stated.

RESULTS

Value:

0.13 mg/L at 25 °C±1°C

Description of solubility;

Of very low solubility

• pH value;

No dissociation group.

pKa value;

There is no pertinent functional group.

Remarks:

Not stated.

### CONCLUSIONS

This chemical is very low solubility in water.

### DATA QUALITY

· Reliabilities:

Key study

Remarks:

Well conducted study, carried out by Chemicals Evaluation and

Research Institute (Kurume, Japan).

### REFERENCES

Ministry of International Trade and Industry (1998)

- Last changed:
- Order number for sorting
- · Remarks:

# ENVIRONMENTAL FATE AND PATHWAYS ELEMENTS

#### STABILITY IN WATER

### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Tokyo Kasei Kogyo Co., Ltd. Lot No. AX01

Purity: 98.5%

#### METHOD

Method/guideline:

OECD TG 111

Type:

Hydrolysis as a function of pH

• GLP:

Yes

Year:

1998

Remarks:

No hydrolysis of test chemical was observed at pH 4 at 50°Q-1°C for 5 days. Hydrolysis rates at pH 7 were determined at 60, 70 and 80 °C, and at pH 9 at 50, 60, and 70°C. They were extrapolated to 25 °C using Arrhenius relationship. Half life at 25 °C was calculated from the rate

constant.

#### RESULTS

Nominal:

ca. 0.2 mg/L

Measured value:

Not stated.

Degradation:

No hydrolysis occurred in 5 days, at 50 °C pH 4. At pH 7 and pH 9.

test chemicals were hydrolysed at all temperatures studied.

Half-life (t<sub>n/2</sub>):

Rate Constant (hr-1)

Half-life(day)

pH7 pH9 1.65 x 10° 2.44 x 10° 17.5 11.9

Breakdown products:

Not stated.

Remarks:

Not stated.

#### CONCLUSIONS

This chemical is stable in aqueous water at pH 4 under the condition studied, but it is hydrolysed at pH 7 and pH 9 at 25 °C with half-life of 17.5 and 11.9 days.

### DATA QUALITY

Reliabilities:

Key study

Remarks:

Well conducted study, carried out by Chemicals Evaluation and

Research Institute (Kurume, Japan).

### REFERENCES

Ministry of International Trade and Industry (1998)

# DRAFT ENV/JM/EXCH(99)13

- Last changed:
- Order number for sorting
- Remarks:

# TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS (FUGACITY)

#### TEST SUBSTANCE

Identity:

Tris(2-cthylhexyl)benzene-1,2,4-triearboxylate

Remarks:

Source: Not applicable.

### METHOD

Test:

Calculation

Method:

Fugacity level III

· Year:

2001

· Remarks

The parameters used are shown in Appendix.

#### RESULTS

Media :

\* Estimated Distribution under three emissioneenarios:

Compartment	Release100 % to air	Release 100 % to water	Release 100 % to soil
Air	19.6 %	0.0 %	0.0%
Water	4.7 %	32.7 %	0.0 %
Soil	66.2 %	0.1%	100.0 %
Sediment	9.5 %	67.2 %	0.0 %

#### Remarks

### CONCLUSIONS

If this chemical is released into water the majority of this chemical is expected to stay in sediment, but if it is released into air or soil, this chemical is expected to stay in soil

### DATA QUALITY

· Reliabilities:

Key sludy.

Remarks:

Not stated.

### REFERENCES

Dainippon lak and Chemicals, Incorporated (2001), unpublished report.

- Last changed:
- Order number for sorting
- Remarks:

### BIODEGRADATION

# TEST SUBSTANCE

Identity:

Tris(2-ethylliexyl)benzene-1,2.4-tricarboxylate

Remarks:

Source: Unavailable

#### METHOD

Method:

OECD TG 302C "Inherent Biodegradability: Modified MITITest(II)"

Test Type:

Acrobic

GLP:

No

Year:

1977

Contact time:

28 days

Inaculum:

The supernatant (500mi) of activated sewage sludge obtained fronten sampling sites and Sliters of supernatant removed from a previously established culture are transferred to a culture vessel. The pH of the culture mixture was adjusted to 7.0±1.0 and constantly acrated. Thirty minutes after stopping acration, discard about 1/3 of the whole volume of the supernatant, and add an equal volume of 0.1% synthetic sewage.

and the acration re-started. Repeat this procedure once a day.

Remarks:

During the aeration, appearance of supernatant and the formation of activated sewage was observed. The sludge was found to form a clear supernatant on settling and formed cloudyflocs when on aeration. Operating temperature, pH and a dissolved oxygen concentration were recorded. The protozoa of sludge were observed under an optical microscope.

\*Incubation apparatus: Respirometry(Closed bottle) Ohkura Electric Co.

\*CO2 absorbent: Soda lime No.1 (Wako pure chemicals Inc.)

\*Stirrer: Magnetic stirrer \*Temperature : 25±1°C

Concentration of test chemical: 30mg/L, 100mg/L.

\*Reference substance: Aniline

#### RESULTS

Degradation:

Results:

4.2% after 28days

Kinetic:

The percentage degradation in term of oxygen consumption was

calculated as follows:

% degradation = (BQD-B)/IQD x 100

BOD: Biological Oxygen Demand of the test material

: Oxygen consumption in basal culture medium to which inoculum is added (control)

TOD: Theoretical oxygen demand to completely oxidize the test

Material

Breakdown products:

Not stated.

· Remarks:

At the end of incubation, measure the residual dissolved organicarbon and test material concentration. The reference substance, aniline attained more than 40% and 60% degradation after 7 and 14days confirming the suitability of the inoculum and culture conditions.

### CONCLUSIONS

This chemical islow biodegradable.

# DATA QUALITY

· Reliabilities:

Key study

· Remarks:

Well conducted study, carried out by Chemical Inspection and Testing

Institute.

#### REFERENCES

Biodegradation and Bioaccumulation Data of Existing Chemicals Based on the CSCL Japan (1992)

Ministry of International Trade and Industry

- Last changed:
- · Order number for sorting
- · Remarks:

#### BIOACCUMULATION

#### TEST SUBSTANCE

· ldentity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Unavailable

METHOD

· Method:

OECD TG 305C

Species:

Cyprinus Carpio (Obtained from Nakajima hatchery in Kumamoto,

Japan)

• GLP:

No

· Year:

1978

Exposure Period:

42 days

Remarks:

Test fish:

Acclimated for ca. 8 weeksbefore testing at 25±2°C. Fish with ca.10cm

in length and ca.30g in weight were selected at random. Lipid content

was 2-6%.

Test condition

Concentrations: 0.2 and 2 mg/L, solubilizer controlled

Type of test: flow-through (200-800mL/min), 100L glass tank.

Dissolved oxygen concentration: 6-8mg/L

Temperature: 25 ±2°C

Water chemistry was tested in the control and two concentrations every

2 times in a week.

Test was conducted in duplicate every 2 weeks for two concentrations.

(The control was done before and after testing.)

RESULTS

Results:

BCF=1-2.7 (concentration: 0.2mg/L)

BCF=0.1-0.23(concentration: 2mg/L)

· Kinetic:

BCF=C1/C2

C1: Concentration of this chemical in Fish C2: Concentration of this chemical in water

Breakdown products:

Not stated.

CONCLUSIONS

This chemical is low bioaccumulation.

DATA QUALITY

· Reliabilities:

Key study

Remarks:

Well conducted study, carried out by Chemical Inspection and

Testing Institute

# REFERENCES

Biodegradation and Bioaccumulation Data of Existing Chemicals Based on the CSCI, Japan(1992)

Ministry of International Trade and Industry

- · Last changed:
- · Order number for sorting
- · Remarks:

# ECOTOXICITY ELEMENTS

#### ACUTE TOXICITY TO FISH

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Tokyo Kasei Kogyo Co., Ltd. Lot No. AX01

Purity: >95.0%

#### METHOD

Method:

OECD TG 203

Type:

Semi-static

GLP:

Yes

Year:

1998.

Species/Strain/Supplier: Oryzias latipes (Medaka): Obtained from commercial domestic

hatcheries.

Analytical monitoring

Yes. Test solutions were measured by IIPLC before and after 24 hours exposure period. Test solutions were replaced every 24 hours to new ones.

Exposure period (h):

Statistical methods:

Not applicable because of no mortality.

Remarks:

Test fish:

Acclimated for more than 12 days before testing; any groups showing no mortality for 7 days before test started. Fish with 22.1 mm (18.3-23.8 mm) in length were selected at random. Average body weight of fishers. 0.1462g (n=10).

Test conditions

Details of test: Semi-static (water changedevery 24 hours)

Dilution water source: Tap water after dechlorinated by passing through

activated carbon.

Dilution water chemistry: Hardness: 25 mg/L as CaCO<sub>3</sub>; pH: 6.7 Stock and test solution and how they are prepared: Pipette or pour the appropriate amount of the solution (0.3 wt% of test chemicalwith solubilizer hydrogenated caster oil HCO-40 3000mg/D into the test

waters.

Concentrations dosing rate, flow-through rate, in what medium: Concentrations of 0, 100 mg/L and dispersant control were tested. Vehicle/solvent and concentrations: Hydrogenated caster oil HCO-40,

100mg/L.

Stability of the test chemical solutions: Stable, measured concentration was 101-103%.

Exposure vessel type: 10 fish per group in 3L glass beaker without acration under room light

Number of replicates, fish per replicate: One replicatewas done. Water chemistry in test (O 2, pH) in the control and all concentration where effects were observed: Dissolved oxygen readings and pH values

were taken daily during 96 h exposure period.

Dissolved oxygen concentration: 5.0-9.2 mg/L.

pH values: 6.7-6.8.

Test temperature range: Water temperature at 23.5-24.1°C.

Method of calculating mean measured concentrations: Geometric mean.

### RESULTS

Nominal concentrations: 0, 100 (mg/L)

Measured concentrations: <1, 103 (0hr), <1, 102 (24hr)

Unit:

mg/L.

Element value

LC<sub>50</sub> at 96 hours >100.0 mg/L based on nominal concentrations.

Statistical results as appropriate: Not applied.

Remarks field for Results:

Biological observations

Not described.

Table showing cumulative mortality:

Percent mortality of Oryzias latipes exposed to the test chemical

Nominal concentration (mg/L)	Camulative number of dead fish (% mortality								
	24 hour	48 hour	72 hour	96 hour					
Control	O(0)	0(0)	0(U)	1(10)					
Dispersant Control	0(0)	0(0)	(0)0	0(D)					
100	0(0)	1(10)	1(10)	1(10)					

Lowest test substance concentration causing 100% mortality:

Not obtained under the test conditions studied.

Mortality of controls:

I fish was dead at 96h.

Abnormal responses:

At 24 hr, one fish showed abnormal breathing behaviour at 100mg/L. Copper(II)sulfate pentahydrate. LC<sub>n</sub> at 96h was 0.43 mg/L.

Reference substances:

Any observations, such as precipitation that might cause a difference between measuredne nominal

values:

It became clouded in 100mg/L concentration, but not precipitation.

# CONCLUSIONS

LC50 (96h) > 100 mg/L for fish.

### DATA QUALITY

Reliabilities:

Klimisch Code: 1=reliable without restrictions.

Remarks field for Data Reliability:

Experimental design and analytical procedure were well documented.

Carried out by Toray Research Center (Japan).

# REFERENCES

Environment Agency of Japan (1998).

- Last changed:
- Order number for sorting:

Remarks field for GeneralRemarks:

### PROLONGED TOXICITY TO FISH

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-trica/boxylate

Remarks:

Source: Tokyo Kasei Kogyo Co., Ltd. Lot No. AX01

Purity: >95.0%

#### **METHOD**

Method:

OECD TG 204

Type:

Flow-through.

• GLP:
• Year:

Yes. 1998.

Species/Strain/Supplier:

Oryzias latipes (Medaka): Obtained from commercial domestic hatcheries,

Analytical monitoring:

Yes. Test solutions were measured by HPLC before and after 7, 14days

exposure period.

Exposure period:

14 day.

Statistical methods

Binomial method (TOXDAT MULTI-METHOD PROGRAM, USEPA)

Dunnet method was used for LGo and for fish body weight difference,

respectively.

Remarks field for Test Conditions:

Test fish:

Acclimated for more than 12 days before testing; any groups showing 9% mortality for 7 days before test started. Fish with 20.0 mm (18.5-21.6 mm) in length were selected at random. Average body weight of fish was 0.484g (0.1182-0.2014g)(n=10). Fish were starved for 24 hours before the test

started.

Test conditions

Details of test: Flow-through.

Dilution water source: Tap water after dechlorinated by passing through

activated carbon.

Dilution water chemistry: Hardness: 15.3mg/L as CaCO<sub>3</sub>; pH: 7.0

Stock and test solution and how they are prepared: The working solution (4.8wt% of test chemical with solibitizer HCO-40 controlled was prepared with the dilution water. The test solution was supplied continuously by mixing the working solution and the dilution water with the help of a

mechanically operated quantitative water-pump.

Concentrations dosing rate, flow-through rate, in what medium: Nominal concentrations of 0, 18.8, 37.5 and 75.0 mg/L and Dispersant control were tested.

Vehicle/solvent and concentrations Hydrogenated caster oil HCO-40,

Max. 75.0 mg/L

Stability of the test chemical solutions: It became clouded in high

concentration, but not precipitation.

Exposure vessel type: 10 fish per group in 3L glass beaker without aeration

under room light

Number of replicates, fish per replicate: One replicate was done

Water chemistry in test (O2 pH) in the control and one concentration where

effects were observed: Dissolved oxygen readings and pH values were

taken every 3 days during the exposure period. Dissolved oxygen concentration: 6.6-7.7 mg/L.

pH values: 6.9~7.2.

Test temperature range

Water temperature at 23.5-24.1 °C (24±2°C).

Method of calculating mean measured:Geometric mean.

#### RESULTS

- Nominal concentrations: 0, 18.8, 37.5, 75.0 (mg/L) and dispersant control
- Measured concentrations:

Measured concentration of the test chemical during a 14-day exposure of orange killifish (Oryzias latipes) under flow-through test conditions

Nominal concentration (mg/L)	Measured concentration (mg/L) (percent of nominal)							
	Oday	7 day	14 day	Mean				
Control	<1.0	< 1.0	< 1.0	<b></b>				
Dispersant Control	<1.0	< 1.0	< 1.0					
18.8	17.7(94.1)	15.8(84.0)	15.5(82.4)	16.3(86.9)				
37.5	35.7(95.2)	33.2(88.5)	30.0(80.0)	33.3(87.9)				
75.0	70.6(94.1)	68.8(91.7)	71.2(94.9)	70.2(93.6)				
mg/L			. ,	•				

Unit:Element value:

LC<sub>so</sub> (7 days) > 75.0mg/L (nominal concentration)

LC<sub>so</sub> (14 days) > 75.0mg/L (nominal concentration)

NOEC (14 days) > 75.0 mg/L (nominal concentration)

Statistical results, as appropriate:

The mean body weight of fish exposed toall concentration of the test chemical was not significantly different from controls during the test periodalfa=0.05. Dunnet).

Remarks field for Results:

Biological observations: Not described.

Cumulative mortality:

Percent mortality of Oryzias latipes exposed to the test chemical under flow-through test Conditions

Nominal conc. (mg/L)		Cumulative number of dead fish (% mortality)						y)			(days)				
•	Ģ	1	2	3	4	5	ó	7	8	9	10	11 -	12	13	14
Control	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	1(10)	1(10)	1(10)
Disp. Cont.	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	O(D)	0(0)	0(0)	0(0)	0(0)	O(O)	0(0)
18.8	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)
37.5	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)
75.0	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)	0(0)

Fish weight:

Numinal conc. (mg/L)		Fish v	reight (	g)							
	Na.1	No.2	No. 3	No.4	No.5	Nu.6	No.7	No.8	No.9	No.10	Ave.
Control	0.1879	0.2526	0.1273	0.2239	0.1139	0.1434	0.1708	0.1789	0.1558	-a	0.1727
Disp. Cont.	0.2205	0.1827	0.1192	0.1884	0.1438	0.1823	0.1563	0.2120	0.1635	0.1580	0.1727
18.8	0.1731	0.1513	0.1593	0.1472	0.2150	0.1548	0.1547	0.1306	0.2104	0.1020	0.1598
37.5	0.1264	0.1495	0.1872	0.1237	0.2055	0.1396	0.1805	0.2101	0.1577	0.1303	0.1611
75.0	0.1746	0.1848	0.1804	0.1625	0.1494	0.1633	0.2103	0.1454	0.1600	Q.1818	0.1713
•	_ (	· No me	seitrement	was mad	te hecaus	e the Ora	nga Killif	ich was di	ead.		

Lowest test substance concentration causing 100% mortality>75.0 mg/mL (nominal).

Mortality of controls:10 % mortality observed during the test period (12 through 14 days).

Food intake:

Fish was fed with TetraMin<sup>®</sup> fish food (2% of fish body weight).

Abnormal responses: No abnormal response showed through 14 days.

Reference substances (if used)- results: Copper (II) sulfate pentahydrate. LC<sub>50</sub> at 96h was 0.30 mg/L.

Any observations, such as precipitation that might cause a difference between measured and nominal values: It became clouded high concentration, butnot precipitation.

#### CONCLUSIONS

LC<sub>50</sub> (7 days) > 75.0 mg/L (nominal concentration) LC<sub>50</sub> (14 days) > 75.0 mg/L (nominal concentration) NOEC (14 days) > 75.0 mg/L (nominal concentration)

# DATA QUALITY

· Reliabilities:

Klimisch Code: 1=reliable without restrictions.

Remarks field for Data Reliability:

Experimental design and analytical procedure were well documented. Carried out by Toray Research Center (Japan).

#### REFERENCES

Environment Agency of Japan (1998).

#### OTHER

- Last changed:
- · Order number forsorting:
- Remarks: GeneralRemarks:

# ACUTE TOXICITY TO AQUATIC INVERTEBRATES (e.g., Daphnia)

# TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Tokyo Kasei Kogyo Co., Ltd. Lot No. AX01

Purity: >95.0%

METHOD

Method:

**OECD TG 202** 

• Type:

Static

• GLP:

Yes

Year:

1998

Species/Strain/Supplier:

Daphnia magna

Analytical monitoring

Yes. Test solutions were measured by HPLC before and after 48 hours

exposure period.

Exposure period (h):

48

Statistical methods:

Not applicable.

### Remarks field for Test Conditions:

Test organisms:

Source, supplier, any pre-treatment, breeding method: Supplied by NIES

(Japan).

Age at study initiation: Juveniles within 24h old.

Control group: Yes.

Test conditions

Stock solutions preparation and stability: No solvent used. Test chemical

was diluted to 1800mg/L (with solubilizer HCO-40 1000mg/L controlled)

with diluting water (Elendt M4) before use.

Test temperature range:

19.9-20.2 °C (average temperature 20°C).

Exposure vessel type: 100mL test solution in a 100 mL glass beaker; 4

beakers per treatment

Dilution water source: Elendt M4(OECD guideline No.211 Annex 2)

Dilution water chemistry: Hardness: 228mg/L as CaCO<sub>3</sub> Lighting: room light 16h:8h light-darkness cycle Water chemistry in test: DO= 8.0-8.6mg/L; pH=7.3-7.8.

Feeding: none

Test design:

Number of replicates=20

Concentrations: 0, 17.1, 30.9, 55.6, 100 and 180 mg/L, because 48h-EiC<sub>50</sub> for parent Daphnia (Acute immobilization test) was>1000mg/L. Dispersant

control was also tested.

Method of calculating mean measured concentrations. Geometric mean.

Exposure period:

48 h

Analytical monitoring:

By HPLC analysis, 95.1-99.6% of the nominal concentration at

preparation; 90.1-97.7% after 48hr.

# ACUTE TOXICITY TO AQUATIC INVERTEBRATES (e.g., Daphnia)

#### TEST SUBSTANCE

· Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Tokyo Kasei Kogyo Co., Ltd. Lot No. AX01

Purity: >95.0%

METHOD

Method:

OECD TG 202

Type:

Static

• GLP:

Yes

Year:

1998

Species/Strain/Supplier:

Daphnia magna

Analytical monitoring

Yes. Test solutions were measured by HPLC before and after 48 hours

exposure period.

Exposure period (h):

48

Statistical methods:

Not applicable.

Remarks field for Test Conditions:

Test organisms:

Source, supplier, any pre-treatment, breeding method: Supplied by NIES

(Japan).

Age at study initiation: Juveniles within 24h old.

Control group: Yes.

Test conditions

Stock solutions preparation and stability: No solvent used. Test chemical

was diluted to 1800mg/L (with solubilizer HCO-40 1000mg/L controlled)

with diluting water (Elendt M4) before use.

Test temperature range:

19.9-20.2 °C (average temperature 20°C).

Exposure vessel type: 100mL test solution in a 100 mL glass beaker; 4

beakers per treatment

Dilution water source: Elendt M4(OECD guideline No.211 Annex 2)

Dilution water chemistry: Hardness: 228mg/L as CaCO<sub>3</sub> Lighting: room light 16h:8h light-darkness cycle Water chemistry in test: DO= 8.0-8.6mg/L; pH=7.3-7.8.

Feeding: none

Test design:

Number of replicates=20

Concentrations: 0, 17.1, 30.9, 55.6, 100 and 180 mg/L, because 48h-EiC<sub>so</sub> for parent Daphnia (Acute immobilization test) was>1000mg/L. Dispersant

control was also tested.

Method of calculating mean measured concentrations. Geometric mean.

Exposure period:

48 h

Analytical monitoring:

By HPLC analysis, 95.1-99.6% of the nominal concentration at

preparation; 90.1-97.7% after 48hr.

#### RESULTS

Nominal concentrations: 17.1, 30.9, 55.6, 100.0, 180.0 (mg/L) (Solubilizer controlled)

Measured concentrations:

Measure Concentrations of test chemicals during a 48hr.

Nominal Concentration	Measur	ed concent	ration(mg/L)	Percent of nominal		
(mg/L)	0hr	48hr	Mean	Ohr	48hr	
Control	< 1.0	< 1.0	-	-	_	
Disp.Cont.	< 1.0	< 1.0	-	-	-	
17.1	16.3	15.4	15.8	95.3	90.1	
30.9	29.4	28.5	28.9	95.1	92.2	
55.6	53.0	52.1	<b>5</b> 2.5	95-3	93.7	
100.0	98.4	96.3	97.3	98.4	96.3	
180.0	179.2	175.8	177.5	99.6	97.7	

Unit:

mg/L.

Element value

EC<sub>sp</sub> at 24 hours >180.0 mg/L

EC<sub>so</sub> at 48 hours >180.0 mg/L

NOEC > 180.0 mg/L LOEC > 180.0 mg/L

- Statistical results as appropriate:Not applied.
- Remarks field for Results:

Biological observations

Not described.

Table showing mortality or immobility

Mortality or immobility of Daphnia magna to the test chemical

_	(Percent Mortality or Immobi						
	24 hour	48 hour					
Control	0(0)	0(0)					
Dissersant Control	0(0)	1(5)					
17.1	0(0)	1(5)					
30.9	(0)0	0(0)					
55.6	0(0)	0(0)					
100.0	0(0)	0(0)					
180.0	0(0)	0(0)					

Lowest test substance concentration causing 100% mortality:

Nominal concentration (mg/L)

Not obtained under the test conditions studied.

Mortality of controls:

No mortality observed during test period.

Abnormal responses:

No abnormal responses observed during test period

Reference substances: Potassium dichromate EC<sub>50</sub> at 48h was 0.87 mg/L.

Any observations, such as precipitation that might cause a difference between measurednd nominal

values:

It became clouded in high concentration, butnot precipitation.

Cumulative number of dead or immobilizes Daphnia

#### CONCLUSIONS

ECso (48h) > 180mg/L and NOEC (48h) > 180mg/L for Daphnia magna.

# DRAFT ENV/JM/EXCH(99)13

# DATA QUALITY

• Reliabilities: Klimisch Code: 1=reliable without restrictions.

· Remarks field for Data Reliability:

Experimental design and analytical procedure were well documented.

Carried out by Toray Research Center (Japan).

## REFERENCES

Environment Agency of Japan (1998).

# **OTHER**

- · Last changed:
- · Order number forsorting:
- Remarks field for GeneralRemarks:

# TOXICITY TO AQUATIC PLANTS (E.G., ALGAE)

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Tokyo Kasei Kogyo Co., Ltd. Lot No. AX01

Purity: >95.0%

**METHOD** 

Method/guideline followed: OECD TG 201

Test type:

Year:

Static.

• GLP:

Yes 1998

Species/strain # and source: Selenastrum capricornutum ATCC22662 (purchased from ATCC)

Element basis:

Area under the growth curve.

Exposure period:

72 h.

Analytical monitoring:

Yes, measured by HPLC at start and end of the test (72hr).

Statistical methods:

Bartlett test for homogeneity in variances and One-wayAnova (EcoTox-Statistics Ver.1.0 beta-edition R1.4) were used for EC<sub>20</sub>, LC<sub>20</sub> and NOEC.

determination (p=0.05).

#### Remarks field for Test Conditions:

Test organisms

Laboratory culture: OECD medium

Method of cultivation: Shaking at 100rpm

Controls: OECD medium. EC<sub>so</sub> of potassium dichromate was 0.41 mg/L.

Test Conditions

Test temperature range: 23±2 °C

Growth/test medium: OECD medium.

Shaking: 100 rpm

Dilution water source: OECD medium.

Exposure vessel type: 100 mL OECD medium in a 300 mL Erlenmeyer

flask with a silicon cap which allows ventilation.

Water chemistry in test (pH) in at least one replicate of each concentration (at start and end of the test); pH=7.3-7.4 at start and 8.3-8.8 at end of the

test (72 h).

Stock solutions preparation: No stock solution was prepared. Test chemical was diluted to 100mg/L (solubilizer, HCO-40 100mg/L) with

OECD medium and sterilised with filter before use.

Light levels and quality during exposure: 4,756-4,822 lux, continuous

illumination.

Test design

Number of replicates: Triplicate

Concentrations: 0, 100 mg/L and dispersant control were tested.

Initial cell number in cells/mL: 1x10<sup>4</sup>

Method of calculating mean measured concentrations Geometric mean.

#### RESULTS

Nominal concentrations:

0, 100 (mg/L) and dispersant control.

Measured concentrations:

At start of the test (0 hr), <1.0, 80.6, <1.0 (mg/L)At end of the test (72 hr), <1.0, 68.7, <1.0 (mg/L)

• Unit:

mg/L

Results:

(calculated based on nominal concentrations)

(1) Growth inhibition (comparison of area under growth curve)

 $EC_{50}$  (0-72 h) > 100 mg/L NOEC (0-72 h) > 100 mg/L

(2) Growth inhibition (comparison of growth rates)

 $EC_{50}$  (24-48) > 100 mg/L  $EC_{50}$  (24-72) > 100 mg/L NOEC (24-72) > 100 mg/L

Was control response satisfactory:

Yes: Mean cell density increased to 270x10<sup>6</sup> cells/mL (270-fold increase) after 72 hr for control. Mean cell density increased to 275x10<sup>6</sup> cells/mL (275-fold increase) after 72 hr for Dispersant control.

Statistical results as appropriate:

Significant difference in the growth curve was not observed between values at 100 mg/L and in each control.

### Remarks field for Results:

Biological observations

Cell density at each flask at each measuring point:

Nominal Concentration (mg/L)		Cell Density	(x104 cells/mL)	
	<b>0</b> hr	24 hr	48 hr	72 hr
Control	1.0±0.00	6.5±0.50	50.5± 3.48	270.5±23.50
Dspersant Control	1.0±0.00	9.3±1.66	57.5± 9.39	275.2±17.22
1 00	1.0±0.00	16.1±7.82	65.1±12.82	283.3± 7.98
	ACT . 1 I		and a second	

(Each value represents the mean of three sample counts.)

Growth curves: Logarithmic growth until end of the test (72 h).

Percent biomass/growth rate inhibition per concentration: Not described.

Observations: Test group(100mg/L) showed normal and similargrowth to that of control (283 fold increase after 72 hr).

## CONCLUSIONS

(1) Growth inhibition (comparison of area under growth curve)

 $EC_{50} (0.72 h) > 100 mg/L$ 

(2) Growth inhibition (comparison of growth rates)

NOEC (0-72 h) > 100 mg/L EC<sub>50</sub> (24-48) > 100 mg/L

 $EC_{50}$  (24-72) > 100 mg/L NOEC (24-72) > 100 mg/L

# DATA QUALITY

Reliabilities:

Klimisch Code: 1=reliable without restrictions.

Remarks field for Data Reliability:

Experimental design and analytical procedure were well documented. Carried out by Toray Research Center (Japan).

# REFERENCES

Environment Agency of Japan (1998).

# OTHER

Last changed:

Order number forsorting:

Remarks field for GeneralRemarks:

# CHRONIC TOXICITY TO AQUATIC INVERTEBRATES (e.g., DAPHNIA) (1)

## TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Nuoplaz 6965

METHOD

• Method:

ASTM and USEPA

Test type:

Flow-through condition

GLP:

Yes

· Year:

1984

Analytical procedures:

Yes. Measured by GLC, on 0,4,7,14,21day)

Species/Strain:

Daphnia magna

Test details:

Dynamic flow-through

Statistical methods:

ANOVA, 2WANOVA, arcsin transformation and Fisher's protected

Least Significant Difference (LSD)

# Remarks field for Test Conditions:

Test organisms:

Source; in house culture

Age at study initiation: Juveniles within 24h old. Control group: Yes (control and solvent control)

Test conditions

Dilution Solvent for Concentrated stock standards: Acetone (1.049mg/ml.)

A proportional diluter system was used for the intermittent introduction of test

material and dilution water into the test chambers.

Test temperature range: 18-22 °C (average temperature 20°C).

Well water was delivered to the chambers as a minimum rate of 2.0mL/min. Exposure vessel type: 900mL test solution in a 1000 mL glass beaker, 4

beakers per treatment

Dilution water chemistry: Hardness and other characteristics are reported.

Dilution water pH in test: pH=8.3-8.4.

Lighting: 37-74 footcandles, 16h:8h light-darkness cycle Feeding: Algae (Selenastrum capricornutum) three times a day

Supplemented with a trout chow suspension at least twice a week

Element (unit) basis:

Mean cumulative numbers of juveniles produced per adult (reproduction)

Growth (length) of parental Daphnia

Long-term survival

Test design:

Number of replicates=4; individuals per replicate=10;

Method of calculating mean measured concentrations Geometric mean.

Exposure period:

21 d

Analytical monitoring:

By GLC analysis. 33-101% of the nominal concentration at Preparation

# RESULTS

Nominal concentrations:

, 0.0074, 0.012, 0.027, 0.048, 0.100 mg/L

Measured concentrations:

Measured concentration of test chemical during 21-day exposure

Nominal concentration

Measured concentration (day, mg/L)

(mg/L)

0

7

14

21

mean

Control			ND	NI	)	ND	ND		ND	N.
Solvent Cont.			ND	NI	)	ND	ND	,	ND	NI
0.0074		(	0.00328		0.00366 0.			246	0.00482	
0.012		(	0.00748	0.0	0626	0.00843		1478	0.00747	
0.027		(	0.0172		150	0.0204	0.01		0.0157	0.01
0.048			0.0305		252	0.0371	0.01		0.0348	0.02
0.100		0	.0824	0.0	766	0.0870	0.06		0.1011	0.08
Cumulative Nun	ber o	f Dead	l Parent	a <b>D</b> aph	nia.					
Nominal conc.	Day	S		-						
(mg/L)	0	3	5	7	10	12	14	17	19	21
Control	0	0	0	0	0	0	0	1	1	2
Solvent Cont.	0	0	0	0	0	1	1	2	3	4
0.0074	0	0	0	Ō	0	1	1	1	1	1
0.012	0	0	0	Ö	0	0	Õ	Ō	ē	ō
0.027	0	0	0	0	0	0	0	0	0	Õ
0.048	0	0	0	0	1	1	1	1	1	1
0.100	0	O	0	0	0	0	0	D	Ô	õ
Control Solvent Cont. 0.0074 0.012 0.027 0.048 0.100		59.5 59.5 59.8 59.8	5 (n=9) 1 (n=7) 5 (n=10) 1 (n=10) 3 (n=10) 5 (n=10) 7 (n=10)	59 58 59 58 59	3.4 (n=9) 9.0 (n=10) 3.5 (n=10) 9.4 (n=10) 3.4 (n=10) 9.6 (n=10) 9.0 (n=10)	5) 59.6 0) 60.3 1) 59.0 0) 59.9 1) 59.8	8 (n=10) 0 (n=9) 1 (n=9) 5 (n=10) 9 (n=10) 8 (n=10)	) <u>:</u>	58.5 (n=1 59.3 (n=1 59.5 (u=1 59.8 (n=1 60.3 (n=1 58.6 (n=1 59.0 (v=1	0) 0) 0) 0) 0)
Mean number	s of in:	star pr	oduced	during	g 21-d.					
Nominal conc.	Day									
(mg/L)	0	3	5	7	10	12	14	17	19	21
Control	-	-	•	-	109	196	317	86	179	170
Solvent Cont.	-	•	•	16	164	178		240	75	156
0.0074	-		-	3	141	202	302	261	75	274
0.012	-		• .	3.5	122	206	373	221	96	265
0.027	-	-	-	8.3	150	189	317	218	138	313
0.048	-	-	•	-	113	203	242	120	233	214
0.100				5.3	135	186	223	180	93	269

# Statistical results as appropriate:

Calculated LC<sub>50</sub> Value for Parental *Duphnia*: LC<sub>50</sub>(21day) > 0.082(mg/L) Calculated EC<sub>50</sub> value for Inhibition of Reproduction: EC<sub>50</sub>(21day) > 0.082(mg/L)

## Remarks field for Results:

Biological observations

Cumulative numbers of dead parental Daphnia: Control: 2 (mortality: 5%), Solv. Cont.: 4 (mortality: 10%) 0.0074 mg/L: 1 (mortality: 2.5%) 0.012 mg/L: 0 (mortality: 0%) 0.027 mg/L: 0 (mortality: 0%) 0.048 mg/L: 1 (mortality: 2.5%)

0.100 mg/ L: 0 (mortality: 0%)

Time of the first production of juveniles:Control:

7-10d

Solvent control:

5-7d

0.0074 mg/L:

5-7d

0.012 mg/L:

5-7d

0.027 mg/L:

5-7d

0.048 mg/L:

7-10d

0.100 mg/ L:

5-7d

Mean cumulative numbers of juveniles produced per adult alive for 21days:

Control:

112.7

Solvent control:

168.5

0.0074 mg/L:

119.6

0.012 mg/L:

139.3

0.027 mg/L:

133.3

0.048 mg/L:

116.0

0.100 mg/L

112.9

Was control response satisfactory: Yes.

# CONCLUSIONS

·NOEC (21-d, reproduction): 0.082 mg/L,

·LOEC (21-d, reproduction):

>0.082 mg/L.

·ECso (21-d, reproduction):

 $>0.082 \text{ mg/L}_{:}$ 

·LCso for parental Daphnia (21-d): >0.082 mg/L

# DATA QUALITY

- Reliabilities:
- Remarks field for Data Reliability:

Experimental design and analytical procedure were well documented.

Carried out by Analytical Biochemistry Laboratories, Inc.,

# REFERENCES

CMA Doc. I.D. 40-8565036 (1985).

## OTHER

- Last changed:
- Order number for sorting ;
- Remarks field for GeneralRemarks:

# CHRONIC TOXICITY TO AQUATIC INVERTEBRATES (e.g., DAPHNIA) (2)

## **TEST SUBSTANCE**

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Tokyo Kasei Kogyo Co., Ltd. Lot No. AX01

Purity: >95.0%

METHOD

Method:

OECD TG 211 (revised edition of No.202).

Test type:

Semi-static.

• GLP:

Yes

Year:

1998

Analytical procedures:

Yes. Measured by HPLC 2-3 times a week (before and afterthe

replacement of the test water)

Species/Strain:

Daphnia magna

Test details:

Semi-static (water renewal: 3 times a week), open-system.

Statistical methods:

Eco-Statics (Version 1.0 beta-edition R1.4)

## Remarks field for Test Conditions:

Test organisms:

Source, supplier, any pre-treatment, breeding method: Supplied by NIES

(Japan)

Age at study initiation: Juveniles within 24h old.

Control group: Yes.

Test conditions

Stock solutions preparation and stability: No solvent used. Test chemical was diluted to 1.0wt.% (with solubilizer HCO-40 1.0wt.% controlled)

with diluting water (Elendt M4) before use. Solubilizer concentration was controlled 100mg/L with working solution (HCO-40 1.0wt.%). Test temperature range: 19. 9-20.8 °C (average temperature 20°C). Exposure vessel type: 80mL test solution in a 100 mL glass beaker: 10

beakers per treatment

Dilution water source: Elendt M4(OECD guideline No.211 Annex 2)

Dilution water chemistry: Hardness: 251mg/L as CaCO<sub>3</sub>

Lighting: <1,200 lx, 16h:8h light-darkness cycle

Water chemistry in test: DO= 7. 0-9.2mg/L; pH=7.4-7.9. Feeding: Chlorella regularis, 0.1-0.2 mgC/day/individual

Element (unit) basis:

Mean cumulative numbers of juveniles produced per adult (reproduction)

Test design:

Number of replicates=10; individuals per replicate=10;

Concentrations: 0, 55.6, and 100 mg/L, because 48h-EiC<sub>50</sub> for parent Daphnia (Acute immobilization test) was>180 mg/L. Dispersant control

was also tested.

Method of calculating mean measured concentrations:Geometric mean.

Exposure period:

21 d

Analytical monitoring:

By HPLC analysis. 99.7-101.3% of the nominal concentration at

preparation; 94.7-99.3% just before the renewal of the test water (after 2

days exposure).

## RESULTS

- Nominal concentrations: 0, 55.6, 100 mg/L
- Measured concentrations: Time-weighted measured concentrations of test chemical during a 21-day exposure were 54.8 and 98.7 mg/L.

# Measured concentration of test chemical during 21-day exposure

Nominal concentration	Measured concentration (day, mg/L)							
(mg/L)	O(new)	2 (old)	7(new)	9(old)	16(new)	19(old)		
Control	< 1.0	< 1.0	< 1.0	< 1.0	< 1.0	< 1.0 É		
Disp.Cont.	< 1.0	<1.0	< 1.0	< 1.0	< 1.0	< 1.0		
55.6 <sup>°</sup>	56.3	54.4	<i>55.</i> 4	53.9	56,3	52.6		
100	100.4	99.3	100.0	98.5	99.8	95.2		

new: freshly prepared test solutions. old: test solution after 2 days exposure.

### Unit:

nig/L

- ·NOEC (21-d, reproduction): 55.6 mg/L,
- ·LOEC (21-d, reproduction): >100 mg/L,
- -ECso (21-d, reproduction): 89.1 mg/L;
- LCse for parental Daphnia (21-d): >100 mg/L; calculated based on nominal concentrations.

# Cumulative Number of Dead Parental Daphnia.

Nominal conc.	Daj	y 3																		•	
( mg/L)	1	2	3	4	5	6	7	8	9 -	10	11	12	13	14	15	16	17	18	19	20	21
Control	0	0	0	0	0	0	0	0	0	0	0	Ò	0	0	0	0	0	0	0	0	0
Disp.Cont.	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
55.6	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0
100	0	0	0	0	0	0	0	0	0	0	0	0	0	0	0	. 0	1	2	2	2	2

# Mean cumulative numbers of juveniles produced per adult during 21-d.

rominal conc.	Days		·	- 1
(mg/L)	1 2 3 4 5	6 7 8 9 10 11	12 13 14 15 16 17 18 19 20 21	1
Control	0.0 0.0 0.0 0.0 0.0	0.0 0.0 1.8 2.2 7.1 7.7	8.2 19.6 20.4 23.2 43.8 48.0 61.6 83.0 88.0 8	8.7
Disp.Cont.	0.0 0.0 0.0 0.0 0.0 0	0.0 0.0 0.3 0.3 8.2 8.2	8.7 29.2 31.9 33.0 55.8 61.5 64.8 72.0 73.8 7	73.8
55.6	0.0 0.0 0.0 0.0 0.0 0	0.0 0.0 0.2 1.0 2.0 2.7	5.1 9.3 13.6 26.6 34.4 43.9 51.4 66.2 74.3 7	اوو
100	0.0 0.0 0.0 0.0 0.0 0	0.0 0.0 0.0 0.0 1.6 2.6	3.6 7.8 9.3 11.0 15.1 17.5 20.3 30.3 33.0 3	3.0

# Cumulative Number of Juveniles produced per Adult Alive for 21-d.

Vessel No.	Cont.	Disp.Cont.	Nominal 55.6	Concentration(mg/L) 100.0
1	74	74	68	37
2	57	71	70	25
3	126	92	65	•
4	127	78	96	•
. 5	90	73	89	36
б	84	70	116	29
7	71	76	78	35
8	94	84	93	28
9	<i>7</i> 8	75	87	34
10	86	45	37	40
Mean (S.D)	88.7(22.524)	73.8(12.072)	79.9(21.533	33.0(5.127)
Inhibition rate(9	6)	0.832	0.901	0.372

Significant difference\*1

-: were not calculated because the parental Daphaia was dead during a 21-days testing period.

1\*: Indicates a significant difference by Dunnet multiple comparison procedure, Two-sides test.

\*\*:Indicates a significant difference (alpha=0.01) from the control.

# Statistical results as appropriate:

Calculated LC<sub>50</sub> Value for Parental Daphnia: LC<sub>50</sub>(21day) >100(mg/L)

Calculated EC<sub>so</sub> value for Inhibition of Reproduction:  $EC_{so}(21\text{day}) = 89.1(\text{mg/L})$ 

(Statistical method: Logit)

#### Remarks field for Results:

Biological observations

Cumulative numbers of dead parental Daphnia: Control: 0 (mortality: 0%),

Disp.Cont.: 0(mortality: 0%) 55.6 mg/L: 0(mortality: 0%)

100 mg/L: 2 (mortality: 20%)

Time of the first production of juveniles: 8-13d for control

8-12d for dispersant control

8-13d for 55.6 mg/L

10-14d for 100 mg/L

Mean cumulative numbers of juveniles produced per adult alive for 21 days:

Control: 88.7, Dispersant control: 73.8

55.6 mg/L: 79.9, 100 mg/L: 33.0

Was control response satisfactory: Yes. Mean cumulative numbers of uveniles produced per

adult was 88.7 and 73.8 > 60.

#### CONCLUSIONS

·NOEC (21-d, reproduction): 55.6 mg/L,

·LOEC (21-d, reproduction): >100 mg/L,

·ECso (21-d, reproduction): 89.1 mg/L;

·LCso for parental Daphnia (21-d): >100 mg/L; calculated based on nominal concentrations.

## DATA QUALITY

Reliabilities:

Klimisch Code: 1=reliable without restrictions.

Remarks field for Data Reliability:

Experimental design and analytical procedure were well documented.

Carried out by Toray Research Center (Japan).

#### REFERENCES

Environment Agency of Japan (1998).

#### OTHER

- Last changed:
- Order number forsorting:

· Remarks field for GeneralRemarks:

# **HEALTH ELEMENTS**

#### ACUTE ORAL TOXICITY

# TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Daihachi Kagaku Kogyo Co., Ltd. Lot. No. N-60601

Purity: >99.0%

Kept at room temperature in a dark place until use. Stability of mixture of

dose was confirmed for 7 days under 4C.

METHOD

Method:

OECD TG 401

Test type:

Single Dose Oral Toxicity Test

• GLP:

Yes

• Year:

1996

Species:

Rat

• Strain:

Crj: CD(SD)

Route of administration:

Oral (by single-dose gavage)

Doses/concentration levels: 0(vehicle) and 2,000 mg/kg

Sex:

Male & Female

· Vehicle:

Corn oil

Post exposure observation period: Two weeks.

Statistical methods:

Not applicable because of no fatality.

# REMARKS FIELD FOR TEST CONDITIONS

Test Subjects:

Age at study initiation: 6 weeks old for both sexes.

Weight at study initiation: 149-163 g for male.

126-140 g for female

No. of animals per sex per dose: 5 per sex per dose group

Study Design:

Vehicle: Com oil. 40.0w/v% for 2000 mg/kg.

Satellite groups and reasons they were added: None Clinical observations performed and frequency:

Each rat was weighed immediately prior to treatment,7 and 14 days after post-treatment observation period. The rats were observed each hour to 6hr, after that, 2 times for one day during this time for signs of toxicity.

RESULTS

LD<sub>50</sub>:

Male :> 2,000 mg/kg

Female: > 2.000 mg/kg

REMARKS FIELD FOR RESULTS.

# DRAFT ENV/JM/EXCH(99)13

Body weight:

The test substance did not cause any changes in body weight.

No detailed body weight data available.

Food/water consumption:

No detailed data available.

Clinical signs:

Loosening erring of the stool attributable to the treatment with corn oil was observed for 3 hours from the administration for both sexes in the groups given 0 and 2000 mg/kg. However, no deaths occurred of either male or

female animals.

Haematology:

Not done

Biochem:

Not done.

Ophthalmologic findings: Not examined.

Mortality and time to death: No deaths were recorded in treated and control group.

Gross pathology incidence and severity: No macroscopic abnormalities that could be attributes to

treatment with the test substance were seen on pathological examination.

Organ weight changes:

Not done.

Histopathology (incidence and severity): Not done.

#### CONCLUSIONS

 $LD_{40}$  was established at > 2,000 mg/kg for both sexes.

## DATA QUALITY

Reliabilities:

Klimisch Code: 1=reliable without restrictions.

Remarks field for Data Reliability:

Well conducted study, carried out by the Biosafety Research Center, Food, Drugs and Pesticides (An-pyo Center), Japan

#### REFERENCES

Toxicity Testing Reports of Environmental Chemicals, vol.4(1996)

Ministry of Health & Welfare, Japan

#### ACUTE INHALATION TOXICITY

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Nouplaz 6959, Batch No. 39049

Purity: 98.95%

#### **METHOD**

Method:

Not specified

• GLP:

Yes

· Year:

1982

Species:

Rat

Strain:

Crj: CD(SD)

Doses/concentration levels: 2,600 mg/m³

Sex:

Male & Female

Post exposure observation period: Two weeks.

Statistical methods:

Not applicable because of no fatality.

#### REMARKS FIELD FOR TEST CONDITIONS

Test Subjects:

Age at study initiation: Not stated.

Weight at study initiation: 210-275 g for both sexes.

No. of animals per sex per dose: 5 per sex per dose group

Study Design:

Inhalation Chamber: A 0.5m<sup>3</sup> stainless steel inhalation chamber was used.

( Youg and Bertke, Cincinnati, Ohio)

The test compound atmosphere was generated directly into the chamber by

means of Jet Nebulizer Mechanism. Chamberconcentrations were

monitored by a filter paper/gravimetric techniquespproximately every 30

min during the exposure period.

The HEPA filtered chamber air-flow was maintained between 10 to 20 air changes per hour during the exposure period with the chamber under

slightly negative pressure.

The temperature in the chamber was maintained at 69-75 degree F with

relative humidity of 30-50%

Satellite groups and reasons they were added: None

Clinical observations performed and frequency:

After the exposure, all animals were observed daily for 14 days for clinical signs of toxicity. Body weights were recorded prior to exposure and weekly thereafter. All animals were subjected to necropsy at termination of the

study.

#### RESULTS

LD<sub>a</sub>:

Male : >  $2,600 \text{ mg/m}^3$ 

Female: > 2,600 mg/m $^{3}$ 

#### REMARKS FIELD FOR RESULTS.

Body weight:

The test substance did not cause any changes in hody weight.

Mean body weight(g) of rats exposed to this chemical

Males Initial weight 265.1(8.40)

First week

297.8(14.02)

Second week

329.7(15.27)

Females Initial weight

213.9(2.66) 223.2(3.96)

First week Second week

238.1(4.82) Mean(S.D.)

Food/water consumption: No detailed data available.

Clinical signs:

All animals (male and female) had matted, drenched coats for the first 2

days, otherwise no visible signs.

Haematology:

Not done

Biochem:

Not done.

Ophthalmologic findings: Not examined.

Mortality and time to death: No deaths were recorded.

Organ weight changes:

Not done.

General necropsy observations: All males and 3/5 females exhibited reddening patches on lungs.

#### CONCLUSIONS

LD<sub>0</sub> was 2,600 mg/m<sup>3</sup> for both sexes.

# DATA QUALITY

Reliabilities:

Klimisch Code: 1=reliable without restrictions.

Remarks field for Data Reliability:

Well conducted study, carried out by Midwest Research Institute.

#### REFERENCES

Nuodex Inc. Acute inhalation toxicity test in SpragueDawley rats using compoundNouplaz 6959

Environmental Protection Agency (1983)

#### ACUTE DERMAL TOXICITY

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Nouplaz 6959, Batch No. 39049

Purity: 98.95%

METHOD

Method:

Procedure set forth in the Federal Insecticide, Fungicide, andRodenticide

Act (FIFRA)

• GLP:

Yes

• Year:

1981

Species:

Rabbits

Strain:

New Zealand albino white rabbits

Doses/concentration levels: 2.0 mL/kg

Sex:

Male & Female

· Post exposure observation period: Two weeks.

Statistical methods:

Not applicable because of no fatality.

# REMARKS FIELD FOR TEST CONDITIONS

Test Subjects:

Age at study initiation: Not stated.

Weight at study initiation: 2.3-3.2 kg for both sexes.

No. of animals per sex per dose: 3 per sex per dose group and 2 per sex

for control.

Study Design:

Procedure: 24 hours prior to treatment the hair on the back of each rabbit was clipped so as to expose approximately 10% of the body surface area. Before dosing, epidermal abrasions were made longitudinally over the exposure area. The abrasions were sufficiently deep to penetrate the

stratum comeum but not so deep as to cause bleeding.

A dosage was applied to the exposure area. A 2 x 2-inch gauze pad was placed on the exposure area to prevent seepage of the compound from the area. Each animal was then wrapped with a rubber dam. After 24 hour of exposure, the rubber dam and gauze pad were removed, and the exposure

area was wiped to remove any remaining test material. Satellite groups and reasons they were added; None Clinical observations performed and frequency:

After the exposure, all animals were observed daily for 14 days for clinical signs of toxicity. A gross necropsy was performed on all animals at the end

of the 14 day observation period.

#### RESULTS

LD<sub>so</sub>:

Male :> 2.0 mL/kg Female:> 2.0 mL/kg

#### REMARKS FIELD FOR RESULTS.

Body weight:

The test substance did not cause any changes in body weight.

Individual Animal Boy Weights

	Sex	Body		
Control		day I	day 7	day 14
	male	3.2	3 <i>A</i>	3.6
		3.2	3.4	3.6
	Tennale	2.7	3.0	3.1
		2,9	3.1	3.3
2.0 mL/kg	male	2.3	2.3	2.5
•	,	2.4	2.4	2.5
		2.3	2.2	2.4
	female	2,3	2.5	2.7
		2.4	2.6	2.7
		2.4	2.5	2.6

Food/water consumption: No detailed data available.

Clinical signs:

No toxic sign.

Haematology:

Not done Not done.

Biochem:

Ophthalmologic findings: Not examined.

Mortality and time to death: No deaths were recorded.

Organ weight changes:

Not done.

Gross Pathology:

Nothing noted.

#### CONCLUSIONS

LD<sub>50</sub> was 2.0 mL/kg for both sexes.

## DATA QUALITY

Reliabilities:

Klimisch Code: 1=reliable without restrictions.

Remarks field for Data Reliability:

Well conducted study, carried out by Midwest Research Institute.

#### REFERENCES

Nuodex Inc. Acute dermal toxicity test of Tenneco Chemicals Inc. compound/ouplaz 6959 in

rabbit.

Environmental Protection Agency (1981)

#### SKIN IRRITATION

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Nouplaz TOTM(Tenneco Chemicals, Inc.)

Purity: 98.95%

#### METHOD

The test method was similar to Section 1500.41. Federal Hazardous Method: Substances Act Regulations - 16 CFR

GLP:

Yes

Year:

1981

Species:

Rabbits

Strain:

New Zealand albino white rabbits

- Doses/concentration levels: 0.5 mL
- Post exposure observation period: 24, 72 hours
- Statistical methods:

Not applicable because of no fatality.

#### REMARKS FIELD FOR TEST CONDITIONS

Husbandry Conditions Temperature - 70 ± 2 degree F

Relative Humidity - 45% ± 5% Light - 12 hour light/dark cycle

Diet - Wayne 15% Rabbit Ration and tap water are provided ad

libitum. Based on our current knowledge no contaminants are known to be in this diet or water that might be expected to

interfere with the objectives of the study.

Caging - Stainless steel with elevated wire mesh flooring 1 rabbit/cage

Bedding - Techbord

Shepherd Products Company Kalamazoo, Michigan 49005

Test method:

A 0.5 mL portion of material was applied to an abraded and an intact akin site on the same rabbit. Gauze patches were then placed over the treated areas and an impervious material was wrapped snugly around the trunks of

the animals to hold the patches in place.

The wrapping was removed at the end of the twenty-four (seventy two) hour period and the treated area were examined. TheDraize method of

scoring was employed.

Evaluation: Draize Scale For Scoring Reactions

Erythema and Eschar Formation:	<b>Value</b>
No erythema	0
Very slight erythema(barely perceptible)	1
Well defined crythem	2
Moderate to severe erythema	.3
Severe erythema (beet redness) to slighteschar formation	
(injuries in depth)	4

Edema Formation	Value
No edema	0
Very slight edema(barely perceptible)	1
Slight edema(edges of area well defined by definite raising)	2
Moderate edema (raised approximately 1 millimeter)	3
Severe edema (raised more than 1 millimater and extending	
beyond the area of exposure)	4

# RESULTS

Primary Irritation Score: 4.16/4 = 1.04

# REMARKS FIELD FOR RESULTS.

	Reading							
Erythema and Eschar Formation	(Hours)	1	2_	3	4	5_	6	Average
Intact skin	24	2	1	2	1	2	1	1.50
Intact skin	72	0	Ð	1	Q	0	0	0.17
Abraded skin	24	2	1	2	1	2	1	1.50
Abraded skin	72	•	0	1	1	Û	0	0.33
						Subto	ta]	3.50
Edema Formation								•
Intact skin	24	1	0	0	0	1	Q	0.33
Intact skin	72	0	0	0	0	0	0	0.00
Abraded skin	24	1	0	Ð	0	1	0	0.33
Abraded skin	72	0	0	0	0	0	O	0.00
						Subtot	e l	0.66
						Tot	ai	4.16

# CONCLUSIONS

Slightly irritating

This report concluded that TOTM was not a primary skin irritant in rabbit. It is not possible to assign a classification.

# **DATA QUALITY**

• Reliabilities: Klimisch Code: 1= reliable without restrictions.

Remarks field for Data Reliability:

Well conducted study, carried out by Biosearch Inc.

# REFERENCES

Nuodex Inc. Primary Skin Irritation - Rabbits. OTS 2065758. Doc ID 878214470,1981

#### EYE IRRITATION

#### TEST SUBSTANCE

Remarks:

Identity;

Tris(2-ethylhexyl)bcnzene-1,2,4-tricarboxylate

Source: Nouplaz TOTM(Tenneco Chemicals, Inc.)

Purity: 98.95%

#### METHOD

Method:

The lest method was similar to Section 1500.42.Federal Hazardous Substances Act

Regulations - 16 CFR.

GLP:

Yes

· Year:

1981 Rabbits

Species:Strain:

New Zealand albino white rabbits

Numbers of animals

6

Doses/concentration levels: 0.1 mL

Sex:

Post exposure observation period: 1,2,3,4,7 days

Statistical methods:

Not applicable because of no fatality.

#### REMARKS FIELD FOR TEST CONDITIONS

Husbandry Conditions Temperature - 70 ± 2 degree F

Relative Humidity - 45%± 5% Light - 12 hour light/dark cycle

Diet - Wayne 15% Rabbit Ration and tap water are provided ad libitum. Based on our current knowledge no contaminants are known to be in this diet or water that might be expected to

interfere with the objectives of the study.

Caging - Stainless steel with elevated wire mesh flooring 1 rabbit/cage

Bedding - Techbord

Shepherd Products Company Kalamazoo, Michigan 49005

Test method:

0.1 mL of the experimental material was instilled into the right eyes of the test animals while the other eyes remained untreated to severe as controls.

The treated eyes were examined at one, two, three, four and seven days

Following instillation of the test materials into the eyes.

Evaluation:

Interpretation of the results was made in accordance with theDraize Scale

of Scoring Ocular Lesions.

Scale of Scoring Ocular Lesions

(1) CORNEA

Value range

A. Opacity - Degree of Density(area most dense taken for reading)

0 - 4

B. Area of Cornea Involved

1 - 4

Score equals A x B x 5 (Total Maximum = 80)

(2) IRIS

A. Values	0 - 2
Score equals A x 5 (Total Maximum = 10)	
(3) CONJUNCTIVAE	
A. Redness (refers to palpebral and bulbar conjunctivae	
excluding cornea and iris)	0 - 3
B. Chemosis	0 - 4
C. Discharge	0 - 3
Score equals (A+B+C) x 2 (Total Maximum =20)	<b>5 5</b>

## RESULTS

Average Ocular IrritationScore: 2.3(1 day), 1.7(2day), 0(3,4,7day)

# REMARKS FIELD FOR RESULTS.

		Reading								
_Rabbi	numher Tissue	Lday_	2 day	3 day	4 day	7day				
1	(1) Cornea tota)	0	•	0	9	Ō				
	(2) Iris total	0	0	0	Ð	Ð				
	(3) Conjunctivae total	2	2	0	0	0				
	Taial Ocular Irritation Score	2	2	0	0	0				
2	(1) Cornea total	Ð	0	0	•	0				
	(2) Iris total	0	0	0	0	0				
	(3) Conjunctivae total	4	2	Ð	0	0				
	Total Ocular Irritation Score	4	2	0	0	•				
3	(1) Cornea total	D	0	0	Ð	0				
	(2) Iris total	0	0	0	0	0				
	(3) Conjunctivae total	2	2	0	0	0				
	Total Ocular Irritation Score	2	2	0	Ø	0				
4	(1) Cornea total	0	0	Ð	0	0				
	(2) Iris total	0	Ð	Q.	0	0				
	(3) Conjunctivae total	2	2	•	0	0				
	Total Ocular Irritation Score	2	2	0	Q	0				
5	(1) Cornen total	0	0	0	0	0				
	(2) Iris total	<b>Q</b> .	Ð	ð	6	Ö				
	(3) Conjunctivae total	2	2	0	0	0				
	Total Ocular Irritation Score	2	2	à	0	0				
6	(1) Cornea total	Ð	0	0	0	0				
	(2) Îris total	0	0	0	0	0				
	(3) Conjunctivae total	2	0	0	0	0				
	Total Ocular Irritation Score	2	0	D	0	0				
	Average Ocular Irritation Score	2.3	1.7	0.0	<b>9</b> .0	0.0				

# CONCLUSIONS

Slightly irritating

This report concluded that TOTM was not a primary skin irritant in rabbit. It is not possible to assign a classification.

# **DATA QUALITY**

• Reliabilities: Klimisch Code: 1=reliable without restrictions.

Remarks field for Data Reliability:

Well conducted study, carried out by Biosearch Inc.

# REFERENCES

Nuodex Inc. Primary Eye Irritation - Rabbits. OTS 2065758. Doc ID 878214471,1983

# DRAFT ENV/JM/EXCH(99)13

#### SENSITIZATION

#### TEST SUBSTANCE

• Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Nouplaz TOTM(Tenneco Chemicals, Inc.)

Purity: 98.95%

#### **METHOD**

Method:

Buehler teat

• GLP:

Yes

Year:

1981

Species:

Guinea pig

Strain:

Albino guinea pig

Numbers of animals

10

Doses/concentration levels: 0.5 mL

Sex:

male

• Post exposure observation period:10 application

Statistical methods:

Not applicable because of no fatality.

## REMARKS FIELD FOR TEST CONDITIONS

Husbandry Conditions Temperature - 70 ± 2 degree F

Relative Humidity-45% ± 5% Light - 12 hour light/dark cycle

Diet - Charless River Guinea Pig Furmula and tap water are provided ad Libitum. Based on our current knowledge no contaminants were Known to be in this diet or water that might be expected to

Interfere with the objectives of the study.

Caging - Stainless steel with clevated wire mesh flooring 5 guinea pigs/cage

Bedding - Deotized Animal CageBoard(DACB)

Shepherd Products Company Kalamazoo, Michigan 49005

Test method:

A 0.5 mL portion of material was applied to the intact akin test site on the guinea pigs. A gauze patch was placed over the treated area and an impervious material was wrapped snugly around the trunks of the animals to hold the patches in place. After a 24 hour contact period the patch was removed and the animals were allowed to rest for one day. Following this rest period another application was applied to the same skin site using a fresh sample. After the tenth application the animals were rested for a two week period. At the termination of the rest period a challenge application was put on skin sites differing from the original test sites. The challenge application remained on for 24 hours.

The sites were examined for reaction using the Draize method of scoring to grade reactions.

Evaluation: Draize Scale For Scoring Reactions

Ervihema and Eschar Formation:

Value

No erythema

0

·	Very slight erythema(barely perceptible)	1
	Well defined erythem	2
	Moderate to severe erythema	3
	Severe erythema (beet redness) to slighteschar formation(injuries	in depth) 4
	Edema Formation	Value
	No edema	O
	Very slight edema(barely perceptible)	1
	Slight edema(edges of area well defined by definite raising)	2
	Moderate edema (raised approximately 1 millimeter)	3
	Severe edema (raised more than 1 millimater and extending	
	beyond the area of exposure)	4

#### RESULTS

# No sensitization

# REMARKS FIELD FOR RESULTS.

		Reading After Application number									Challenge			
Guinea pig No.		L_	2_	_3_	_4_	5	<u>.</u> 6_	7	_8_	9	10		24hours 48hours	
1	Erythema	. 0	0	0	0	0	0	0	0	Ð	Đ	0	0	
	Edems	0	0	0	0	0	0	0	Œ	Ð	0	0	Ð	
2	Erythema	0	0	0	0	Ð	0	0	0	D	0	0	0	
	Edema	0	0	0	0	0	0	0	0	0	0	0	Ō	
3	Erythema	0	0	0	0	D	0	ø	ø	0	0	0	0	
	Edema	0	0	D	0	û	ø	0	0	ũ	0	0.	o o	
4	Erythema	0	0	D	0	0	Ò	ø	0	0	0	0	Ō	
	Edema	0	0	0	B	0	0	0	0	0	0	0	0	
5	Erythema	0	0	0	ø	0	Û	Ð	Û	0	0	Q	0	
	Edema	0	0	0	Ð	â	•	0	0	0	0	Û	0	
6	Erythema	•	0	0	0	0	O	0	0	Q	0	0	O	
	Edema	0	0	0	0	0	•	0	0	0	Ð	0	0	
7	Erythema	0	0	0	0	0	0	Ð	0	9	•	0	0	
	Edema	0	Q.	e	0	Ď	0	0	0	0	0	0	0	
8	Erythema	Ð	O	0	0	0	0	0	0	0	0	0	Ð	
	Edema	ō	0	0	Û	0	ð	0	0	Ø	0	D	Ō	
9	Erythema	D	0	0	0	0	9	0	Ð	0	Ð	Ü	Đ	
	Edema	0	0	0	0	0	0	0	Ð	ō	0	0	o o	
10	Erythema	0	0	0	0	0	0	0	0	0	O	0	O	
. –	Edema	0	0	Ō	0	0	Ð	o	0	Ð	0	6	å	

# CONCLUSIONS

No sensuization

# DATA QUALITY

Reliabilities:

Klimisch Code: 1=reliable without restrictions.

Remarks field for Data Reliability:

Well conducted study, carried out by Biosearch Inc.

# REFERENCES

Nuodex Inc. Guinea Pig Contact Dermal Irritatiom/Sensitization-Modified Buehler Method OTS 266574. Doc ID 878214475,1981

# REPEATED DOSE TOXICITY (a)

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Nuoplaz 6959

Purity: 98.2% (GC/FID) 97.9% (HPLC)

Impurities were detected at level than 0.1-0.5%, one being di(2-ethylhexyl)

phthalate (DEHP).

METHOD

Method:

**BIBRA Standard Operating Procedures** 

Test type:

Repeat Dose Toxicity

GLP:

Yes

Year:

1984

Species:

Rat

Strain:

Fischer 344

Route of administration

Oral Doses/concentration levels: 0(0), 0.2(184), 0.67(650) and 2(1826) % (mg/kg bw/day)

Vehicle:

Rodent diet

Sex:

Male & Female

Exposure period:

28 days

Frequency of treatment:

Once daily

Control group and treatment: Dietary level 0% and reference compound DEHP 0.67%.

Post exposure observation period: None

Duration of test:

Males and females; for 28 days

Statistical methods:

The control and TOTM treated groups were subject to analysis of

variance, and if this was significant the treated groups were compared with

the controls using the Least Significant Difference test.

The controls and DEHP groups were compared using a two-tailed pooled student t test with Welch's correction. In all cased a probability level of

P<0.05 was taken to indicate statistical significance.

#### REMARKS FIELD FOR TEST CONDITIONS

Test Subjects:

Age at study initiation: 48-51 days old for males and females

Weight at study initiation: 137-154g for male.

111-132g for female.

No. of animals per sex per dose: 5 Rats per sex per dose group

Study Design:

Vehicle: Dict

Satellite groups and reasons they were added: None Clinical observations performed and frequency:

Body wt. was recorded immediately prior to the first exposure and again for each

animal 1, 3, 7, 10, 14, 17, 21, 24, 27 <sup>6</sup> days.

Twice each day the animals were observed in their cages forvariations in behaviour or condition, and once weekly a more detailed examination was made at the time of

a weighing.

Food intakes were measured over the period day-3 to 0 and continuos intakes were measured at twice-weekly intervals until the day preceding autopsy. The intakes of test article or reference compound for each animal were calculated twice weekly using the analysed dietary concentrations of TOTM or DEHP, and the individual valued for bodyweight and food intake.

Hematologic parameters were evaluated for each animal. On the day preceding the start of the autopsies a sample of blood was collected from a caudal voin of each animal.

Autopsy: At the end of the 28th day treatment period the rats were deprived of food overnight, with water available. On the day of autopsy each animal was weighted and then killed. The blood was used to provide serum for clinical chemistry. During the autopsy any abnormalities of the external condition and of the thoracic or abdominal viscera were noted.

Organs: The weight of the following organs were recorded: adrenal glands, lungs, brain, ovaries, heart, spleen, kidneys, testes, liver, thyroid.

Electron microscopy: Two thin slices of liver, one from the left lobe, the other from the median lobe, were fixed for analysis. (The remainder of the liver was used for biochemical analysis.)

Biochemical analysis of the liver: Whole homogenates were prepared and assayed for protein and cyanide-insensitivepalmitoyl-CoA.

#### RESULTS

NOAEL

184 mg/kg bw

LOAEL

650 mg/kg bw

## REMARKS FIELD FOR RESULTS.

Body weight:

No statistically significant differences of bodyweight between the control and TOTM or DEHP treated groups of either sex. There was a trend for the male rats from all the TOTM treated groups to be lighter than the controls. In the females, this trend was only evident in the 2.0% TOTM group.

Food/water consumption: Female rats fed 2.0% TOTM consumed significantly less diet than the controls during first seven days of treatment after which their intakes increased but remained lower than those of the controls. In the males there were no statistically significant differences between the control and TOTM fed groups during the treatment period.

Haematology: In both sexes haemoglobin concentration of the rats given diet containing 0.67 or 2.0% TOTM were statistically significantly lower than the control. In the males there was a small lowering of erythrocyte count in all groups given TOTM but this was not reproduced in the females.

Both sexes given the two higher dietary concentrations of TOTM had higherleucocyte counts than the control, but the differences were statistically significant onlyin the males. These male groups also had lower proportions of the leucocytes assosinophils and monocytes.

Significantly lower values for haemotocrit and mean cell volume were limited to females given the two lower dose levels of TOTM.

In both sexes the liver weights, and liver weights relative to bodyweight, were Organ weights:

increased in the TOTM and DEHP treated animals compared to the controls. These differences were small and not statistically significant in the 0.2% TOTM group. The increase seen in the rats given 2.0% TOTM was less than that in those given DEHP. In the males fed TOTM the higher values for brain weights relative to body weight, in the absence of any significant differences in the recorded weight probably reflect the lower bodyweights in the groups concerned. In the females there were statistically significant higher lung weights in the rats fed 0.2 or 0.67% TOTM when compered to the controls. In the case of the TOTM treatedanimals this difference was not dose related and not statistically significant when expressed relative to bodyweight.

Serum analyses: Analysis of serum from the males and females showed statistically significantly increased levels of albumin in the groups given 0.67 or 2.0% TOTM. In the males there were statistically significantly higher cholesterol levels in the 0.67 and 2.0% TOTM groups.

> Concentrations of serum urea were statistically significantly increased in the male 2.0% TOTM group to the control values. In the females there was also an isolated statistically significantly lower value for lipid concentration in the 0.2% TOTM group.

Liver Biochemistry: Neither TOTM or DEHP treatment influenced to a statistically significant degree the concentration of hepatic protein. After TOTM treatmentPCoA activity was statistically significantly higher than controls in both sexes at the highest dose and in the males at the lower two doses. In the groups given TOTM only the highest dose level males had statistically significant increases of enzyme level. Both sexes given 0.67 or 2.0% TOTM had statistically significantly increased carnitine

acetyltransferase activity with little difference between the two sexes.

Histology:

No abnormalities were detected in themajority of the animals. The only lesions occurring with any frequency were focal interstitialpneumonitis and nephrocalcinosis in the females. The observations were not firmly dose related. Thorneumonitis was of limited extent, often only a single focus. Two female rats fed 2.0% TOTM showed reductions in sytoplasmic basophilia in liver although it was only marginal.

Electron Microscopy: In the hepatocytes from the control rats theperoxisomes varied in size from small to moderately large. They had uniformly electron dense contents and some possessed a lattice core. They were ubiquitously distributed throughout the cytoplasm. Feeding diet containing 2% TOTM produced a slight increase in the numbers of peroxisomes, which varied between cells. No difference was seen between the centrilobular and periportal areas.

## CONCLUSIONS

The NOAEL for repeated dose toxicity is considered to be 184 mg/kg and the LOAEL is Considered 650 mg/kg for both sexes.

## DATA QUALITY

Klimisch Code: 1=reliable without restrictions. Reliabilities:

Remarks field for Data Reliability:

Well conducted study, carried out by the British Industrial Biological Research Associations

# DRAFT ENV/JM/EXCH(99)13

# REFERENCES

Chemical Manufacturers Association, Project No. 3.0496. Report No. 0496/1/85

CMA Reference. TM-3.0-BT-B1B

# REPEATED DOSE TOXICITY (b)

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Daihachi Kagaku Kogyo Co., Ltd. Lot. No. N-60601

Purity: >99.0% Kept at room temperature in a dark place until use.

METHOD

Method:

Guidelines for 28-day Repeated Dose Toxicity Testing of Chemicals

(Japan)

Test type:

Repeat Dose Toxicity

GLP:

Yes

Year:

1996

Species:

Rat

Strain:

· Crj:CD(SD)

Route of administration

Doses/concentration levels: O(vehicle) 100, 300 and 1,000 mg/kg/day

Vehicle:

Com oil

Sexi

Male & Female

Exposure period:

28 days

Frequency of treatment:

Once daily

Control group and treatment: Vehicle (corn oil)

Post exposure observation period: 2 weeks for 0 and 1,000 mg/kg/day dose.

Duration of test:

Males and females; for 28 days

Statistical methods:

Bartlett's test, Dunnett's test or Kruskal-Wallis test depending on whether

or not the data were nonhomogeneous or homogeneous.

Fisher 's test for the pathological result. Jonckheere's test for the

correlation of dosage

#### REMARKS FIELD FOR TEST CONDITIONS

Test Subjects:

Age at study initiation: 6 weeks old for males and females

Weight at study initiation: 130-151g for male.

110-121g for female.

No. of animals per sex per dose: 5 Rats per sex per dose group

Study Design:

Vehicle: Com oil

Satellite groups and reasons they were added: None Clinical observations performed and frequency:

Body wt, was recorded immediately prior to the first exposure and again for each

animal every week

Hematologic parameters were evaluated for each animal. Bloodsamples for the hematologic determinations were taken fromabdominal artery in rats after 16 hr fast. Clinical chemistry analyses were performed on serum samples from each animal. Urinalyses were performed for each rat. Urine samples were collected from each rat

on the day prior to scheduled termination.

Organs examined at necropsy:

Organ weight; brain, liver, kidney, spleen, adrenal, spermary (male) and ovary

(females) for each animal.

Microscopic: heart, liver, kidneys, spleen, adrenal and bone marrow from rats in the

control and high-exposure groups and kidney from all dosage male.

### RESULTS

NOAEL

Male: >1,000 mg/kg/day Female: >1,000 mg/kg/day

#### REMARKS FIELD FOR RESULTS.

The mean body weight of treatment groups of rats for males and femalesnot Body weight:

Significantly different from controls at any time during the course of the study.

Food/water consumption: No significantly different from controls at any time during dosing and

recovering period for both sexes.

Clinical signs: No unusual clinical observations during the study.

Males:

No dose-related change in general clinical signs.

Females:

No dose-related change in general clinical signs.

Haematology:

at the end of dosing

Males and females: No dose-related significant changes inhematology.

In the blood clotting test, prothrombin times for males were slightly extended, but they were considered within the physiological change. For females, no significant

changes in all test.

after recovering period

Males:

In hematology, hemoglobin amounts for males at 1000mg/kg dosing were slightly increased, but they were considered within the physiological change. In the blood

clotting test, no significant changes in all tests.

Females:

No significant change in all tests.

Biochem:

at the end of dosing

Males:

No dose-related significant adverse treatment-related effect in clinical chemistry.

Females:

At 300, and 1,000 mg/kg dosing, chlorine contents were low.

after recovering period

Males:

At 1,000 mg/kg dosing, potassium contents were slightly high.

Females:

At 1,000 mg/kg dosing, GOT were slightly high.

But both changes were considered to be no meaning, because at the end of treatment these changes were not recognised

Urinalysis:

at the end of dosing

Males and Female: At 1,000 mg/kg dosing, some of rats (both sexes), amounts of urinary increased, but the mean urinary specific gravity values in the 1,000 mg/kg dosing group

was not significant change from control group.

after recovering period:

Males and Females: No dose-related significant change in all tests.

Mortality and time to death: No deaths prior to scheduled termination.

Organ weight changes:

at the end of dosing

Male:

No dose-related change in all testedorgans.

Female:

Relative liver weight were slightly increased at 100 mg/kg dosing, but no

dose-related change. Other organs, no significant change.

after recovering period:

Males:

At 1,000 mg/kg dosing, relative kidneyweight were slightly low.

Female:

At 1,000 mg/kg dosing, absolute and relative adrenal weight were lightly

high.

But both changes were considered no related to dosing and recovering of this chemical. Gross pathlogy and histopathlogy:

at the end of dosing:

Males:

Coloured patch/zone of lungs were observed 1 of 100 mg/kg, 2 of 300 mg/kg and 3 animals of 1,000 mg/kg dosing group. Also hypertrophy of the kidney,

hypertrophy of parathyroid, and etc. were observed.

Amounts of eosinophilic body in the kidney were slightly increased in dosing

group. But all these changes were considered no related the dosing and

recovering of this chemical, because the degree and rate of changes were same

of all the group included control.

Females:

Red patch/zone of thymus dilated lumen of the uterus and etc. were observed. But all these changes were considered no related the dosing and recovering of this chemical, because the degree and rate of changes were same of all the

group included control.

after recovering period:

Males and Females: No dose-related significant change in all tests.

#### CONCLUSIONS

No test substance related changes were noted in terms of clinical signs, body weight, food consumption, and hematology, blood chemical examination, urinally sis, and pathological

The NOEL for repeated dose toxicity is considered to be 1,000 mg/kg/day for both sexes.

# DATA QUALITY

Reliabilities:

Klimisch Code: 1=reliable without restrictions.

Remarks field for Data Reliability:

Well conducted study, carried out by the Biosafety Research Center, Food, Drugs and Pesticides (An-pyo Center), Japan

#### REFERENCES

Toxicity Testing Reports of Environmental Chemicals, vol. 4(1996) Ministry of Health & Welfare, Japan

#### TOXICITY TO REPRODUCTION

#### TEST SUBSTANCE

Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Daihachi Kagaku Kogyo Co., Ltd. Lot. No. N-80301

Purity: >99.0% Kept at room temperature in a dark place until use.

#### METHOD

Method:

OECD Preliminary reproductive toxicity screening test

Test type:

Preliminary reproduction toxicity screening test.

GLP:

Yes

Years

1998

Species:

Rat

Strain:

Crj;CD (SD)

Route of administration:

Oral (by gavage)

Doses/concentration levels: 0(vehicle) 100, 300, 1,000 mg/kg/day

Vehicle:

Corn oil

Sex:

Male & Female

Administration period:

Male; for 46 days from 2 weeks prior to mating

Female; from 2 weeks prior to mating to day 3of lactation

Frequency of treatment:

Once daily.

Control group and treatment: Vehicle (com oil)

Post exposure observation period: None.

Terminal kill

Male: day 47

Female: day 4 of lactation

Statistical methods:

Chi square test for 1 grade positive data and Fisher's test for another. Bartlett's test or Kruskal-Wallis' test for 2 or more grade positive data. And used Dunnett's test of Mann-Whitney's U-test for examination

#### REMARKS FIELD FOR TEST CONDITIONS

Test Subjects:

Age at study initiation: 10 week old for both sexes.

Weight at study initiation: 373-435 g for males, 217-257 g for females

No. of animals per sex per dose: 12 per sex per dose group

Study Design:

The animals were sacrificed on the day 4 of lactation for females. Males and females with nomated were killed 1 day after the mating period. Females with no

delivery killed 26th day of gestation period.

Vehicle: Corn oil

Satellite groups and reasons they were added: None

Mating procedures: Male/female per cage; 1/1, length of cohabitation; with in the limit of 14 days until proof of pregnancy (formation sperm detection in vagina)

was observed.

Clinical observations performed and frequency:

Parent: General appearance once a day

Foetus: General appearance once a day after birth

Organs examined at necropsy:

Parent: Males and females: Grosspathlogy of all organs were tested. Males: Organ weight: Testis and epididymis of all animals.

Female: Organ weight: Ovary of all animals.

Count: Implantation sites and corpusluteum of ovary of all animals.

Microscopic: Males: Testis and epididymis. Count of sertoli sells, spermatocytes. round spermatids and elongatespermatids in seminiferous tubules of 5animals of all dosing groups. (Stage I-VI, VII-VIII, IX-XI, XII-XIV of spermatozoon formative cycle.)

Females: Ovary

Pup: Gross pathlogy of all organs were tested. Dead pups and abnormal organs were tested histopathogy.

#### Parameters assessed during study:

Body weight. Males: Prior to the first dosing and 2, 5, 7, 10, 14 day. After that once a week, the daysacrificed. Females: Prior to the first dosing and 2, 5, 7, 10, 14 day. During gestation period, 0, 1,3, 5, 7, 10, 17 and 20 day. During lactation period, 0, 1, and 4. During cohabitation period, the same day with male. Pups: Day 0 and 4

Food/water consumption. The same day when body wt. determined except lactation period and the day sacrificed for males, also, 0 day of gestation and lactation for female.

No. of pairs with successful copulation, copulation index (No.of pairs with Successful copulation/No. of pairs mated) x 100, duration of mating No. of pregnant females, fertility index = (No.of pregnant animals/No. of pairs with successful copulation x 100, No. of corpora lutea, No. of implantation sites, implantation index (No. of implantation sites/No. of corpora lutea) x 100, No. of pups born, delivery index (No.of pups born/No. of implantation sates)x 100, No. of love pups born, live birth index (No. of love pups born/No. of pups born) x 100, sex ratio of pups, No. of dead pups born, gestation length, gestation index (No. of females with live pops delivered/ Noof pregnant females) x 100, nursing index (No. of females nursing live pups/No. of females with normal delivery) x 100, No. of live pups on day 4, viability index (No. of live pups on day 4/No. of live pups born) x 100,

#### RESULTS

Repeat dose toxicity: NOEL 100 mg/kg/day for males 1,000 mg/kg/day for female

Reproductive and developmental toxicity: NOEL100 mg/kg/day for males 1,000 mg/kg/day for female

1,000 mg/kg/day for offspring

#### REMARKS FIELD FOR RESULTS.

Mortality and day of death: None.

No statistical significant difference from controls. Body weight:

No statistical significant difference from controls. Food/water consumption:

Reproductive data:

No statistical significant difference from controls.

Pups data:

Body weight and weight gain of 300 mg/kg dosing group for both sexes were slightly low. But all pups of 100 and 1000 mg/kg dosing group were not

statistical significant difference from controls.

At the other tests, no statistical significant difference from controls.

Grossly visible abnormalities, external, soft tissue and skeletal abnormalities:

For males:

Slightly decrease of spermatocytes and spermatids: 2 animals of 300 mg/kg dosing group. 11 of 1000 mg/kg dosing group.

Moderate decrease of spermatocytes and spermatids: 1 of 1000 mg/kg/dosing group.

At this animal, a few multinucleate giant cell were appeared and slightly acuolization of sertoli sells were observed. Also, at the epididymis, moderate amount of cell debris moderate decrease of spermatids and slightly granuloms of spermatic were observed. For the control group, atrophy of seminiferous tubule were observed 2 animals. At these animals, slightly amount of cell debris were observed one of these animals, slight decrease of spermatids was also observed.

Number of cells in seminiferous tubules:

Group 1(Stage I-VI)

: Low value of spermatids at 300 mg/kg dosing group.

Low values of spermatocytes and spermatids at 1000 mg/kg dosing group. Group 2(Stage VII-VIII):Low values of round spermatids and ratio of sertoli cells at 1000 mg/kg. Group 3(stage IX-XI) :Low values of elongatespermatids and ratio of serioli cells at 1000 mg/kg. Group 4(stage XII-XIV): Low values of spermatocytes, elongatespermatids, and ratio of sertoli cells at 1000 mg/kg dosing group.

#### For females:

Cyst of corpus luteum of ovary was observed 2 animals of 300 mg/kg dosing group. No abnormal ovary observed at the female of 100 mg/kg dosing without successful copulation, females of control and 100 mg/kg dosing without pregnant.

#### Histopathological finding in rats

		dose (mg/kg)			
Items		0	100	300	1,000
No. of male animals examined		12	12	12	12
Organ: Findings					
	Grade				
Testis:					
Decrease, spermatocyte and spermatid	Total	0	O	. 2	12**
•	+	0	0	2	11
	+ +	. 0	0	0	1
Multinuclear glant cell, seminiferous tubi	ule +	0	0	0	1
Vacuolozation, Sertoli cell	• 🔸	0	0	0	1
Atrophy, seminiferous tubule	+	2	٥	0	0
Epididymis:					•
Cell debris, lumen	Total	2	0	0	. 1
	+	2	Q	0	0
	++	0	O	0	1
Decrease, sperm	Total	1	0	Q	1
Treet const about it	+	$\overline{1}$	0	0	0
	++	ā	C	0	1
Granuloma, spermatic	+	0	0	Ō	1

Ovary:  Cyst, corpus luteum  Values are no, of animals with fine				
Cyst, corpus luteum				
	<+ >	0	0 2	0
	ding.			
Grade: +=slight, ++=moderate cha		ted		
Significantly different from 0 mg/l				
Number of cells in seminiferous tubules	s of male rats.	dose (m	a/ka)	
Items	0	100	300	1,000
No. of animals examined	5	5	5	5
Group 1 (Stage I-VI)	-	•	•	•
Na. of Sertoll cells	20.12(3.18)	19.08(1.49)	18.52(1.45)	18.08(1.45)
Spermatogonia	20122(0.10)	13.00(1.43)	10120(1(40)	10.00(1.45)
No.	16.80(5.65)	20.52(2.58)	18.48(3.17)	15 76/7 61\
ratio <sup>s)</sup>	0.85(0.29)	1.08(0.19)	1.01(0.21)	15.76(2.61) 0.87(0.11)
	0.03(0.23)	1.00(0.13)	1.01(0.21)	0.07(0.11)
Spermatocyles	ደስ ወሰጣ ላላ ነ	C1 00/4 04\	42.64(2.63)	40 04/E /25#
No.	50.80(7.44 )	51.80(4.84)	• •	40.84(5.63)*
ratio	2.53(0.13)	2.72(0.26)	2.37(0.24)	2.25(0.16)
Round spermatids		100 00/2 00	44F /Aze #ALA	440 200
No.	138.36(17.20)	128.00(8.89)	117.68(5.59)*	112.60(3.11)
ratio	6.91(0.35)	6.75(0.84)	6.39(0.70)	6.26(0.48)
Elongate spermatids				
No.	130.00(21.71)	132.32(11.17)	103.28(12.34)	, ,
ratio	6.53(1.15)	6.98(0.88)	5.62(0.90)	5.30(0.69)
Group 2 (Stage VII-VIII)				•
No. of Sertoli cells	16.96(2.63)	17.04(2.17)	16.64(2.73)	16.52(2.23)
Spermatogonia				
No.	2.92(1.06)	2.40(0.93)	2.04(0.68)	2.60(1.10)
ratio	0.18(0.09)	0.14(0.05)	0.12(0.03)	0.16(0.06)
Spermatocytes	, ,			•
No.	91.68(10.37)	94.68(6.55)	84.44(6.99)	82.32(6.70)
ratio	5,45 (0.56)	5.60(0.51)	5.16(0.79)	5.03(0.54)
Round spermatids	, ,	` ,	• •	
No.	142.08(13.39)	131.64(13.72)	123.96(8.23)	118.76(8.28)
ratio	8.45(0.62)	7.75(0.39)	7.66().66)	7.25(0.62)*
Elongate spermatids	(,			(3.32)
No.	129,24(17 37)	128.32(16.88)	114,72/9,80\	105.65(13.47)
	7.78(1.54)	7.56(0.72)	7.09(1.62)	6.46(1.05)
ratio	1.10(1.04)	1400(0114)	,100(1,00)	5.40(1.00)
Group 3 (Stage VII-VIII)				48888
No. of Sertoli cells	19,28(1.92)	20.52(1.55)	19.20(1.58)	19.32(2.18)
Spermatogonia				
No.	4.52(1.32)	4.20(1.50)	4.92(1.63)	3.32(1.02)
ratio	0.23(0.05)	0.21(0.08)	0.26(0.11)	0.18(0.05)
Spermatocytes	<b>,</b> ,	• /	• •	· •,
No.	102.52(10.83)	99.08(8.42)	97.56(4.50)	89.04(9.00)
ratio	5.34(0.56)}	4.85(0.50)	5.10(0.36)	4.62(0.32)
	J. 1 (0.30))		5,22(5,25)	
Elongate spermatids	145.24(11.01)	130.64(9.90))	131.68(19.71)	119.24(15.90
No.	149:54(11:01)	*30'04(3'20))	*21.00(12.11)	ムニノ・ムマリムン・ブリ

	<del>~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~</del>			
Group 4 (Stage VII-VIII)				
No. of Sertoli cells	19.16(2.81)	20.92(1.73)	18.64(1.72)	16.72(0.92)
Spermatogonia				
No.	4.04(0.89)	3.72(0.72)	3.64(0.48)	3.64(0.71)
ratio	0.21(0.05)	0.18(0.03)	0.20(0.02)	0.22(0.05)
Spermatocytes				
No.	109.80(13.15)	110.36(9.22)	99.44(4.54)	88.76(4.33)**
ratio	5.76 (0.29)	5.28(0.12)	5.36(0.34)	5.32(0.46)
Elongate spermatids				
No.	159.76(15.91)	150.28(18.99)	137.08(17.70)	105.16(18.34)**
ratio	8.39(0.63)	7.19(0.71)	7,35(0.62)	6.33(1.31)**
TO Describe an English was the training of the				

Values are expressed as Mean(S.D.)

Significantly different from 0 mg/kg group; \*  $p \le 0.05$ , \*\*  $p \le 0.01$ 

a): (No. of spermatogenic cells/no. of sertoli cells in a seminiferous tubule)

#### Influence on reproductive performances of rats

dose (mg/kg)					
0	100	300	1,000		
12	12	12	12		
12	12	12	12		
2.1(1.2)	2.3(1.3)	2.7(1.2)	2.7(1.1)		
100.0	91.7	100.0	100.0		
11	10	12	12		
91.7	90.9	100.0	100.0		
	12 12 2.1(1.2) 100.0 11	0 100 12 12 12 12 2.1(1.2) 2.3(1.3) 100.0 91.7 11 10	12 12 12 12 12 12 2.1(1.2) 2.3(1.3) 2.7(1.2) 100.0 91.7 100.0 11 10 12		

<sup>\*(</sup>No.of pairs with successful copulation/no.of pairs mated) x 100

# Influence on developmental performances of rats

		dose (mg/kg)							
Items	. <b>0</b> ),	100	300	1,000					
No. of male animals examined	12	12	12	12					
No. of corpora lutea	16.8(1.5)	17.3(1.3)	17.0(2.3)	17.9(2.2)					
No. of implantation sites	15.5(1.7)	16.6(1.3)	16.0(2.0)	16.3(2.3)					
Implantation index(%)	92.5(7.2)	96.2(6.6)	94.5(8.4)	91.3(8.8)					
No. of pups born(%)	13.7(3.1)	15.0(1.7)	15.0(1.8)	15.1(2.7)					
Delivery index(%) b	87.6(15.4)	90.3(6.8)	94.1(7.2)	92.2(9.6)					
Live pups born									
No.	13.3(2.9)	14.7(2.0)	14.9(2.0)	15.0(2.7)					
Live birth index(%)	97.1(5.6)	97.8(3.6)	99.2(2.6)	99.4(2.1)					
Sex ratio(M/F)	1.09(0.69)	1.05(0.50)	1.17(0.75)	0.76(0.44)					
Dead pups born									
No.	0.5(0.9)	0.3(0.5)	0.1(0.3)	0.1(0.3)					
Gestation length(day)	22.7(0.5)	22.7(0.5)	22.5(0.5)	11.6(0,5)					
Gestation index(%)	100.0	100.0	100.0	100.0					
Nursing index(%)	100.0	100.0	100.0	100.0					
Live pups on day 4									
No.	13.2(2.8)	14.6(2.1)	14.4(2.9)	14.5(2.9)					
Viability index(%) 6	99.5(1.8)	99.3(2.3)	95.6(11.5)	96.7(6.7)					
Body weight of pups(g)									
Male Day 0	7.32(0.77)	7.13(0.52)	6.69(0.55)	6.87(0.84)					
Day 4	11.71(1.76)	11.09(0.93)	10.23(0.98)*	•					
Day 0-4, gain(g)	4.39(1.04)	3.96(0.53)	3.54(0.77)*	3.73(0.80)					
Body weight gain(%) "	59.41(8.87)	55,54(6.16)	53.19(11.91)	54.39(9.50)					

<sup>\*\*(</sup>No. of pregnant animals/no.of pairs with successful copulation) x 100

Female	Day 0	6.93(0.83)	6.63(0.64)	6.33(0.58)	6.58(0.62)	
	Day 4	11.08(1.71)	10.28(1.01)	9.84(1.01)*	10.03(1.46)	
	Day 0-4, gain(g)	4.16(1.00)	3.65(0.56)	3.14(0.79)*	3.46(0.96)	
	Body weight gain(%)	59.63(10.42)	55.24(8.07)	49.95(13.09)	52.17(11.10)	

Values are expressed as Mean (S.D.)

Significantly difference from 0 mg/kg group;  $p \le 0.05$ 

- a): (No. of implantation sites/no. of corpora lutea) x 100
- b): (No. of pups born/no. of implantation sites) x 100
- c): (No. of live pups born/no. of pups born) x 100
- d): (No. of females with live pups delivered/ no. of pregnant remales) x 100
- e): (No. of females nursing live pups/no. of females with normal delivery) x 100
- f): (No. of live pups on day 4/ no. of live pups born) x 100
- g): (Body weight gain/body weight on day 0) x 100

#### **CONCLUSIONS**

#### Repeat dose toxicity

Histopathological examination of the testes, demonstrated decrease of permatocytes and spermatids in males of the 300 and 1000 mg/kg group. No effects of this chemical on general appearance, body weight, food consumption, autopsy findings, weights of the reproductive organs of both sexes, or histopathlogical features of the ovary were detected.

The NOELs are considered to be 100 mg/kg/day for males, and 1,000 mg/kg/day for females.

#### Reproductive and developmental toxicity

Except for the effects in males observed onhistopathological examination, no influence of this chemical was detected regarding reproductive ability, organ weight ohistopathological feature of the ovary, delivery or maternal behaviour of dams. No effects of this chemical were detected oniability, general appearance, body weights or autopsy findings for offspring.

The NOELs are considered to be 100 mg/kg/day for males, 1,000 mg/kg/day for females, and 1,000 mg/kg/day for offspring.

# DATA QUALITY

- Reliabilities: Klimisch Code: 1=reliable without restrictions.
- Remarks field for Data Reliability:

Well conducted study, carried out by the Safety Research Institute for Chemical Compounds Co., Ltd.(Japan)

#### REFERENCES

Toxicity Testing Reports of Environmental Chemicals, vol. 6(1998)

Ministry of Health & Welfare, Japan

#### GENERAL REMARKS

#### GENETIC TOXICITY IN VITRO (BACTERIAL TEST)

#### TEST SUBSTANCE

· Identity:

Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate

Remarks:

Source: Daihachi Kagaku Kogyo Co., Ltd. Lot. No. N-60601

Purity: >99.0% Kept at room temperature in a dark place until use.

**METHOD** 

Method:

Guideline for ScreeningMutagenicity Testing of Chemicals(Japan) and

OECD TG 471 and 472

Test type:

Reverse mutation assay

• GLP:

Yes

Year:
Species/Strain:

1996

Salmonella typhimurium TA100, TA1535, TA98, TA1537

Escherichia coli WP2 uvrA

Positive controls:

-S9 mix, 2-(2-Furyl)-3-(5-nitro-2-furyl)acrylamide (TA100, WP2, TA98)

Sodium azide (TA1535) 9-Aminoacridine (TA 1537)

+S9 mix, 20Aminoanthracene (five strains)

S9:

Rat liver, induced with phenobarbital and 5,6-benzoflavone

Statistical methods

No statistical analysis was done.

#### REMARKS FIELD FOR TEST CONDITIONS

Study Design:

Concentration: -S9: 0, 313, 625, 1,250, 2,500, 5,000 ug/plate (five strains) +S9: 0, 313, 625, 1,250, 2,500, 5,000 ug/plate (five strains)

Number of replicates: 2

Plates/test: 3

Procedure: Plate incorporation method

Solvent: Acetone Positive controls:

-S9 mix, 2-(2-Furyl)-3-(5-nitro-2-furyl)acrylamide (TA100, WP2, TA98)

Sodium azide (TA1535)
9-Aminoacridine (TA 1537)

+S9 mix, 20Aminoanthracene (five strains)

#### RESULTS

Cytotoxic concentration:

Toxicity was not observed up to 5,000 ug/plate in five strains with and without metabolic activation (S9 mix).

Genotoxic effects:
+ ? -
With metabolic activation: [ ] [ x ]
Without metabolic activation: [ ] [ x ]
REMARKS FIELD FOR RESULTS.
CONCLUSIONS
Bacterial gene mutation is negative with and without metabolic activation.
DATA QUALITY
Reliabilities: Valid without restriction.
Remarks field for Data Reliability
Well conducted study, carried out by Halano Research Institute, Food and Drug Safety Center (Hadano, Japan).
REFERENCES
Toxicity Testing Reports of Environmental Chemicals,vol.4(1996)
Ministry of Health & Welfare, Japan
GENERAL REMARKS

# GENETIC TOXICITY IN VITRO (NON-BACTERIAL IN VITRO TEST)

TES	ST SUBSTANCE	
	Identity:	Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate
4	Remarks:	Source: Daihachi Kagaku Kogyo Co., Ltd. Lot. No. N-60601 Purity: >99.0% Kept at room temperature in a dark place until use
ME	THOD	
•	Method:	Guideline for Screening Toxicity Testing of Chemicals (Japan)
•	Test type:	Chromosomal aberration test
•	GLP.	Yes .
•	Year:	1996
•	Species/Strain:	CHL/IU cell
•	Metabolic activation:	with and without S9 from rat liver, induced withphenobarbital and 5,6-benzoflavone.
•	Statistical methods	Fisher's exact analysis
RE	MARKS FIELD FOR TES	ST CONDITIONS
•	Study Design:	
		For continuous treatment, cells were treated for 24 or 48 hrs without S9.
		For short-term treatment, cells were treated for 6 hrs with and without S9 and cultivated with fresh media for 18 hrs.
		Concentration: -S9 (continuous treatment): 0, 1.3, 2.5, 5.0 mg/mlS9 (short-term treatment): 0, 1.3, 2.5, 5.0 mg/ml.
		+S9 (short-term treatment): 0, 1.3, 2.5, 5.0 mg/mL
		Plates/test: 2
		Solvent: Acetone
		Positive controls: Mitomycin C for continuous treatment
		Cyclophosphamide for short-term treatment
RE	SULTS	
•	Cytotoxic concentration	n:
	Toxicity was not o without S9 mix.	bserved up to 5.0 mg/ml in continuous and short-term treatment with or
•	Genotoxic effects:	Clastogenicity polyploidy
1		+ ? - + ? -
	With metabolic Without metaboli	
47.4	CMARKS FIELD FOR RE	יכווז דכ
	CIVERSON BERLEITEN DE NO.	.71141415.

REMARKS FIELD FOR RESULTS.

# GENETIC TOXICITY IN VITRO (NON-BACTERIAL IN VITRO TEST)

TES	T SUBSTANCE	
•	Identity:	Tris(2-ethylhexyl)benzene-1,2,4-tricarboxylate
	Remarks:	Source: Daihachi Kagaku Kogyo Co., Ltd. Lot. No. N-60601
		Purity: >99.0% Kept at room temperature in a dark place until use
ME	тнор	
•	Method:	Guideline for Screening Toxicity Testing of Chemicals (Japan)
•	Test type:	Chromosomal aberration test
•	GLP:	Yes ·
•	Year:	1996
•	Species/Strain:	CHL/IU cell
•	Metabolic activation:	with and without S9 from rat liver, induced withphenobarbital and 5,6-benzoflavone.
•	Statistical methods	Fisher's exact analysis
RE	MARKS FIELD FOR TES	ST CONDITIONS
ļ		
•	Study Design:	The second control of the second for 76 at 49 him without 60
		For continuous treatment, cells were treated for 24 or 48 hrs without S9.  For short-term treatment, cells were treated for 6 hrs with and without S9.
		and cultivated with fresh media for 18 hrs.
	•	Concentration: -S9 (continuous treatment): 0, 1.3, 2.5, 5.0 mg/ml.
		-S9 (short-term treatment): 0, 1.3, 2.5, 5.0 mg/ml.
		+S9 (short-term treatment): 0, 1.3, 2.5, 5.0 mg/ml.
		Plates/test: 2
		Solvent: Acetone
-		Positive controls: Mitomycin C for continuous treatment  Cyclophosphamide for short-term treatment
		Cyclophospinames is such activities
RE	SULTS	
•	Cytotoxic concentration	1:
	Toxicity was not o without S9 mix.	bserved up to 5.0 mg/ml in continuous and short-term treatment with or
	Genotoxic effects:	Clastogenicity polyploidy
		+ 3 - + 3 -
	With metabolic	activation: [][][x][][x]
	Without metabolic	
1		

Appendix I Parameters used in caluculation of distribution by Mackay level III fugacity model.

# Physico-chemical Parameter for TOTM

			M
molecu	lar weight	546.79	Measured.
melting	point [°C]	-50	Measured
vapor pr	essure [Pa]	2.80E-04	Estimated
water solu	bility [g/m²]	0.13	Measured
log Kow		5.94	Measured
	in air	12	Estimated
half life [h]	in water	288	Estimated
in soil		288	Estimated
	in sediment	864	Estimated

Temp. [°C]

25

#### Environmental Parameter

		Volume	depth	area	organic	lipid content	density	residence
		[m³]	[m]	[m³]	carbon[-]	[-]	[kg/m³]	time [h]
	air	1.0E+13					1.2	100
bulk air	particles	2.0E+03						
	total	1.0E+13	1000	1E+10				
	water	2.0E+10					1000	1000
bulk water	particles	1.0E+06			0.04		1500	
[	Fish	2.0E+05				0.05	1000	
	Total	2.0E+10	10	2E+09				
	Дir	3.2E+08					1.2	
bulk soil	Water	4.8E+08					1000	
	Solid	8.0E+08			0.04		2400	
	Total	1.6E+09	0.2	8E+09				
bulk	Water	8.0E+07					1000	
sediment	Solid	2.0E+07			0.06		2400	50000
	Total	1.0E+08	0.05	2E+09				

# Intermadea Transport Parameter (m/h)

air side air-water MTC	5	soil air boundary layer MTC	5
water side air water MTC	0.05	sediment-water MTC	1E-04
rain rate	1E-04	sediment deposition	5E-07
aerosol deposition	6E-10	sedimentresuspension	2E-07
soil air phase diffusion MTC	0.02	soil water runoff	5E-05
soil water phase diffusion MTC	1 <b>E</b> -05	soil solid runoff	1E-08

# Theoretical Distribution of TOTM

### scenario 1

	emission rate	conc.	amount	percent	Transformation rate [kg/h]	
	[kg/h]	[g/m³]	[kg]	[%]	Reaction	advection
Air	1,000	1.3.E-07	1.3.E+04	19.6	7.5E+02	1.3.E+02
Water	0	1.6.E-05	3.10.E+03	4.7	7.6E+00	3.1.E+00
Soil	0	2.5.E-03	4.4.E+04	66.2	1.1E+02	•
Sediment	•	1.3.E-02	6.3.E+03	9.5	5.1E+00	1.2.E-01
		total amount	6.7.E+04			

### scenario 2

	Emission rate	conc.	Amount	percent [%]	Transformation rate [kg/h]	
•	[kg/h]	[g/m³]	[kg]		Reaction	advection
air	0	1.8.E-09	1.8.E+02	0.0	1.0.E+01	1.8.E+00
water	1000	9.7.E-04	1.9.E+05	32.7	4.7.E+02	1.9.E+02
soil	0	3.4.E-05	6.2.E+02	0.1	1.5.E+00	
sediment		7.9.E-01	3.9.E+05	67.2	3.2.E+02	7.9.E+00
		total amount	5.9.E+05		-	

# DRAFT ENV/JM/EXCH(99)13

# scenario 3

	emission rate	conc.	Amount	percent [%]	Transformation rate [kg/h]	
	[kg/h]	[g/m³]	[kg]		Reaction	advection
аіг	0	7.0.E-13	7.0.E-02	0.0	4.1.E-03	7.0.E-04
water	0	5.2.E-08	1.0.E+01	0.0	2.5.E-02	1.0.E-02
soil	1000	2.3.E-02	4.2.E+05	100.0	1.0.E+03	
sediment		4.2.E-05	2.1.E+01	0.0	1.7.E-02	4.2.E-04
		total amount	4.2.E+05			

scenario 4

	emission rate	conc.	Amount	percent [%]	Transformation rate [kg/h]	
	[kg/h]	[g/m³]	[kg]		Reaction	advection
air	600	7.8.E-08	7.8.E+03	3.0	4.5.E+02	7.8.E+01
water	300	3.0.E-04	6.0.E+04	23.5	1.5.E+02	6.0. <b>E</b> +01
soil	100	3.8.E-03	6.8.E+04	26.6	1.6.E+02	
sediment		2.4.E+01	1.2.E+05	46.9	9.8.E+01	2.4.E+00
		total amount	2.6.E+05			

# IUCLID

# **Data Set**

: 1,2,4-Benzenetricarboxylic acid, decyl octyl ester

**Existing Chemical** 

CAS No.

EINECS Name

EINECS No.

ISCA Name

TSCA Name

: 1,2,4-Benzenetricarboxylic acid, decyl octyl ester

Molecular Formula : C10H22O.xC9H6O6.xC8H18O

**Producer Related Part** 

Company Creation date : ExxonMobil Biomedical Sciences Inc.

: 02.11.2000

: ID: 67989-23-5

: 67989-23-5

: 268-007-3

**Substance Related Part** 

Company Creation date : ExxonMobil Biomedical Sciences Inc.

: 02.11.2000

Memo

: ACC Phthalate Esters Panel HPV Testing Group

Printing date Revision date

•

: 11

Date of last Update

: 30.10.2001

: 13.12.2001

Number of Pages

Chapter (profile) : Reliability (profile) :

Flags (profile)

#### 1. General Information

ld 67989-23-5 Date 13.12.2001

#### 1.0.1 OECD AND COMPANY INFORMATION

Type

: lead organisation

Name

ACC Phthalate Esters Panel HPV Testing Group

Partner

Dr. Marian Stanley

Date

:

Street Town : 1300 Wilson Blvd.: 22209 Arlington, VA

Country Phone : United States : (703) 741-5623 : (703) 741-6091

Telefax Telex

:

Cedex Remark

The American Chemistry Council Phthalate Esters Panel sponsoring this

test plan includes the following member companies:

Eastman Chemical Company
ExxonMobil Chemical Company

Sunoco Chemicals Teknor Apex Company

Flag

26.10.2001

Critical study for SIDS endpoint

#### I.1. GENERAL SUBSTANCE INFORMATION

Substance type Physical status : organic : liquid

Physica Purity

.% w/w

09.10.2001

#### 1.1.0 DETAILS ON TEMPLATE

Comment

This chemical is part of the Trimellitate category. The category includes the following four CAS numbers: 3319-31-1, 27251-75-8, 53894-23-8 and

67989-23-5.

Remark

**DESCRIPTION OF THE TRIMELLITATES CATEGORY** 

The trimellitates comprise a family of chemicals synthesized by esterifying trimellitic anhydride with alcohols with average carbon numbers ranging from approximately C7-C10, in the presence of an acid catalyst. The category includes the four trimellitates: 3319-31-1 (TOTM), 27251-75-8

(TIOTM),

53894-23-8 (TINTM), and 67989-23-5 (DOTM). Trimellitates in this category are all 1,2,4-benzenetricarboxylic acids with side chain ester groups ranging from C8 to C10. The structural formula for trimellitates varies somewhat depending on the isomeric composition of the alcohols used in their manufacture. The specific alcohols used are 2-ethylhexanol (TOTM), iso-octyl alcohol (TIOTM), iso-nonyl alcohol (TINTM), and a mixture of linear and branched decyl (40%) and octyl (60%) alcohols (DOTM).

Trimellitates are colorless to slightly yellow liquids with high boiling points (> 250oC) and low vapor pressures; properties which contribute to their high physical stability. They are readily soluble in most organic solvents and miscible with alcohol, ether and most oils, but essentially insoluble in water. Because of the similarity in structure as well as physicochemical

#### 1. General Information

ld 67989-23-5 Date 13.12.2001

properties, the trimellitates were grouped into a single category containing four substances with carboxylic side chain ester groups ranging from C8-

C10.

Flag 09.10.2001 Critical study for SIDS endpoint

#### 1.7 USE PATTERN

Type Category Remark : industrial

Polymers industry

: Trimellitates are used predominantly as plasticizers for production of flexible PVC. Because of their relatively high molecular weight (>500 g/mole) and bulky structure, they have lower volatility and greater resistance to migration than the corresponding phthalate ester plasticizers. They are predominantly used in the manufacture of high temperature PVC cables (Wilson, 1996). Since these chemicals are produced in closed systems, there is essentially no occupational exposure to these

substances except at the flexible PVC production facility. Usually, these substances have been already blended to the compound as plasticizer, so it is not expected that downstream users or consumers are directly

exposed to trimellitates.

Flag 13.12.2001 : Critical study for SIDS endpoint

(3)

ld 67989-23-5 Date 13.12.2001

#### 2.1 **MELTING POINT**

Value

234 ° C

Decomposition

no at °C

Sublimation Method

no other

Year

2000

GLP

Test substance Method

Melting point calculation by MPBPWIN ver. 1.40 using calculation methods

of Joback and Gold and Ogle.

Remark

: EPIWIN is used and advocated by the US EPA for chemical property

estimation

Melting point calculation seems to give erroneously high results for the

thhis class of chemicals.

Source

: ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability

: (3) invalid

10.10.2001

(2)

#### 2.2 BOILING POINT

Value

585 °C at 1013 hPa

Decomposition

no other

Method Year

: 2000

GLP

Test substance

Method

Boiling point calculation by MPBPWIN ver. 1.40 using calculation methods

of Stein and Brown.

Remark

: EPIWIN is used and advocated by the US EPA for chemical property

estimation

Source

: ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability

(2) valid with restrictions

10.10.2001

(2)

#### 2.4 VAPOUR PRESSURE

Value

.0000000000014 hPa at 25° C

Decomposition

Method

other (calculated)

Year

2000

**GLP** 

Test substance Decomposition

Method

Vapor pressure calculation by MPBPWIN ver. 1.40 using calculation

method of Grain.

Remark

: EPIWIN is used and advocated by the US EPA for chemical property

estimation

Source

: ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability 15.10.2001 : (2) valid with restrictions

(2)

ld 67989-23-5 Date 13.12.2001

#### **PARTITION COEFFICIENT** 2.5

Log pow

12.79 at 25° C

Method

other (calculated)

Year

2000

**GLP** 

Test substance

Method

Partition coefficient by LOGKOWWIN ver. 1.65 using an atom/fragment

calculation method of Mevlan and Howard.

Remark

: EPIWIN is used and advocated by the US EPA for chemical property

estimation

Source

ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability

20.12.2000

: (2) valid with restrictions

#### 2.6.1 WATER SOLUBILITY

Value

2.78 other: ng/l at 25 ° C

Qualitative

at 25 ° C

Pka at and °C PH

Method other 2000 Year

**GLP** 

**Test substance** 

Method

Water solubility calculated using WSKOWWIN ver 1.36 based on Kow

correlation method of Meylan and Howard

EPIWIN is used and advocated by the US EPA for chemical property Remark

estimation

ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA Source

Reliability 10.10.2001 : (2) valid with restrictions

(2)

(2)

ld 67989-23-5 Date 13.12.2001

#### 3.1.1 PHOTODEGRADATION

Type

Light source : Sun light Light spect. : nm

Rel. intensity : 1 based on Intensity of Sunlight

: air

Conc. of subst. : at 25 degree C

Indirect photolysis

Sensitizer : OH

Conc. of sens. : 1500000 molecule/cm3

Rate constant : .0000000000335 cm3/(molecule\*sec)

Degradation : % after
Deg. Product : not measured
Method : other (calculated)

Year : 2000

GLP Test substance

Method : Photodegradation rate calculated by AOPWIN ver. 1.89 based on the

methods of Atkinson.

Remark : EPIWIN is used and advocated by the US EPA for chemical property

estimation

Source : ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability : (2) valid with restrictions

10.10.2001 (2)

#### 3.1.2 STABILITY IN WATER

Type : abiotic

t1/2 pH4 : at degree C

t1/2 pH7 : 1 year at 25 degree C t1/2 pH9 : at degree C

Deg. Product : not measured
Method : other (calculated)

Year : 2000

GLP

Test substance

Method : Hydrolysis rate calculated by HYDROWIN ver. 1.67 based on work for EPA

by T. Mill et al.

Remark : EPIWIN is used and advocated by the US EPA for chemical property

estimation

Source : ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability : (2) valid with restrictions

20.12.2000 (2)

#### 3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

Type : fugacity model level I

Media : other
Air (level I) : 0
Water (level I) : 0
Soil (level I) : 97.8

Biota (level II / III)

Soil (level II / III)

Method : other Year : 2000

ld 67989-23-5 Date 13.12.2001

Source

: ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability 15.10.2001 : (2) valid with restrictions

(1)

#### 3.3.2 DISTRIBUTION

Media

: air - biota - sediment(s) - soil - water

Method

: other (calculation)

Year

: 2000

Result

: Soil - 97.8%

Air - 0.00000102%

Water - 0.000000172% Sediment - 2.17%

Suspended sed. - 0.068%

Source

: ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability

: (2) valid with restrictions

20.12.2000

(1)

4.	Ec	oto	xic	itv
		~		

ld 67989-23-5 **Date** 13.12.2001

- 4.1 ACUTE/PROLONGED TOXICITY TO FISH
- 4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES
- 4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

5.1.1	ACUTE ORAL TOXICITY
5.1.2	ACUTE INHALATION TOXICITY
5.1.3	ACUTE DERMAL TOXICITY
5.4	REPEATED DOSE TOXICITY
5,5	GENETIC TOXICITY 'IN VITRO'
5.8	TOXICITY TO REPRODUCTION
5.9	DEVELOPMENTAL TOXICITY/TERATOGENICITY

5. Toxicity

**ld** 67989-23-5

Date 13.12.2001

### 6. References

ld 67989-23-5 Date 13.12.2001

(1) Mackay, D., A. DiGuardo, S. Paterson and C. Cowan, EQC Model ver. 1.01, 1997, available form the Emvironmental Centre, Trent Univ. Canada.

- (2) Meylan, M. Syracuse Research Corporation (1994-1999) Calculation program contained in EPIWIN (Esitmate ver. 3.04) available from SRC.
- (3) Wilson, A., (1996). Plasticizers Selection, Applications and Implications. Rapra Review Reports 8:15-16.

#### 7. Risk Assessment

ld 67989-23-5 Date 13.12.2001

#### 7.1 END POINT SUMMARY

#### 7.2 HAZARD SUMMARY

Chapter Remark : Chapters 4 & 5

Because of the similarity in chemical structure, the Panel believes that the toxicological properties of the substances in this category will be similar as well. Thus, the Panel considers that the data for the best tested member of this category, tris-2-(ethylhexyl) trimellitate (TOTM), also represents the potential for human and environmental effects of the other members of this category.

TOTM has been sponsored under the OECD SIDS program through ICCA. A review of the available data for TOTM (see attached Table) indicates that all endpoints have been adequately addressed, and that TOTM exhibits a low order of toxicity.

Due to their higher molecular weight and bulky side chains, the remaining members of this category are expected to demonstrate a lower order of toxicity than TOTM. This is supported by a similar structural-activity relationship observed with phthalate ester compounds, i.e., the higher molecular weight phthalates (ester side chains >C7) are less active that the transitional phthalates (ester side chains C4-C6). Thus, the use of TOTM to represent the potential hazards of the other category members is a conservative position.

Attached doc.

Flag

13.12.2001

Summary of SIDS Information on Trimellitates.doc

Critical study for SIDS endpoint

7.3 RISK ASSESSMENT

# IUCLID

# **Data Set**

**Existing Chemical** 

CAS No.

: ID: 3319-31-1

**EINECS Name** 

: 3319-31-1

: tris(2-ethylhexyl) benzene-1,2,4-tricarboxylate

EINECS No.

: 222-020-0

TSCA Name

: 1,2,4-Benzenetricarboxylic acid, tris(2-ethylhexyl) ester

Molecular Formula

: C33H54O6

**Producer Related Part** 

Company

: ExxonMobil Biomedical Sciences Inc.

Creation date

: 02.11.2000

**Substance Related Part** 

Company

: ExxonMobil Biomedical Sciences Inc.

Creation date

: 02.11.2000

Memo

: ACC Phthalate Esters Panel HPV Testing Group

Printing date

: 13.12.2001

Revision date

**Date of last Update** 

: 13.12.2001

**Number of Pages** 

: 16

#### 1. General Information

ld 3319-31-1 Date 13.12.2001

#### 1.0.1 OECD AND COMPANY INFORMATION

Type

: lead organisation

Name

ACC Phthalate Esters Panel HPV Testing Group

Partner

: Dr. Marian Stanley

Date

Street Town

: 1300 Wilson Blvd. : 22209 Arlington, VA

Country Phone

: United States

Telefax

: (703) 741-5623 (703) 741-6091

Telex

Cedex

Remark

The American Chemistry Council Phthalate Esters Panel sponsoring this

test plan includes the following member companies:

Eastman Chemical Company ExxonMobil Chemical Company

Sunoco Chemicals **Teknor Apex Company** 

Flag

26.10.2001

: Critical study for SIDS endpoint

#### 1.1 GENERAL SUBSTANCE INFORMATION

Substance type Physical status

: organic : liquid

Purity

% w/w

09.10.2001

#### 1.10 DETAILS ON TEMPLATE

Comment

This chemical is part of the Trimellitate category. The category includes the following four CAS numbers: 3319-31-1, 27251-75-8, 53894-23-8 and

67989-23-5.

Remark

DESCRIPTION OF THE TRIMELLITATES CATEGORY

The trimellitates comprise a family of chemicals synthesized by esterifying trimellitic anhydride with alcohols with average carbon numbers ranging from approximately C7-C10, in the presence of an acid catalyst. The category includes the four trimellitates: 3319-31-1 (TOTM), 27251-75-8 (TIOTM), 53894-23-8 (TINTM), and 67989-23-5 (DOTM). Trimellitates in this category are all 1,2,4-benzenetricarboxylic acids with side chain ester groups ranging from C8 to C10. The structural formula for trimellitates varies somewhat depending on the isomeric composition of the alcohols used in their manufacture. The specific alcohols used are 2-ethylhexanol (TOTM), iso-octyl alcohol (TIOTM), iso-nonyl alcohol (TINTM), and a mixture of linear and branched decyl (40%) and octyl (60%) alcohols (DOTM).

Trimellitates are colorless to slightly yellow liquids with high boiling points (> 250oC) and low vapor pressures; properties which contribute to their high physical stability. They are readily soluble in most organic solvents and miscible with alcohol, ether and most oils, but essentially insoluble in water. Because of the similarity in structure as well as physicochemical

### 1. General Information

ld 3319-31-1 Date 13.12.2001

properties, the trimellitates were grouped into a single category containing four substances with carboxylic side chain ester groups ranging from C8-

Flag 09.10.2001 Critical study for SIDS endpoint

#### 1.7 USE PATTERN

Type Category Remark

: industrial

: Polymers industry

: Trimellitates are used predominantly as plasticizers for production of flexible PVC. Because of their relatively high molecular weight (>500 g/mole) and bulky structure, they have lower volatility and greater resistance to migration than the corresponding phthalate ester plasticizers. They are predominantly used in the manufacture of high temperature PVC cables (Wilson, 1996). Since these chemicals are produced in closed systems, there is essentially no occupational exposure to these substances except at the flexible PVC production facility. Usually, these substances have been already blended to the compound as plasticizer, so

it is not expected that downstream users or consumers are directly

exposed to trimellitates.

Flag 13.12.2001 : Critical study for SIDS endpoint

(9)

**Id** 3319-31-1 Date 13.12.2001

#### **MELTING POINT** 2.1

Value

: -46 ° C

Remark Source

: pour point : ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability

: (4) not assignable

20.12.2000

(7)

Value Decomposition

97 ° C : no at °C

Sublimation Method

: no : other : 2000

Year GLP

Test substance

Method

Melting point calculation by MPBPWIN ver. 1.40 using calculation methods

of Joback and Gold and Ogle.

Remark

: EPIWIN is used and advocated by the US EPA for chemical property

Melting point calculation seems to give erroneously high results for this

class of chemicals.

Source

: ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability

: (3) invalid

16.10.2001

(8)

#### 2.2 BOILING POINT

Value

541 °C at 1013 hPa

Decomposition Method

: no : other : 2000

Year

**GLP** 

**Test substance** 

: Boiling point calculation by MPBPWIN ver. 1.40 using calculation method

of Stein and Brown.

Remark

Method

: EPIWIN is used and advocated by the US EPA for chemical property

estimation

Source

: ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability

: (2) valid with restrictions

16.10.2001

(8)

#### 2.4 VAPOUR PRESSURE

Value

.0000000000525 hPa at 25° C

Decomposition

: no

Method

other (calculated)

Year

: 2000

**GLP** 

Test substance

Decomposition

: no

Method

: Vapor pressure calculation by MPBPWIN ver. 1.40 using calculation

method of Grain.

Remark

: EPIWIN is used and advocated by the US EPA for chemical property

estimation

Source

: ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability

: (2) valid with restrictions

4/16

ld 3319-31-1 Date 13.12.2001

16.10.2001

(8)

Value

: .133 hPa at 200° C

Source

: ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability

: (4) not assignable

16.10.2001

(7)

#### **PARTITION COEFFICIENT**

Log pow

4.35 at 25° C

Method Year

other (measured) 1984

**GLP** 

: yes

Test substance

as prescribed by 1.1 - 1.4

Remark

The study was conducted following the methods outlined in the ABC protocol # A-8003 (revised 6 August, 1984) for CMA Environmental Effects Testing Program with TOTM. 0.4% solutions of TOTM (supplied by CMA) were prepared in n-octanol and 40 ml portions were shaken for 24 hours with 400 ml water. After a 48 hour settling period, aliquots from both phases were drawn to analyse their TOTM concentrations using GC or

HPLC.

Source

: Internaltional Speciality Chemicals Ltd. Hythe

FMC Corporation Manchester.

16.10.2001

Log pow

: 5.94 at 25° C

Method

OECD Guide-line 107 "Partition Coefficient (n-octanol/water), Flask-

shaking Method"

Year GLP : 2000 : ves

Test substance

: as prescribed by 1.1 - 1.4

Source

: Chemicals Evaluation and Research Institute, Japan Ministry of

International Trade and Industry (1998)

Reliability

16.10.2001

: (2) valid with restrictions

Log pow

11.59 at 25° C

Method Year

other (calculated)

2000

**GLP** 

Test substance

Method

Partition coefficient by LOGKOWWIN ver. 1.65 using an atom/fragment

calculation method of Meylan and Howard.

Remark

: EPIWIN is used and advocated by the US EPA for chemical property

estimation.

Source

: ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability

: (2) valid with restrictions

30.10.2001

#### 2.6.1 WATER SOLUBILITY

Value

.00005 other: ug/L at 25 ° C

Qualitative

at 25 ° C :

Pka PH

at and °C

Method Year

: other

2000

**GLP** 

5/16

ld 3319-31-1 Date 13.12.2001

Test substance

Method

Water solubility calculated using WSKOWWIN ver. 1.36 based on Kow

correlation method of Meylan and Howard.

Remark

: EPIWIN is used and advocated by the US EPA for chemical property

Source

: ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA : (2) valid with restrictions

Reliability

30.10.2001

Value Qualitative

.00039 mg/l at 25 ° C of very low solubility

Pka

at 25 ° C at and °C

PH Method

OECD Guide-line 105 "Water Solubility"

Year **GLP** 

yes

**Test substance** 

as prescribed by 1.1 - 1.4

Source

Chemicals Evaluation and Research Institute, Japan Ministry of International Trade and Industry (1998)

16.10.2001

ld 3319-31-1 Date 13.12.2001

#### 3.1.1 PHOTODEGRADATION

Type : air

Light source : Sun light Light spect. : nm

Rel. intensity : 1 based on Intensity of Sunlight

Conc. of subst. : at 25 degree C

Indirect photolysis

Sensitizer : OH

Conc. of sens. : 1500000 molecule/cm3

Rate constant : .0000000003277 cm3/(molecule\*sec)

Degradation : % after

Deg. Product :

Method : other (calculated)

Year : 2000

GLP

Test substance

Method : Photodegradation rate calculated by AOPWIN ver. 1.89 based on the

methods of Atkinson.

Remark : EPIWIN is used and advocated by the US EPA for chemical property

estimation

Source : ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability : (2) valid with restrictions

20.12.2000 (8)

#### 3.12 STABILITY IN WATER

Type : abiotic

t1/2 pH4 : at degree C

t1/2 pH7 : .3 year at 25 degree C

t1/2 pH9 : at degree C
Deg. Product : not measured
Method : other (calculated)

Year : 2000

GLP

Test substance

Method : Hydrolysis rate calculated by HYDROWIN ver. 1.67 based on work for EPA

by T. Mill et al.

Remark : EPIWIN is used and advocated by the US EPA for chemical property

estimation

Source : ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability : (2) valid with restrictions

20.12.2000 (8)

#### 3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

Type : fugacity model level I

 Media
 : other

 Air (level I)
 : 0

 Water (level I)
 : 0

 Soil (level I)
 : 97.8

Biota (level II / III)
Soil (level II / III)

Method : other Year : 2000

7/16

ld 3319-31-1 Date 13.12.2001

Source

ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability

: (2) valid with restrictions

20.12.2000

(6)

#### 3.3.2 DISTRIBUTION

Media Method

: air - biota - sediment(s) - soil - water : Calculation according Mackay, Level I

Year

: 2000

Result

: Soil - 97.8%

Air - 0.00000364% Water - 0.000000284% Sediment - 2.17%

Suspended sed. - 0.068%

Source

: ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability

: (2) valid with restrictions

20.12.2000

(6)

#### 3.5 BIODEGRADATION

Type

: aerobic

Inoculum

: domestic sewage

Concentration

: 10mg/l related to Test substance

related to

Contact time

: 28 day

Degradation

ca. 68.3 - 71.1 % after readily biodegradable

Result Deg. Product

Method

other 1985

Year **GLP** 

yes

Test substance

as prescribed by 1.1 - 1.4

Method

: Method/Guideline-USEPA 1982, CO2 Evolution, Shake Flask.

Domestic sewage, mixed liquor.

Kinetics-Not Reported

**Degradation Products-Not Reported** 

Analytical Monitoring-No

Result

: The results of the first and third test are reported (68.3 and 71.1%

biodegrdation respectively).

Test condition

: Inoculum consisted of deionized water, mineral stocks, native soil, aerated mixed liquor and raw sewage. Inoculum was aged prior to test initiation. The test chemicals were added to flasks containing medium and inoculum. The flask were incubated and shaken in the dark for 28 days. Twelve flasks were prepared; 3 controls, 3 dextrose, 3 test substance and 3 with test substance and HgCl2 (to prevent microbial growth). The CO2

production was captured in KOH solution.

500ml Erlenmeyer flasks were used as test vessels. Test flasks were shaken at a rate of 60rpm at 25 +/- 2 deg C. Plate count at initiation was 1.7 x 105 colony/ml. The pH at initiation was not reported.

Three test trials were conducted. The methods described are those of trial #3.

Test substance

Nominal test concentration for all substances = 10mg/L : Tris (2-ethylhexyl) Trimellitate (CAS# 3319-31-1)

(1,2,-benzenedicarboxylic acid, Tris (2-ethylhexyl) Ester)

ld 3319-31-1

Date 13.12.2001

Conclusion

Synonym: TOTM

: The substance is readily biodegradable using mixed populations of

Reliability Flag 29.11.2000

microorganisms,
: (2) valid with restrictions
: Critical study for SIDS endpoint

(1)

# 4. Ecotoxicity

ld 3319-31-1 Date 13.12.2001

### 4.1 ACUTE/PROLONGED TOXICITY TO FISH

See attached TOTM SIAR document

# 4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES

See attached TOTM SIAR document

# 4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

See attached TOTM SIAR document

ld 3319-31-1 Date 13.12.2001

#### 5.1.1 ACUTE ORAL TOXICITY

Type

LD50

Species

rat

Strain Sex

male

Number of animals

: 20

Vehicle

: other

Value Method : > 3200 mg/kg bw

Year **GLP** 

: other : 1971

Test substance

: no : other TS

Method

: Rats and mice

Remark

: No animals died. All animals gained weight post-exposure

Test condition

: Two male rats and two male mice per dose level were treated with 200, 400, 800, 1600, or 3200 mg/kg neat test substance by oral gavage. The animals were observed for a period of 14 days for survival and body weight

Test substance

: 1,2,4-benzenetricarboxylic, tris(2-ethylhexyl)ester (tri-2-ethylhexyl

trimellitate)

Reliability

: (2) valid with restrictions

28.12.2000

(4)

(5)

#### 5.1.2 ACUTE INHALATION TOXICITY

Type

: LC50

Species Strain

: rat

: no data

Sex

3

Number of animals

Vehicle Exposure time

: 6 hour(s)

Value

other

Method

Year

: 1971

**GLP** Test substance

: other TS

Remark

: Three rats were exposed for 6 hours to a nominal concentration of 4.17, 2.64, or 0.23 mg/L of the test substance in a whole-body inhalation chamber. The test substance was heated to 180oC to generate the test atmosphere which was likely a mixture of aerosol and heated vapor. The

animals were observed for a period of 14 days for survival.

Test condition

: 1,2,4-benzenetricarboxylic, tris(2-ethylhexyl)ester (tri-2-ethylhexyl

Test substance

: Three rats were exposed for 6 hours to a nominal concentration of 4.17, 2.64, or 0.23 mg/L of the test substance in a whole-body inhalation chamber. The test substance was heated to 180oC to generate the test atmosphere which was likely a mixture of aerosol and heated vapor. The animals were observed for a period of 14 days for survival.

Conclusion Reliability 28.12.2000

: 100% mortality at >2.64 mg/L (nominal). LC50 not determined.

: (2) valid with restrictions

5.1.3 ACUTE DERMAL TOXICITY

11/16

ld 3319-31-1 Date 13.12.2001

Type Species : LD50 : guinea pig

Strain

Sex

no data

Number of animals Vehicle

: other

Venicie Value

: > 20 ml/kg bw

Method Year : other

rear GLP

: no

Test substance

: other TS

Remark

: No animals died. Moderate to severe edema and moderate erythema were observed at 24 hours. All animals appeared normal after one week.

**Test condition** 

Three guinea pigs were shaved and depilated prior to dosing. Dose levels of 5, 10, or 20 ml/kg of the neat test substance were applied to the skin (one animal per dose level) and the area wrapped with an impervious material for 24 hours. Following unwrapping, the skin was evaluated for signs of irritation. The animals were observed for a period of 14 days for

Test substance

: 1,2,4-benzenetricarboxylic, tris(2-ethylhexyl)ester (tri-2-ethylhexyl

trimellitate)

Conclusion

: Under the conditions of this study, the test substance has a low order of

acute dermal toxicity in rats.

Reliability 28.12.2000

: (2) valid with restrictions

(3)

#### 5.4 REPEATED DOSE TOXICITY

Species

rat

Sex

:

Strain
Route of admin.
Exposure period

: Fischer 344: oral feed: 28 days: Daily

Frequency of treatment

: No

Post obs. period

: 0, 0.2, 0.67 or 2.0% (0, 183, 654, 1734 mg/kg/day

Doses
Control group

: yes : = .67 %

NOAEL Method

: OECD Guide-line 407 "Repeated Dose Oral Toxicity - Rodent: 28-day or 14-d Study"

Year GLP : 1985 : yes

Test substance

: Analysis of variance with significant groups compared by Least Significant

Method

Difference test (p < 0.05)

Remark

There were no statistically significant differences between the body weights of control and treated animals. Initially, there was a significant decrease in food intake for females (2%); however, food intake gradually increased throughout the study. In males, there were no treatment-related effects on food consumption. Serum albumin levels were significantly increased in males and females at the mid and high dose. Similarly, leukocyte counts were increased in both sexes at the two higher dose levels; however, this increase was significant only in males. At the two lower dose levels, hematocrit and mean cell volume decreased in female rats. In both sexes, the absolute and relative liver weights increased at the mid-dose level, but not at the highest dose. The high dose animals showed slight increases in the number but not size of peroxisomes. There were no deaths related to treatment in this study.

## 5. Toxicity

ld 3319-31-1 Date 13.12.2001

#### **Test condition**

: Male and female rats were randomly assigned to various treatment groups. Following the acclimation period, rats were administered dietary doses of either the control or the test substance for 28 days. DEHP was used in this study as a reference compound. Animals were monitored twice each day. Food intake was measured from Days -3 to day 0 and continuous intakes were measured at twice-weekly intervals until the day preceding the autopsy. One day prior to autopsy, blood was collected from each animal and the following endpoints were evaluated: differential leukocyte and erythrocyte counts, mean cell volume, packed cell volume, total leukocyte count, platelet count and reticulocyte count. Serum chemistry analysis of several endpoints was also performed. On the day of sacrifice, the following organs were retained in 10% neutral buffered formalin: cecum, colon, pancreas, prostate, skeletal muscle, small intestine, stomach, thymus, and uterus. Two slices of liver were subjected to electron microscopy for evaluation of peroxisome proliferation.

Test substance

: 1,2,4-benzenetricarboxylic, tris(2-ethylhexyl)ester (tri-2-ethylhexyl

trimellitate)

Conclusion

: The test substance caused a slight peroxisome proliferation at the high

dose but was less potent than comparable doses of DEHP.

**Reliability** 28.12.2000

: (1) valid without restriction

## 5.5 GENETIC TOXICITY (IN VITRO)

Type

: Ames test

System of testing

Bacterial

Concentration

100, 333, 1000, 3333, 10000 mg/ml.

Cycotoxic conc. Metabolic activation

: with and without

Result

: negative

Method

: OECD Guide-line 471 "Genetic Toxicology: Salmonella thyphimurium

Reverse Mutation Assay"

Year GLP : 1988

Test substance

: yes

Method

: other TS

: Chemicals were judged to be mutagenic if the test results produced a dose-related, reproducible increase in histidine revertants over control. It was not a requirement for mutagenic responses to reach two-fold over

**Test condition** 

background.

Prior to assay initiation, a toxicity pretest was performed using tester strain

TA100. Based on these results, the doses for the final assay were determined. In the definitive assay, each of the five strains was dosed with either the test substance; a vehicle control (DMSO); or a nontreated control and a positive control. The test mixture containing the tester strain and test substance with or without S9 was added to the surface of petri dishes containing Vogel-Bonner medium. The histidine-independent colonies that formed on the plates were counted following a two-day incubation at 37°C. Positive controls were as follows: 2-aminoanthracene (all strains with S9); sodium azide (without S9, TA1535, TA100), 4-nitro-o-phenylenediamine (without S9, TA98) and 9-aminoacridine (without S9, TA 97, TA1537). There were 3 plates/dose group/strain/treatment. The test results were verified by repeating the assay. If the results were negative, they were

Test substance

repeated first without S9 and then with 30% S9.

1,2,4-benzenetricarboxylic, tris(2-ethylhexyl)ester (tri-2-ethylhexyl)

Conclusion

trimellitate)
: Under the conditions of this study, tri (2-ethylhexyl) trimellitate was not

mutagenic at doses up to 10,000 mg/ml.

**Reliability** 28.12.2000

: (1) valid without restriction

## 5. Toxicity

ld 3319-31-1 Date 13.12.2001

#### 5.8 TOXICITY TO REPRODUCTION

See attached TOTM SIAR document

## 5.9 DEVELOPMENTAL TOXICITY/TERATOGENICITY

See attached TOTM SIAR document

## 5.10 OTHER RELEVANT INFORMATION

See attached TOTM SIAR document

# 6. References ld 3319-31-1 Date 13.12.2001

- (1) ABC Final Report, # 31891, Shake Flask Biodegradation of 14C-Tris (2-ethylhexyl) Trimellitate (TOTM). 1986, Sponsored by CMA. Performed by: ABC Laboratories Columbia, MO.
- (2) E. Zeiger, B. Anderson, S. Haworth, T. Lawlor and K. Mortelmans. (1988). Salmonella mutagenicity tests. IV. Results from the testing of 300 chemicals. Environmental and Molecular Mutagenesis 11(12):1-158.
- (3) Eastman Kodak Company (1971). Tri(2-ethylhexyl)trimellitate. Acute dermal toxicity. Unpublished report.
- (4) Eastman Kodak Company (1971). Tri(2-ethylhexyl)trimellitate. Acute oral toxicity. Unpublished report.
- (5) Eastman Kodak Company (1971). Tri(2-ethylhexyl)trimellitate. Acute inhalation toxicity. Unpublished report.
- (6) Mackay, D., A. DiGuardo, S. Paterson and C. Cowan, EQC Model ver. 1.01, 1997, available form the Emvironmental Centre, Trent Univ. Canada.
- (7) Manufacturer Safety Data Sheet
- (8) Meylan, M. Syracuse Research Corporation (1994-1999) Calculation program contained in EPIWIN (Esitmate ver. 3.04) available from SRC.
- (9) Wilson, A., (1996). Plasticizers Selection, Applications and Implications. Rapra Review Reports 8:15-16.

id 3319-31-1

Date 13.12.2001

#### 7.2 **HAZARD SUMMARY**

Chapter Remark

: Chapters 4 & 5

Because of the similarity in chemical structure, the Panel believes that the toxicological properties of the substances in this category will be similar as well. Thus, the Panel considers that the data for the best tested member of this category, tris-2-(ethylhexyl) trimellitate (TOTM), also represents the potential for human and environmental effects of the other members of this category.

TOTM has been sponsored under the OECD SIDS program through ICCA. A review of the available data for TOTM (see attached Table) indicates that all endpoints have been adequately addressed, and that TOTM exhibits a low order of toxicity.

Due to their higher molecular weight and bulky side chains, the remaining members of this category are expected to demonstrate a lower order of toxicity than TOTM. This is supported by a similar structural-activity relationship observed with phthalate ester compounds, i.e., the higher molecular weight phthalates (ester side chains >C7) are less active that the transitional phthalates (ester side chains C4-C6). Thus, the use of TOTM to represent the potential hazards of the other category members is a conservative position.

Attached doc.

Summary of SIDS Information on Trimellitates.doc

Flag

Critical study for SIDS endpoint

13.12.2001

#### 73 RISK ASSESSMENT

Memo

: SIDS Initial Assessment Profile (SIAP), SIDS Initial Assessment Report (SIAR) and Robust Summary for TOTM -- submitted by Japan under ICCA HPV program.

Attached doc.

: TOTM SIAR.pdf

Flag

13.12.2001

: Critical study for SIDS endpoint

# Summary of SIDS Information on Trimellitates A. Physical/Chemical Properties of Trimellitates

(R)							Water	Photodeg	Hydrolysis	Transport (%) c			
Carbon Chain Length	CAS Number	Chemical Name	MP* (°C)	BP** (°C)	VP (hPa@25°C)	PC (log Pow)	Solubility (mg/L @25°C)	Half-life	Half-life (yrs)	Soil	Air	Water	Sediment
C8	3319-31-1	tris 2-ethylhexyl (TOTM)	-46 97 c	>300 541 c	<0.0001*** 5.25E-11 c	5.94 11.59 c	3.9E-04 4.51E-08 c	0.33 c	0.05 0.32 c	97.8	3.6E - 6	2.8E - 7	2.17
C8	27251-75-8	triisooctyl ester	<0 197 c	541 c	5.25E-11 c	11.59 с	4.51E-08 c	0.35 с	0.43 с	97.8	3.64E - 6	2.8E - 7	2.17
C9	53894-23-8	triisononyl ester	<0 224 c	>300 575 c	3.17E-12 c	13.06 c	1.32E-09 c	0.31 c	0.86 с	97.8	2.74E - 7	9.61E -9	2.17
C8,C10	67989-23-5	decyl, octyl ester	<0 234 c	585 c	1.37E-12 c	12.79 c	2.78E-09 c	0.32 c	0.98 с	97.8	1.02E - 7	1.79E - 8	2.17

c = calculated data using EPWIN; all other values are derived from measurements

<sup>\* =</sup> All of these trimellitates are liquids at zero degrees C. Modeled data do not accurately reflect melting points for these substances

<sup>\*\* =</sup> Measured boiling points were determined to be >300°C at 0.66 kPa

<sup>\*\*\* =</sup> vapor pressure of TOTM 13 Pa @ 200°C

## **Summary of SIDS Information on Trimellitates** B. Toxicology Data on Trimellitates

(R) Carbon Chain Length	CAS Number	Chemical Name	Acute Oral LD50	Acute Dermal LD50	Acute Inhalation LC50	Repeated Dose Toxicity	GeneTox (Ames)	GeneTox (Chrom. Abs.)	Toxicity to Reproduction	Developmental Toxicity / Teratogenicity	Acute Fish (A) mg/L	Daphnia (B) mg/L	Algal (C) mg/L	Biodegradation %
C8	3319-31-1	tris 2-ethylhexyl (TOTM)	> 3.2 g/kg (rat, mouse)	>20 ml/kg (guinea pig) >2.0 ml/kg (rabbit)	<2.64 mg/L (rat, nominal)	NOAEL (rat, dietary) 654 mg/kg/day	Negative	Negative (CHL/IU cells)	NOAEL (rat, oral) 1000 mg/kg/day	NOAEL (rat, oral) 1000 mg/kg/day (3)	>100	>180	>100	68-71 (1) 4.2 (2)
C8	27251-75-8	Triisooctyl ester												
C9	53894-23-8	Triisononyl ester	> 10 g/kg (rat)											
C8, C10	67989-23-5	decyl, octyl ester												

- Footnotes: A) Japanese Medaka (Oryzias latipes), 96 hr LC50 & NOEC
  - B) Daphnia magna, 48-hr EC50
  - C) Selenastrum capricomutum, 72-hr EC50 & NOEC
  - (1) Inherent biodegradation by Shake Flask Method
  - (2) Ready biodegradation by MITI method (OECD 301C)
  - (3) OECD Preliminary reproduction toxicity screening test; indirect measure of develomental effects

# IUCLID

## **Data Set**

**Existing Chemical** 

CAS No.

TSCA Name Generic name : ID: 27251-75-8

: 27251-75-8

: 1,2,4-benzenetricarboxcylic acid, triisooctyl ester

: triisooctyl ester trimellitate

**Producer Related Part** 

Company Creation date : ExxonMobil Biomedical Sciences Inc.

: 26.10.2000

**Substance Related Part** 

Company Creation date : ExxonMobil Biomedical Sciences Inc.

: 26.10.2000

Memo

: ACC Phthalate Esters HPV Panel

Printing date

: 13.12.2001

Revision date

•

**Date of last Update** 

: 30.10.2001

**Number of Pages** 

: 11

Chapter (profile)
Reliability (profile)
Flags (profile)

:

#### 1. General Information

ld 27251-75-8

Date 13.12.2001

#### 1.0.1 OECD AND COMPANY INFORMATION

Type

: lead organisation

Name

ACC Phthalate Esters Panel HPV Testing Group

Partner

Dr. Marian Stanley

Date

Street Town

1300 Wilson Blvd. 22209 Arlington, VA

Country Phone

**United States** (703) 741-5623

Telefax

(703) 741-6091

Telex Cedex

Remark

The American Chemistry Council Phthalate Esters Panel sponsoring this

test plan includes the following member companies:

Eastman Chemical Company ExxonMobil Chemical Company

Sunoco Chemicals **Teknor Apex Company** 

Flag

26.10.2001

: Critical study for SIDS endpoint

## 1.1 GENERAL SUBSTANCE INFORMATION

Substance type

organic

Physical status

liquid

**Purity** 

% w/w

09.10.2001

## 1.1.0 DETAILS ON TEMPLATE

Comment

This chemical is part of the Trimellitate category. The category includes

the following four CAS numbers: 3319-31-1, 27251-75-8, 53894-23-8 and

67989-23-5.

Remark

DESCRIPTION OF THE TRIMELLITATES CATEGORY

The trimellitates comprise a family of chemicals synthesized by esterifying trimellitic anhydride with alcohols with average carbon numbers ranging from approximately C7-C10, in the presence of an acid catalyst. The category includes the four trimellitates: 3319-31-1 (TOTM), 27251-75-8

(TIOTM),

53894-23-8 (TINTM), and 67989-23-5 (DOTM). Trimellitates in this category are all 1,2,4-benzenetricarboxylic acids with side chain ester groups ranging from C8 to C10. The structural formula for trimellitates varies somewhat depending on the isomeric composition of the alcohols used in their manufacture. The specific alcohols used are 2-ethylhexanol (TOTM), iso-octyl alcohol (TIOTM), iso-nonyl alcohol (TINTM), and a mixture of linear and branched decyl (40%) and octyl (60%) alcohols (DOTM).

Trimellitates are colorless to slightly yellow liquids with high boiling points (> 250oC) and low vapor pressures; properties which contribute to their high physical stability. They are readily soluble in most organic solvents and miscible with alcohol, ether and most oils, but essentially insoluble in

#### 1. General Information

ld 27251-75-8 Date 13.12.2001

water. Because of the similarity in structure as well as physicochemical properties, the trimellitates were grouped into a single category containing four substances with carboxylic side chain ester groups ranging from C8-

Flag 09.10.2001 Critical study for SIDS endpoint

#### 1.7 USE PATTERN

Type Category Remark : industrial

Polymers industry

: Trimellitates are used predominantly as plasticizers for production of flexible PVC. Because of their relatively high molecular weight (>500 g/mole) and bulky structure, they have lower volatility and greater resistance to migration than the corresponding phthalate ester plasticizers. They are predominantly used in the manufacture of high temperature PVC cables (Wilson, 1996). Since these chemicals are produced in closed systems, there is essentially no occupational exposure to these substances except at the flexible PVC production facility. Usually, these substances have been already blended to the compound as plasticizer, so it is not expected that downstream users or consumers are directly

exposed to trimellitates.

Flag

13.12.2001

: Critical study for SIDS endpoint

(3)

## 2. Physico-Chemical Data

ld 27251-75-8 Date 13.12.2001

#### 2.1 **MELTING POINT**

Value

: = 197 ° C

Decomposition Sublimation

: no at : no

Method Year

: other : 2000

GLP

Test substance Method

: Melting point calculation by MPBPWIN ver. 1.40 using calculation methods

of Joback and Gold and Ogle.

Remark

: EPIWIN is used and advocated by the US EPA for chemical property

Melting point calculation seems to give erroneously high results for the this

class of chemicals.

Source

: ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability

29.10.2001

: (3) invalid

(2)

## 2.2 BOILING POINT PARTY OF THE PROPERTY OF THE

Year

: = 541 °C at 1013 hPa

Decomposition Method

: no other

**GLP** 

2000

Test substance

Method

: Boiling point calculation by MPBPWIN ver. 1.40 using calculation method

of Stein and Brown.

Remark

: EPIWIN is used and advocated by the US EPA for chemical property

estimation

Source

: ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability

: (2) valid with restrictions

29.10.2001

(2)

#### 2.4 VAPOUR PRESSURE

Value

: = .000000000524 hPa at 25° C

Decomposition

Method

other (calculated)

Year

2000

GLP Test substance

Decomposition

Method

: Vapor pressure calculation by MPBPWIN ver. 1.40 using calculation

method of Grain.

Remark

: EPIWIN is used and advocated by the US EPA for chemical property

estimation.

Source

: ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability : (2) valid with restrictions

29.10.2001

(2)

#### 2.5 **PARTITION COEFFICIENT**

## 2. Physico-Chemical Data

ld 27251-75-8 Date 13.12.2001

Log pow Method

: = 11.59 at 25° C other (calculated)

Year

2000

**GLP** 

Test substance

Method

Partition coefficient by LOGKOWWIN ver. 1.65 using an atom/fragment

calculation method of Meylan and Howard.

Remark

: EPIWIN is used and advocated by the US EPA for chemical property

estimation.

Source

: ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability

29.10.2001

: (2) valid with restrictions

(2)

### 2.6.1 WATER SOLUBILITY

Value

.00005 other: ug/L at 25 ° C

Qualitative

Pka

at 25 ° C at and °C

PH Method

other: calculated

2000 Year

**GLP** 

Test substance

Method

Water solubility calculated using WSKOWWIN ver. 1.36 based on Kow

correlation method of Meylan and Howard.

Remark

EPIWIN is used and advocated by the US EPA for chemical property

estimation.

Source

ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability 30.10.2001 : (2) valid with restrictions

## 3. Environmental Fate and Pathways

ld 27251-75-8 Date 13.12.2001

#### 3.1.1 PHOTODEGRADATION

Type : air
Light source : Sun light

Light spect. : nm

Rel. intensity : 1 based on Intensity of Sunlight

Conc. of subst. : at 25 degree C

Indirect photolysis

Sensitizer : OH

Conc. of sens. : 1500000 molecule/cm3

Rate constant : .0000000003068 cm3/(molecule\*sec)

Degradation : % after
Deg. Product : not measured
Method : other (calculated)

Year : 2000

GLP Test substance

Method : Photodegradation rate calculated by AOPWIN ver. 1.89 based on the

methods of Atkinson.

Remark : EPIWIN is used and advocated by the US EPA for chemical property

estimation.

Source : ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability : (2) valid with restrictions

29.10.2001 (2)

## 3.1.2 STABILITY IN WATER

Type : abiotic

t1/2 pH4 : at degree C

t1/2 pH7 : .4 year at 25 degree C

t1/2 pH9 : at degree C
Deg. Product : not measured
Method : other (calculated)

Year : 2000

GLP

Test substance

Method : Hydrolysis rate calculated by HYDROWIN ver. 1.67 based on work for EPA

by T. Mill et al.

Remark : EPIWIN is used and advocated by the US EPA for chemical property

estimation.

Source : ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability : (2) valid with restrictions

29.10.2001 (2)

## 3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

Type : fugacity model level !

 Media
 : other

 Air (level I)
 : 0

 Water (level I)
 : 0

 Soil (level I)
 : 97.8

Biota (level II / III)

Soil (level II / III)

Method : other Year : 2000

6/11

## 3. Environmental Fate and Pathways

ld 27251-75-8

Date 13.12.2001

Source

: ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability 20.12.2000

: (2) valid with restrictions

(1)

#### 3.3.2 DISTRIBUTION

Media Method : air - biota - sediment(s) - soil - water: Calculation according Mackay, Level I

Year

: 2000

Result

: Soil - 97.8%

Air - 0.00000364% Water - 0.000000284% Sediment - 2.17%

Suspended sed. - 0.068%

Source Reliability : ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

: (2) valid with restrictions

20.12.2000

(1)

4.	<b>Ecotoxicit</b>	٧
₹.	LCCLOXICIL	y

ld 27251-75-8

Date 13.12.2001

- 4.1 ACUTE/PROLONGED TOXICITY TO FISH
- 4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES
- 4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

	<b>Date</b> 13.12.2001
5.1.1 ACUTE ORAL TOXICITY	
5.1.2 ACUTE INHALATION TOXICITY	
5.1.3 ACUTE DERMAL TOXICITY	
5.4 REPEATED DOSE TOXICITY	
5.5 GENETIC TOXICITY 'IN VITRO'	
5.8 TOXICITY TO REPRODUCTION	
5.9 DEVELOPMENTAL TOXICITY/TERATO	GENICITY

ld 27251-75-8

5. Toxicity

## 6. References

ld 27251-75-8 **Date** 13.12.2001

- (1) Mackay, D., A. DiGuardo, S. Paterson and C. Cowan, EQC Model ver. 1.01, 1997, available form the Emvironmental Centre, Trent Univ. Canada.
- (2) Meylan, M. Syracuse Research Corporation (1994-1999) Calculation program contained in EPIWIN (Esitmate ver. 3.04) available from SRC.
- (3) Wilson, A., (1996). Plasticizers Selection, Applications and Implications. Rapra Review Reports 8:15-16.

ld 27251-75-8

Date 13.12.2001

#### 7.1 END POINT SUMMARY

#### 7.2 HAZARD SUMMARY

Chapter Remark

: Chapters 4 & 5

: Because of the similarity in chemical structure, the Panel believes that the toxicological properties of the substances in this category will be similar as well. Thus, the Panel considers that the data for the best tested member of this category, tris-2-(ethylhexyl) trimellitate (TOTM), also represents the potential for human and environmental effects of the other members of this category.

TOTM has been sponsored under the OECD SIDS program through ICCA. A review of the available data for TOTM (see attached Table) indicates that all endpoints have been adequately addressed, and that TOTM exhibits a low order of toxicity.

Due to their higher molecular weight and bulky side chains, the remaining members of this category are expected to demonstrate a lower order of toxicity than TOTM. This is supported by a similar structural-activity relationship observed with phthalate ester compounds, i.e., the higher molecular weight phthalates (ester side chains >C7) are less active that the transitional phthalates (ester side chains C4-C6). Thus, the use of TOTM to represent the potential hazards of the other category members is a conservative position.

Attached doc.

: Summary of SIDS Information on Trimellitates.doc

Flag 13.12.2001 : Critical study for SIDS endpoint

7:3 RISK ASSESSMENT

# IUCLID

# **Data Set**

**Existing Chemical** 

CAS No.

**TSCA Name** 

Generic name

: ID: 27251-75-8

: 27251-75-8

: 1,2,4-benzenetricarboxcylic acid, triisooctyl ester

: triisooctyl ester trimellitate

**Producer Related Part** 

Company Creation date : ExxonMobil Biomedical Sciences Inc.

: 26.10.2000

**Substance Related Part** 

Company Creation date : ExxonMobil Biomedical Sciences Inc.

: 26.10.2000

Memo

: ACC Phthalate Esters HPV Panel

Printing date Revision date : 13.12.2001

**Date of last Update** 

: 30.10.2001

**Number of Pages** 

: 11

## 1. General Information

ld 27251-75-8

Date 13.12.2001

#### 1.0.1 OECD AND COMPANY INFORMATION

Type

lead organisation

Name

ACC Phthalate Esters Panel HPV Testing Group

Partner

Dr. Marian Stanley

Date

Street Town Country 1300 Wilson Blvd. 22209 Arlington, VA **United States** 

Phone Telefax (703) 741-5623 (703) 741-6091

Telex

Cedex

Remark

The American Chemistry Council Phthalate Esters Panel sponsoring this

test plan includes the following member companies:

Eastman Chemical Company **ExxonMobil Chemical Company** 

Sunoco Chemicals **Teknor Apex Company** 

Flag 26.10.2001

Critical study for SIDS endpoint

### 1.1 GENERAL SUBSTANCE INFORMATION

Substance type

organic liquid

Physical status **Purity** 

% w/w

09.10.2001

#### 1.1.0 DETAILS ON TEMPLATE

Comment

This chemical is part of the Trimellitate category. The category includes the following four CAS numbers: 3319-31-1, 27251-75-8, 53894-23-8 and

67989-23-5.

Remark

DESCRIPTION OF THE TRIMELLITATES CATEGORY

The trimellitates comprise a family of chemicals synthesized by esterifying trimellitic anhydride with alcohols with average carbon numbers ranging from approximately C7-C10, in the presence of an acid catalyst. The category includes the four trimellitates: 3319-31-1 (TOTM), 27251-75-8 (TIOTM),

53894-23-8 (TINTM), and 67989-23-5 (DOTM). Trimellitates in this category are all 1,2,4-benzenetricarboxylic acids with side chain ester groups ranging from C8 to C10. The structural formula for trimellitates varies somewhat depending on the isomeric composition of the alcohols used in their manufacture. The specific alcohols used are 2-ethylhexanol (TOTM), iso-octyl alcohol (TIOTM), iso-nonyl alcohol (TINTM), and a mixture of linear and branched decyl (40%) and octyl (60%) alcohols (DOTM).

Trimellitates are colorless to slightly yellow liquids with high boiling points (> 250oC) and low vapor pressures; properties which contribute to their high physical stability. They are readily soluble in most organic solvents and miscible with alcohol, ether and most oils, but essentially insoluble in

### 1. General Information

ld 27251-75-8

Date 13.12.2001

water. Because of the similarity in structure as well as physicochemical properties, the trimellitates were grouped into a single category containing four substances with carboxylic side chain ester groups ranging from C8-

Flag 09.10.2001 Critical study for SIDS endpoint

### 1.7 USE PATTERN

Type Category Remark : industrial

Polymers industry

Trimellitates are used predominantly as plasticizers for production of flexible PVC. Because of their relatively high molecular weight (>500 g/mole) and bulky structure, they have lower volatility and greater resistance to migration than the corresponding phthalate ester plasticizers. They are predominantly used in the manufacture of high temperature PVC cables (Wilson, 1996). Since these chemicals are produced in closed systems, there is essentially no occupational exposure to these

substances except at the flexible PVC production facility. Usually, these substances have been already blended to the compound as plasticizer, so it is not expected that downstream users or consumers are directly

exposed to trimellitates.

Flag 13.12.2001 : Critical study for SIDS endpoint

(3)

## 2. Physico-Chemical Data

ld 27251-75-8 Date 13.12.2001

#### 2.1 **MELTING POINT**

Value

: = 197 ° C

Decomposition

: no at : no

Sublimation Method

: other : 2000

Year **GLP** 

Test substance

Method

: Melting point calculation by MPBPWIN ver. 1.40 using calculation methods

of Joback and Gold and Ogle.

Remark

: EPIWIN is used and advocated by the US EPA for chemical property

Melting point calculation seems to give erroneously high results for the this

class of chemicals.

Source

: ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability

: (3) invalid

29.10.2001

2.2 BOILING POINT

: = 541 °C at 1013 hPa **Value** 

Decomposition Method

no other

**GLP** 

2000

Test substance

Method

: Boiling point calculation by MPBPWIN ver. 1.40 using calculation method

of Stein and Brown.

: EPIWIN is used and advocated by the US EPA for chemical property Remark

estimation

Source

Year

: ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability

: (2) valid with restrictions

29.10.2001

(2)

### 2.4 VAPOUR PRESSURE

Value

= .0000000000524 hPa at 25° C

Decomposition

Method

other (calculated)

Year

2000

GLP

Test substance

: no

Decomposition

Method

: Vapor pressure calculation by MPBPWIN ver. 1.40 using calculation

method of Grain.

Remark

: EPIWIN is used and advocated by the US EPA for chemical property

estimation.

Source

: ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability

: (2) valid with restrictions

29.10.2001

(2)

(2)

## 2. Physico-Chemical Data

ld 27251-75-8 Date 13.12.2001

(2)

#### 2.5 PARTITION COEFFICIENT

Log pow

: = 11.59 at 25° C other (calculated)

Method Year

2000

**GLP** 

Test substance Method

Partition coefficient by LOGKOWWIN ver. 1.65 using an atom/fragment

calculation method of Meylan and Howard.

Remark

EPIWIN is used and advocated by the US EPA for chemical property

estimation.

Source

: ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability

: (2) valid with restrictions

29.10.2001

## 2.6.1 WATER SOLUBILITY

Value .00005 other: ug/L at 25 ° C

Qualitative

at 25 ° C

PH Method

at and °C other: calculated

Year

2000

**GLP** 

Pka

Test substance

Method

: Water solubility calculated using WSKOWWIN ver. 1.36 based on Kow

correlation method of Meylan and Howard.

Remark : EPIWIN is used and advocated by the US EPA for chemical property

estimation.

Source ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability

: (2) valid with restrictions

30.10.2001

## 3. Environmental Fate and Pathways

ld 27251-75-8 **Date** 13.12.2001

#### 3.1.1 PHOTODEGRADATION

Type : air Light source : Sun light

Light spect. : nm

Rel. intensity : 1 based on Intensity of Sunlight

Conc. of subst. : at 25 degree C

Indirect photolysis

Sensitizer : OH

Conc. of sens. : 1500000 molecule/cm3

Rate constant : .00000000003068 cm3/(molecule\*sec)

Degradation: % afterDeg. Product: not measuredMethod: other (calculated)

Year : 2000

GLP

Test substance

Method : Photodegradation rate calculated by AOPWIN ver. 1.89 based on the

methods of Atkinson.

Remark : EPIWIN is used and advocated by the US EPA for chemical property

estimation

Source : ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability : (2) valid with restrictions

29.10.2001 (2)

#### 3.1.2 STABILITY IN WATER

Type : abiotic

t1/2 pH4 : at degree C

t1/2 pH7 : .4 year at 25 degree C

t1/2 pH9 : at degree C
Deg. Product : not measured
Method : other (calculated)

Year : 2000

GLP

Test substance

Method : Hydrolysis rate calculated by HYDROWIN ver. 1.67 based on work for EPA

by T. Mill et al.

Remark : EPIWIN is used and advocated by the US EPA for chemical property

estimation.

Source : ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability : (2) valid with restrictions

29.10.2001 (2)

#### 3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

Type : fugacity model level !

 Media
 : other

 Air (level I)
 : 0

 Water (level I)
 : 0

 Soil (level I)
 : 97.8

Biota (level II / III)
Soil (level II / III)

Method : other Year : 2000

6/11

## 3. Environmental Fate and Pathways

ld 27251-75-8 Date 13.12.2001

Source

: ExxonMobil Biomedical Sciences, Inc. Annandale, N.J. USA

Reliability 20.12.2000 : (2) valid with restrictions

(1)

#### 3.3.2 DISTRIBUTION

Media Method air - biota - sediment(s) - soil - water
Calculation according Mackay, Level I

Year

: 2000

Result : Soil - 97.8%

Air - 0.00000364% Water - 0.000000284% Sediment - 2.17%

Suspended sed. - 0.068%

Source

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Reliability

: (2) valid with restrictions

20.12.2000

(1)

4. Ecotoxicity	4.	Ecc	otoxi	city
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ld 27251-75-8

Date 13.12.2001

- 4.1 ACUTE/PROLONGED TOXICITY TO FISH
- 4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES
- 4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

5. Toxicity	ld 27251	-75-8
•	<b>Date</b> 13.12.	2001
5.1.1 ACUTE ORAL TOXICITY		
5.1.2 ACUTE INHALATION TOXICITY	en e	
5.1.3 ACUTE DERMAL TOXICITY		in the second se
5.4 REPEATED DOSE TOXICITY		
5.5 GENETIC TOXICITY 'IN VITRO'		
5.8 TOXICITY TO REPRODUCTION		
5.9 DEVELOPMENTAL TOXICITY/TERATOGENICITY		

ld 27251-75-8

## 6. References

ld 27251-75-8

Date 13.12.2001

- (1) Mackay, D., A. DiGuardo, S. Paterson and C. Cowan, EQC Model ver. 1.01, 1997, available form the Emvironmental Centre, Trent Univ. Canada.
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ld 27251-75-8 Date 13.12.2001

#### 7.1 END POINT SUMMARY

#### 7.2 HAZARD SUMMARY

Chapter Remark : Chapters 4 & 5

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Attached doc.

Flag

13.12.2001

: Summary of SIDS Information on Trimellitates.doc

: Critical study for SIDS endpoint

7.3 RISK ASSESSMENT